# Acid-Base Potential Characterisation in the Southern Highveld Coalfield of Mpumalanga

by

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Dissertation submitted in fulfilment of the requirements in respect of the Master's Degree qualification MSc (Geohydrology) at the Institute for Groundwater Studies in the Faculty of Natural and Agricultural Sciences at the University of the Free State.

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July 2017





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#### CHAPTER 1 INTRODUCTION

#### 1.1 Background information

Coal is valued for its energy content and has been widely used to generate electricity. 77% percent of South Africa's total electrical energy is generated from coal. According to Trending Top Most report of 2017 (www.trendingtopmost.com), South Africa is ranked seventh in the global coal production after China, United States, India, Australia, Indonesia and Russia. It produces about 260 million tons of coal and is the sixth largest coal exporter having traded about 74 million tons in 2012. The coal is mainly exported to Europe, China and India.

Coal is generally mined by opencast/open pit or underground methods in South Africa. Fiftyone percent of South African coal is produced by underground mining and 49 percent is produced by open cast methods (Department of Energy, 2014). During both coal mining processes, a variety of rock types with different compositions are removed from the surrounding and exposed to atmospheric condition and undergo accelerated weathering (Bhuiyan *et al.*, 2010). These materials are often deposited nearby as mine waste rocks and mine dust which causes a number of environmental problems such as Acid Mine Drainage (AMD).

By definition, AMD is acidic water that is produced when sulphide minerals are exposed to air and water resulting in a chemical reaction that produces acidic mine water (www.miningfacts.org). It is widely accepted that AMD and its potential impact on groundwater resources is one of the most serious environmental concern associated with coal mining (Brady *et al.*, 1997 and Bell *et al.*, 2001). It is therefore seen as one of the great threats to the water resources in South Africa, and it is imperative that the mining industries are able to predict and evaluate the environmental consequences (Usher, 2003) resulting from the AMD. In the mining areas, AMD is the main contaminant of water resources which can render it unsafe for consumption, industrial and agricultural applications. Vermeulen and Usher (2009) reported that water related problems, largely associated with water quality deterioration, due to pyrite oxidation, occur as a result of mining in the South African coal fields. When the acidic mine water is released into the environment, the high salinities of this drainage degrade the water quality considerably (Vermeulen & Usher, 2009).

Today, there are many measures that can be used to investigate the potential impact or effects of AMD, such methods include: Kinetic method, Modelling of oxidation, pollutant generation and release, Modelling of material composition and Acid Base Accounting (ABA). Due to its simplicity, the ABA method is a commonly and preferred method. The ABA provides the basis for evaluating acid mine production potential of ore and waste rock. This

study was aimed at assessing the Acid Base Potential of two mines located in the Southern Highveld Coalfield of Mpumalanga using the ABA method.

# 1.1.1 Research aims and objectives

The aim of the study was to assess the acid base potential and leaching of trace elements at Mine 1 and 2 in the Highveld Coalfields in Mpumalanga. The aims were achieved through the following specific objectives:

- Review of the literature on coal mining and its potential environmental impacts in South Africa,
- Collection of geological samples,
- Analysis of mineralogy of samples and determination of the chemical composition of the whole rock and
- Conduct acid base accounting tests.

# 1.1.2 Structure of the dissertation

This dissertation consists of 6 chapters:

- Chapter 1 is the introduction of the dissertation,
- Chapter 2 is a discussion of the study area which includes location, geology, geohydrology and climate,
- Chapter 3 gives a literature background of AMD and the factors that are involved in the process of AMD,
- Chapter 4 investigates the methods that are necessary for geochemical characterisation that were used during analysis,
- Chapter 5 provides results and discussions, and
- Chapter 6 provides conclusion and recommendation of the overall study.

# CHAPTER 2 STUDY AREA DESCRIPTION

#### 2.1 Study area location

The study was conducted at two mines, namely Mines 1 and 2 that are located about 10 km from each other in the Highveld coal in Mpumalanga Province of South Africa. Figure 1 is the geographically map showing the location of the study sites in Mpumalanga Province of South African and also the Highveld coal fields where the site is located.

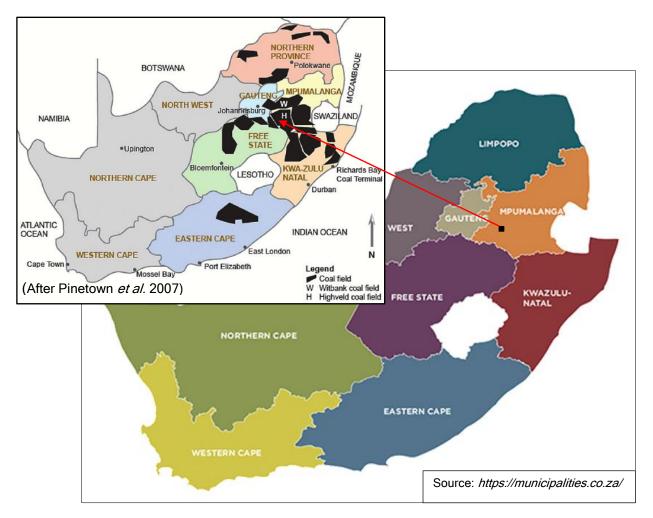


Figure 1: Geographically map showing the location of the study sites in Mpumalanga Province of South African and also the Highveld coal field where the site are located

# 2.2 Geology

# 2.2.1 General

The study sites are located in the Highveld Coalfields of Mpumalanga province. The Coal seams of the Highveld Coalfields are situated in the Vryheid Formation of the Ecca group of

the Karoo Supergroup located in the Main Karoo Basin. The Main Karoo Basin overlies the central and eastern parts of South Africa (Figure 2). The geological formations of the Karoo Basin comprise of a succession of 5 groups (Dwyka, Ecca, Beaufort, Stromberg and Drakensberg groups).

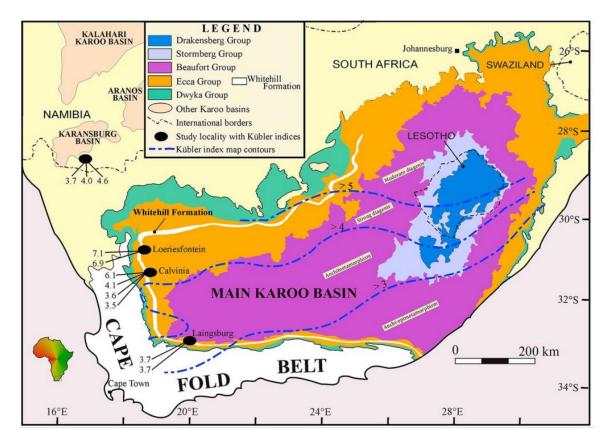


Figure 2 Geological groups of the Main Karoo Basin

#### Source: http://sajg.geoscienceworld.org/content/gssajg/118/4/489/F1.large.jpg

Table 1 show a simplified stratigraphic column of the Karoo Supergroup in the northern portion of the Karoo basin (after South African Committee for Stratigraphy 1980) where the study area is located. In the Highveld Coalfields (Table 1), the Coal seams are situated in the Vryheid Formation of the Ecca group. Thus the geology of the Vryheid Formation is discussed in detail below.

Table 1: Simplified stratigraphic column of the Karoo Supergroup in the northern portion of the Karoo basin (after South African Committee for Stratigraphy 1980)

PERIOD (AGE)	GROUP	FORMATION	ROCK TYPES
Jurassic (150 my)		Drakensberg	Basaltic lava
		Clarens	Fine-grained sandstone
Triassic (195 my)		Elliot	Red sandstone, mudstone
		Molteno	sandstone, mudstone
	Beaufort	Tarkastad	Sandstone, shale
		Estcourt	Sandstone, shale, sub-ordinate coal
Permian (225 my)		Volksrust	Shale, sandstone, sub-ordinate coal
	Ecca	Vryheid	Sandstone, shale, coal
		Pietermaritzburg	Shale
Upper Carboniferous (285 my)		Dwyka	Tillite, varved shale

#### 2.2.2 Vryheid Formation

The stratigraphy of the Vryheid Formation was described by Cadle *et al.*, (1982) as a succession of five coarsening upward sequences which exhibit lateral continuity across the entire region of the Main Karoo Basin. In a complete succession each of the five coarsening-upward sequences starts with fine-grained marine facies grading upwards into coarser delta front and delta plain-fluvial facies (Hancox and Götz 2014).

Vryheid Formation is the only coal bearing formation of the Ecca group and mainly comprises of sandstone, shale and coal sedimentary rocks. The majority of the economically extracted coal in South Africa occurs in rocks of the Vryheid Formation which ranges in thickness from less than 70 m to over 500 m (Hancox and Götz 2014). Six bituminous coal seams are present in a 120m succession of sedimentary lithologies which overlie Dwyka Formation diamictite or pre-Karoo basement and these are described and discussed in detail by Winter (1987). Several coal seams that occur in the Vryheid Formation are associated predominantly with the coarser-grained fluvial facies at the top of each sequence (Hancox and Götz 2014).

Although there are some differences (Winter 1987), the regional stratigraphy in the northern Highveld Coalfield is generally similar to that of the Vryheid Formation in the adjacent central Witbank Coalfield (Le Blanc Smith 1980) and the east Witbank Coalfield (Cairncross and Cadle 1987).

#### 2.2.3 Geohydrology

There are two groundwater systems that are present in the study area, namely the weathered and unweathered Ecca Group/Vryheid Formation aquifers (Azzie, 1999 and Grobbelaar, 2001).

#### 2.2.3.1 Weathered Ecca Aquifer

The weathered formation lies between depths of 5 and 12 m below surface and occurs at the interface of soil and bedrock (Pinetown and Boer, 2006). This aquifer is recharged by rainfall. The percentage recharge to this aquifer is estimated to be in the order of 1 - 3% of the annual rainfall (Kirchner *et al.*, 1991). According to Hodgson and Krantz, 1998, the aquifer within the weathered zone is usually generally low yielded (range 100 - 2000 L/hour) because of its insignificant thickness.

Rainfall that infiltrates into the weathered rock reappears on surface at springs where the flow paths are obstructed by a barrier, such as a dolerite dyke, paleo-topographic highs in the bedrock, or where the surface topography cuts into the groundwater level at streams. It is suggested that less than 60% of the water recharged to the weathered zone eventually emanates in streams (Hodgson and Krantz, 1998) and the rest of the water is evapotranspirated or drained by some other means.

#### 2.2.3.2 Unweathered Ecca Aquifers

Unweathered Vryheid Formation consists of sandstones, siltstones, shales and coal. Groundwater within these sediments will be contained within fractures, joints and bedding planes. The Ogies Dyke is impermeable over much of its length and thus compartmentalizes the groundwater. Of all the unweathered sediments in the Ecca the coal seams have the highest hydraulic conductivity (Grobbelaar 2001). The pores within the Ecca sediments are too well-cemented to allow any significant flow of water.

#### 2.2.4 Climate and Rainfall

Mpumalanga is a province of two halves, namely the high-lying grassland savannah of the Highveld escarpment and the subtropical Lowveld plains. The capital of the Mpumalanga province is Nelspruit and together with the Kruger National park, they both fall in the Lowveld area. The Lowveld has a tropical climate with warm sub-tropical temperatures and experiences high summer rainfalls. Between the month of September and March this area receives approximately rainfall of 620 mm. (*www.southafrica.com*).

# CHAPTER 3 LITERATURE REVIEW

#### 3.1 Introduction

This chapter gives a review of the coal mining activities in South Africa, more specifically in Mpumalanga province. Potential environmental impacts of coal mining activities are discussed with emphasis on Acid Mine Drainage (AMD). AMD is one of the great threats to the water resources in South Africa therefore it is important that the mining industries are able to predict and evaluate the environmental consequences (Usher, 2003). Measures to prevent and reduce the impacts of AMD are discussed.

# 3.2 Coal mining in South Africa

Coal mines in South Africa play an important role in the country's economy, with 90% of all primary energy needs being provided for by coal. Coal is valued for its energy content and since the 1880s, has been widely used to generate electricity. Coal is found in South Africa in 19 coalfields as shown in Figure 3, located mainly in KZN, Mpumalanga, Limpopo and the Free State with lesser amount of coal in Gauteng, North West and Eastern Cape (Jeffrey. 2005). The study area is situated in coal field number 12 in Figure 3. Approximately half of all coal mines in South Africa use open pit mining techniques (opencast) while other half relies on subsurface techniques (underground). Steel and cement industries use coal as a fuel for extraction of iron ore and for cement production. In the US, UK and South Africa, a coal mine together with its structures is a colliery. In Australia, Colliery generally refers to an underground coal mine.

By international standards, South Africa's coal deposit is relatively shallow with thick seams, which make them easier and usually cheaper to mine. At the present production rate, there should be more than 50 years of coal supply left (Dept. of Energy, overview 2014).

#### 3.3 Environmental impacts of coal Mining

While coal mining is a pivotal part of the South African economy as it provides energy and jobs to the country, it is nevertheless associated with serious and damaging environmental impacts. The main concern is the impacts of acid mine drainage on the environment. It is widely accepted that acid mine drainage (AMD) and its potential impact on groundwater resources is one of the most serious environmental concern associated with coal mining (Brady *et al.*, 1997; Rose and Cravotta, 1998; Bell *et al.*, 2001).

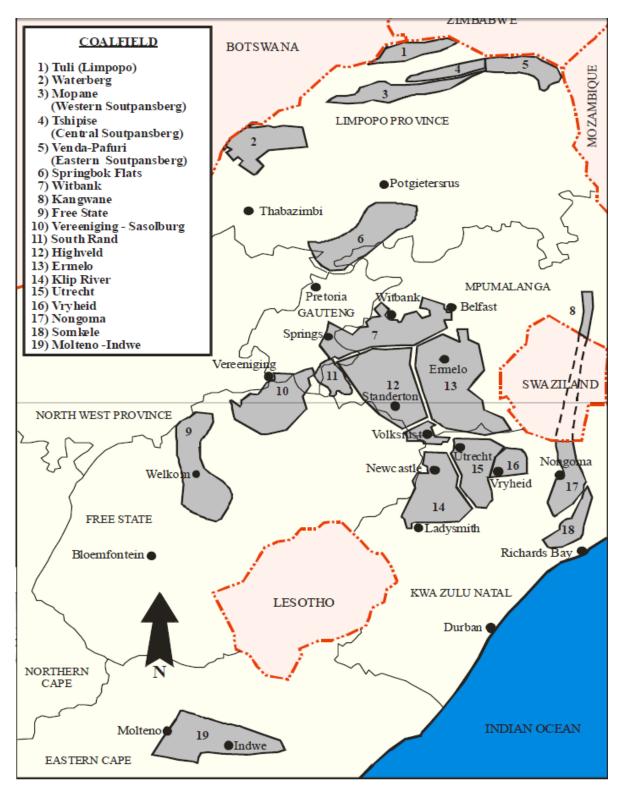


Figure 3: South Africa Coalfields (Jeffrey 2015)

Hydroxide occurs over long distances. Rain penetrates overburden and acquires a certain alkalinity, usually by dissolution of calcite (Caruccio & Geidel, 1978). The amount of alkalinity acquired is determined by the PCO<sub>2</sub> of the water and the solubility of calcite which at first is neutralised by the alkalinity in the groundwater. If the acidity generated is greater than the initial alkalinity of water, all the alkalinity will be consumed resulting in acid water. If sufficient

oxygen is present, the amount of acidity generated is determined by the amount of reactive pyrite in the coal (Drever, 2002).

In the absence of mining, acid waters are uncommon because dissolved oxygen in gas is insufficient to produce acidity greater than the alkalinity of the groundwater. When mining occurs, additional oxygen is introduced and water movement through the system is accelerated. Oxidation is no longer limited by groundwater transport of oxygen and acidity may result. The bacteria that catalyse the acidity-producing reaction thrive only under acid conditions so once acidity is initiated, acid production becomes more rapid and the acidity problem increases rapidly (Drever, 2002).

Mining practices, present and past, cause environmental problems that can damage ecosystems and human health. Mining disturbs geologic formations that took millions of years to form; likewise, related natural systems and processes are disturbed, e.g. hydrology. Once disruption has taken place a variety of problems may arise, from physical hazards to pollution of water and soil. The most severe and widespread environmental problems almost always have to do with water. Hodgson *et al.*, (2001) pointed out that through advanced planning much can be done to minimise the impact on the mine water during and after mining.

### 3.4 Treatment of mine sites

Treatment of mine sites generally requires pH adjustment, oxidising or reducing (redox) conditions and/ or stabilisation of waste (Costello, 2003). There are two types of treatment technologies:

#### 3.4.1 Traditional treatment

Traditional treatments rely on conventional, well-recognised technology to raise pH or create redox conditions. These traditional conditions are Water treatment plant, relocation of waste, covering of waste piles, water diversion tactics and vegetation.

#### 3.4.2 Wastewater Treatment Plant

In this type of treatment, waters are removed from their course, treated and then discharged. There are other treatment that are similar to traditional wastewater treatment plant i.e. Oxidation Dosing with Alkali and Sedimentation (ODAS), sulfidisation, sorption and ion exchange and membrane processes like filtration and reverse osmosis. One of the advantages of this treatment is precision, which means it is useful for active mining sites because of its frequent changing water characteristics (Younger *et al.*, 2002).

#### 3.4.3 Relocation of waste site

Wastes are shipped off-side to landfills and treatment plants, which is costly. Instead of shipping off the waste, this can be avoided by covering the waste with multiple layer of plastic, cement, soil, rock, vegetation etc. By doing this, solid materials high in metals and acid materials, will not be exposed to the elements and will not cause typical problems associated with mine waste (Pioneer Technical Services, 2002).

#### 3.4.4 Divergent of Water

Another tactic is by controlling waste flow near a waste pile by installing trenches and culverts to divert water from the pile.

#### 3.4.5 Innovative treatment

Innovative treatment is done by wide range of technologies e.g. Limestone drains, Constructed wetlands, etc.

#### 3.4.6 Limestone drains

In this method of treatment, the water is allowed to settle in the pond or wetland to allow metals to precipitate and settle. The only problem with this method is that it causes the limestone to become inactive and causes clogging of the drain (Cravotta *et al.*, 2002).

#### 3.4.7 Constructed wetlands

Wetlands are capable of treating water and retaining toxics forms the basis of most passive treatment technologies. There are 2 types of wetlands used to treat mine drainage, namely aerobic and anaerobic. If metal of concern is iron, an aerobic wetland is used as treatment. Anaerobic wetland generally consists of organic substrate, often compost and can be mixed with lime to increase alkalinity (USEPA, 1994).

#### 3.5 Acid Mine Drainage

Acid Mine Drainage is a widespread phenomenon in the mining industry worldwide affecting the quality of water at many South African Collieries (Vermeulen and Usher, 2006). Acid Mine Drainage is produced when sulphide-bearing material is exposed to oxygen and water (Akcil *et al.*, 2006). This result in the generation of sulphates, metals and acidity that can have numerous environmental consequences (Usher *et al.*, 2003)

#### 3.5.1 Oxidation of metal sulphide

Acid is generated at mine sites when metal sulphide minerals are oxidized. Oxidation of these minerals and the formation of sulphuric acid is a function of natural weathering processes. Oxidation of sulphide minerals consists of several reactions. Each sulphide minerals has a different oxidation rate, example: marcasite and framboidal pyrite will oxidise quickly while crystalline pyrite will oxidise slowly.

Minerals	Composition
Pyrite	FeS <sub>2</sub>
Marcasite	FeS <sub>2</sub>
Chalcopyrite	CuFeS <sub>2</sub>
Chalcocite	Cu <sub>2</sub> S
Sphalerite	ZnS
Galena	PbS
Millerite	NiS
Pyrrhotite	Fe <sub>1-x</sub> S (where 0 <x<0.2)< td=""></x<0.2)<>
Arsenopyrite	FeAsS
Cinnabar	HgS

Source: Ferguson and Erickson 1988

$$2FeS_{2(s)} + 2H_2O + 7O_2 \rightarrow 4H^+ + 4SO_4^{2-} + 2Fe^{2+}$$

Equation 1

In Equation 1,  $S_2^{2^-}$  is oxidised to form hydrogen ions and sulphate, the dissociation products to sulphuric acid in solution. Soluble  $Fe^{2^+}$  is also free to further react with oxygen (Equation 2). Oxidation of the ferrous ion to ferric ion occurs more slowly at lower pH values:

$$4Fe^{2+} + O_2 + 4H^+ \to 4Fe^{3+} + 2H_2$$

Equation 2

If the ferric ion is formed in contact with pyrite the following reaction can occur (Equation 3), dissolving the pyrite.

$$2FeS_{2(s)} + 14Fe^{3+} + 8H_20 \rightarrow 15Fe^{3+} + 2SO_4^{2-} + 16H^+$$

Equation 3

This reaction generates more acid. Ferric iron precipitates as hydrated iron oxide as indicated in the following reaction (Equation 4):

$$Fe^{3+} + 3H_2O \rightarrow Fe(OH)_{3(s)} + 3H^+$$

Equation 4

Fe  $(OH)_3$  precipitates and is identifiable as the deposit of amorphous, yellow orange or red deposit on stream bottoms (Yellow boy) as shown in Figure 4.



Figure 4: Precipitation of a yellow boy at a South African Colliery (Usher (2003))

#### 3.5.2 Contributing factors to AMD

The potential for a mine to generate acid and release contaminants is dependent on many factors and is site specific. Ferguson and Erickson (1988) identified primary, secondary and tertiary factors that control acid drainage.

#### 3.5.2.1 Primary Factors

Primary factors required for the generation of AMD include sulphide minerals, water, oxygen, and ferric ion, bacteria to catalyse the oxidation reaction. Both water and oxygen are necessary to generate acid drainage. Water serves as both a reactant and a medium for bacteria in the oxidation process. It also transports the oxidant products and oxygen is required to drive the oxidation reaction (Ferguson and Erickson, 1988).

#### 3.5.2.2 Secondary Factors

Secondary factor act to either neutralise the acid production by oxidation of sulphides or may change the effluent character by adding metals ions mobilised by residual acid. Neutralisation of acid by the alkalinity released when acid react with carbonate minerals is an important mean of moderating acid production. The most common neutralising minerals are calcite and dolomite (Ferguson and Erickson, 1988).

#### 3.5.3 Tertiary Factors

Some of tertiary factors affecting acid drainage are the physical characteristics of the material, how acid generating and acid neutralising material are placed, waste and the hydrologic regime in the vicinity. The physical nature of the material such as particle size, permeability and physical weathering characteristics, is important to the acid generation potential. Particle size is a fundamental concern since it affects the surface area exposed to weathering and oxidation (Ferguson and Erickson, 1988).

Very coarse grain material as is found in waste dumps exposes less surface area but may allow air and water to penetrate deeper into the unit exposing more material to oxidation and ultimately produce more acid. In contrast, fine grain material may retard while air and much fined material may limit water flow, however, finer grains expose more surface area to oxidation. The relationships between particle size, surface area and oxidation play a prominent role in acid prediction method.

The hydrology of the area surrounding mine workings and waste units is also important in the analysis of acid generation potential. When acid generation material occurs below the water table, the slow diffusion of oxygen in water retards acid production. This is reflected in the portion of pits or underground workings located below the water table. Where mine walls and underground workings extend above the water table, the flow of water and oxygen in joints may be a source of acids (USEPA, 1994).

# 3.6 Assessment of the AMD potential

Application of acid base accounting procedures has made it possible to quantify the potential of a particular rock or coal sample to produce acid or alkaline waters under mine drainage conditions. A detailed explanation of the experimental procedure followed in applying this technique to the present study can be found in Usher *et al.*, (2003).

The method involves:

#### 3.6.1 Hydrogen Peroxide (H<sub>2</sub>O<sub>2</sub>)

- Adding 120 or 80ml of H<sub>2</sub>O<sub>2</sub> reagent to 1 to 4g of a pulverised sample and allowing it to vivacious or give off bubbles.
- The final pH is measured once the bubbles has ceased or come to an end.
- The supernatant liquid is then analysed for sulphate.
- The %S (total) can also be determined by Leco element analyser if required.
- The actual acid produced during the oxidation of pyrite by H<sub>2</sub>O<sub>2</sub> is called potential acidity.

• The reaction (Equation 5) which represents the complete oxidation of pyrite is as follows:

$$FeS_2 + 3.75O_{2(aq)} + 1.5H_2O_{(aq)} \leftrightarrow Fe(OH)_{3(s)} + 2SO_4^{2-} + 4H^+$$

Equation 5

The overall pyrite oxidation reaction in Equation 5 will produce 4 moles of  $H^+$ , 1 mole of Fe(OH)<sub>3(s)</sub> and 2 moles of SO<sub>4</sub><sup>2-</sup> for each mole of pyrite oxidised.

#### 3.6.2 Neutralising Potential (0.06N of H2SO4)

The neutralising potential is determined by adding 20 ml of 0.06 N of standardised  $H_2SO_4$  to 5 g of a pulverised sample. The pH of the mixture must be 2.5 after 24 hours before back titration to a pH of 7. Back titration is carried out with 0.06 NaOH. If the pH is still above 3 after 24 hours, additional acid is added and the process is repeated until the correct pH is obtained (Usher *et. al.* 2003)

The solubility of calcite is different for open and closed system and thus the acid potential (AP) and neutralising potential (NP) for both cases was determined.

**OPEN SYSTEM** - In an open system, carbon dioxide  $(CO_2)$  dissolves into the atmosphere. Therefore, 1 mole of FeS<sub>2</sub> is neutralised by 2 mole of CaCO<sub>3</sub>.

**CLOSED SYSTEM** - In a closed system, carbon dioxide (CO<sub>2</sub>) is dissolved in the water and carbonic acid or  $H_2CO_3$ , is formed.

It follows that the 4 mole of CaCO<sub>3</sub> is needed to neutralise 1 mole of FeS<sub>2</sub> (Pyrite).

#### 3.6.3 Acid Potential (AP)

AP is a measure of the potential of a sample to generate acidity. The amount of calcite required to neutralise a given amount of acid mine drainage depends on the behaviour of  $CO_2$  during neutralisation and on the pH reached. If the acid-mine drainage is to be neutralised to pH 6.3 or above, then the following reaction (Equation 6) may be written:

$$FeS_2 + 2CaCO_3 + 3.75O_2 + 1.5H_2O \leftrightarrow Fe(OH)_3 + 2SO_4^{2-} + 2Ca^{2+} + 2CO_2$$

Equation 6

For each mole of pyrite that is oxidised, two mole of calcite are required for acid neutralisation (Equation 6). On a mass ratio basis, for each gram of sulphur present, 3.125g of calcite is required for acid neutralisation.

When expressed in parts per thousand (ppt) of spoil, for each 10 ppt of sulphur (S) present 31.25 ppt of calcite is required for acid neutralisation. The stoichiometry in the previous

equation is based on the exsolving of  $CO_2$  gas out of the spoil system. In closed system,  $CO_2$  is not exsolved and additional acidity from carbonic acid is generated. Cravotta *et al* (1990) proposed that up to 4 mole of calcite might be needed for acid neutralisation. Twice as much calcite would be required for acid neutralisation in a closed system, compared to an open system. On a mass basis for each 10ppt of sulphur present, 62.5ppt of calcite is needed for acid neutralisation in one thousand tonnes of spoil (Cravotta *et.al*, 1990).

Results obtained from the laboratory experimental procedure are used in calculating the acid potential (AP), neutralising potential (NP) and the net neutralising potential (NP) as follows:

1. 
$$AP = \left(\frac{SO_4\left(\frac{mg}{l}\right)}{weight(g)} / 1000\right) \times ml H_2 0 \text{ or } H_2 O_2 = \frac{kgSO_4}{t \text{ of sample}}$$

2. 
$$AP(Open)(CaCO_3kg/t) = \frac{SO_4kg/t}{48} \times 50$$

3. 
$$NP\left(\frac{CaCO_{3}kg}{t}\right) = \frac{(NH_{2}SO_{4} \times ml \ acid) - (N \ NaOH \ \times ml \ alkali)}{weight \ (g) \times 50}$$

Thus, the NNP is determined by subtracting the acid potential from the neutralising potential.

In a closed system

NNP (Closed) = NP - AP (Closed) (Hodgson & Krantz, 1998)

There is various type of screening criteria used to interpret acid-base accounting results. In this study the NNP was used as screening criteria. Research has shown that there is a range from -20 to 20kg/t CaCO<sub>3</sub> where a sample can become acidic or remain neutral. Thus a sample with a NNP<20 is potentially acid generating and a sample with a NNP>20 might not generate acid (Usher *et. al*, 2003).

#### 3.7 Prevention of acid generation

The main strategies to prevent acid generations are prevention or minimization of water circulation through acid generating material by covering with an impermeable cap, which may simply be a soil in relatively arid climate. Another approach is to dispose the materials under water, for example in a flooded mine pit. The solubility of oxygen in water is quite low, so sulphide oxidation in the saturated zone is generally low, limited by the availability of oxygen (Drever, 2002).

### CHAPTER 4 METHODS AND MATERIALS

# 4.1 Introduction

Static method was used to determine whether the samples analysed has the capacity to generate or neutralise an acid and it was also used as a screening method to determine the difference between the acid generating capability and the acid neutralising potential of samples analysed. A description of sampling method from the core logs is provided. The static Acid-Base method (ABA) is used to determine the acid mine drainage potential of the samples. The mineralogical (X-Ray Diffraction (XRD)) and chemical (X-Ray Fraction (XRF)) information is used in conjunction with the ABA to give evidence of the ABA results.

#### 4.2 Collection of geological samples

The geological samples were collected from borehole logs (Figure 5) at Mine 1 and Mine 2 which were drilled at different areas. Appendix 1 and Appendix 2 shows how the samples were collected at different depths. Most of the samples consist of sandstone, siltstone, gritstone, dolerite, carbonaceous shale and coal.

#### 4.2.1 Mine 1 Samples

Mine 1 sample were collected from 13 borehole logs, namely: F142441, F142446, F142471, O105016, P110030, P110087, R146043, T139228, W569001, Y106148, Z145168, Z145198, and Z145199. Table 3 shows how many samples were collected from each borehole log.

Borehole Name	Samples collected	Depths (m)
F142441	6	1.7 - 128.56
F142446	9	5.8 - 161.50
F142471	10	4.68 - 109.90
O105016	9	7.3 - 85.56
P110030	14	3.8 - 141.55
P110087	11	6.9 - 140.18
R 146 043	14	5.33 - 170. 34
T139228	11	2.2 - 145.24
W569001	8	2.4 - 90.8
Y106048	5	5.95 - 75.27
Z145168	8	1.6 - 142.08
Z145198	6	3.33 - 184
Z145199	7	0 - 162.3
Total samples collected		118

Table 3: Mine 1 Borehole logs and number of sample collected

Coal samples are also included in the 118 samples collected and were collected from 4 borehole logs; namely: F142471 (4 samples), P110030 (4 samples), P110087 (2 samples) and W569001 (3 samples):

- F142471 coal samples were collected from: No4 lower coal seam (C4L)
  No3 coal seam (C3)
  No2 coal seam (C2) and
  No5 coal seam (C5)
- P110030 coal samples were collected from: No5 lower coal seam (C5L)
  No4 lower coal seam (C4L)
  No3 coal seam (C3) and
  No2 coal seam (C2)
- P110087 coal samples were collected from: No5 lower coal seam (C5L) No4 lower coal seam (C4L)
- W569001 coal samples were collected from: No4 lower coal seam (C4L)
  No3 coal seam (C3)
  No2 coal seam (C2)

# 4.2.2 Mine 2 Samples

Mine 2 samples were collected from 4 borehole logs, namely: R100001, R100002, Z124027, Z124029 and Z124030. Table 4 shows how many samples were collected from each borehole log:

Table 4: Mine 2 Borehole logs and number of samples collected

Borehole Name	Samples collected	Depths (m)
R 100 001	13	1.62 - 143.1
R 100 002	8	14.1 - 124.4
Z124027	21	4.13 - 200.16
Z124029	13	6.9 - 174.53
Z124030	16	11.6 - 185.66
Total of samples collected		71

Coal samples are also included in the 71 samples collected; each borehole logs contain coal samples. R100001 consist of 2 coal samples (C5L, H &M and C4L), R100002 have 1 coal samples (C4L), Z124027 consist of 3 coal samples (C5L, C4L and C4L), Z124029 have only 1 coal sample (C4L) and Z124030 have 4 coal samples (C5H, C5L, C4H & C4L). These samples were collected to determine the acid base potential, whole rock chemical analysis and to supply the mineralogy of the analysed samples.



Figure 5: Mine 1 and Mine 2 Borehole core logs

# 4.3 Mineralogical and whole rock analysis

The X-Ray Diffraction (XRD) technique was used to determine the mineralogy and the X-ray Fluorescence (XRF) technique was used to determine the chemical analysis or whole rock analysis of the samples analysed. Both the XRD and XRF assisted in understanding the process of Acid Mine Drainage in both areas.

# 4.3.1 Mineralogical identification

Mineralogical analysis of the rock sample was done by using X-Ray Diffraction (XRD) technique. The XRD involves the scattering of x-rays by minerals crystals with accompanying variation in intensity due to interferences effects. The XRD analysis evaluates the crystal structure of the materials by passing x-rays through and recording the diffraction or scattering image of the rays. It has been established as probably the most important, convenient and unambiguous technique applied to the study of soil and overburden

mineralogical composition (Bish, 1994). The occurrence of pyrite and other sulphide minerals as well as calcite and dolomite can be determined by x-ray diffraction.

#### 4.3.2 Whole rock analysis

The whole rock analysis is determined by using X-Ray Fluorescence (XRF) technique, which is used to determine the chemical composition of a rock sample by analysing several elements (Taggart *et al*, 1987). The oxides in the sample were identified using Panalytical Axios XRF machine, the machine has a Rh end window tube, with 4kW anode (consisting of Rh) and a W cathode (filament). It has sequential wavelength dispersive XRF, which measures one element at a time and it measures the wavelength of the X-rays instead of the energy. This machine also has the additional crystals for diffraction when compared to the energy dispersive machines which splits the peaks of the elements for better identification. It also consists of two detectors that are attached to it, namely: a flow detector contain P10 gas (for X-rays with low energies). These two detectors can be used for intermediate energies to enhance sensitivity and to analyse a wider range of transition metals (Bruker 2006).

# 4.4 Static Method

Static method which is one of the methods of Acid-Base Accounting (ABA) was used to determine acid-base potential for investigated geochemical layers. Static method provide a rough indication of the acid generating potentials of the various lithological units, it determines the difference between the acid-generating capability and the acid neutralising potential of a particular sample or set of samples.

# 4.5 Determination using acid-base accounting method

Acid-base Accounting (ABA) is an excellent first-order tool to determine whether or not mine waste has the potential to form acidic drainage (Usher *et al*, 2003). The tool was developed by Richard Smith and his associates at the West Virginia University in the late 1960's and was designed to evaluate the acid producing capability of coal mine wastes and can also be used to determine if the rock samples has acid or base producing potential. The following tests are used for ABA method:

#### 4.5.1 Paste or Initial pH

5g of sample is measured to 50ml of distilled water and stirred for 1 hour. The solution is then left to stand 24 hours to allow the solubility reaction to be more complete. The initial pH is measured and recorded after 24 hours. The solution is filtered and the leachate analysed by ICP-OES Inductively Coupled Plasma Optical Emission Spectroscopy to determine the

water soluble parameters/elements. The results obtained provide the current state of the sample and the pH recording gives an indication of whether the sample is acidic or basic.

#### 4.5.2 Acid potential using Hydrogen Peroxide (H<sub>2</sub>O<sub>2</sub>)

80 ml of 30%  $H_2O_2$  is added to 2g of pulverised sample in 5ml increments. The sample is then allowed to react until boiling stops. The higher the sulphur content in the sample, the more violent it will react with the hydrogen peroxide. The solution is left to stand for 24 hours to cool to room temperature; the pH is then recorded after 18-24 hours as oxidised pH. The solution is filtered and analysed for sulphate and other ions.

Hydrogen peroxide is used in this method for its ability to oxidise sulphides (e.g. pyrite) present in a rock or coal sample to sulphate (Price, 1997).

#### 4.5.3 Acid Leachable elements/products

5g of sample is measured and mixed with 50ml of  $0.1N H_2SO_4$  and stirred for 1 hour. The solution is then left to stand for 24 hours to allow the solubility reaction to complete. The pH must be recorded only when it is less than 2.5, if not, few drops of  $H_2SO_4$  must be added and left for another 24 hours. Samples are stirred thoroughly after addition of any acid. The pH of less than 2.5 is measured and recorded after 24 hours. The solution is filtered and analysed by ICP-OES.

# 4.5.4 Neutralising potential using Sulphuric Acid (H<sub>2</sub>SO<sub>4</sub>)

20 ml of  $\pm 0,06N$  H<sub>2</sub>SO<sub>4</sub> is added to 1g of sample. The pH of this slurry must be below 2.5 after 24 hours, before back titration to pH7 is done with  $\pm 0.06$  NaOH. If the pH is >2.5, more H<sub>2</sub>SO<sub>4</sub> is added and the sample is left another 24 hours for the reactions to complete. If the pH is not below 2.5 the next day the above procedure is followed again until the pH is below 2.5, then the sample is titrated to a pH of 7 with the standardised Sodium Hydroxide (NaOH).

The reason for using  $H_2SO_4$  for this method is that Acid Mine Drainage contains sulphuric acid; therefore, it will provide a better simulation of the field condition (Price, 1997).

# 4.6 The advantages and the limitations of acid base accounting

Acid-Base Accounting is simply a screening process. It provides no information on the speed or kinetic rate with which acid generation or neutralisation will proceed and because of this limitation, the test work procedures used are referred to as Static procedures Ziemkiewicz and Meek, 1994)

#### 4.6.1 The primary advantages of Acid Base Accounting method are:

- a. Short turn-around time for sample processing.
- b. Low cost.

- c. Relatively simple interpretation of results
- d. Interpretation is based on decades of international research or experience.
- e. Correlation to field has been shown by case studies.
- 4.6.2 The limitations of Acid Base Accounting (ABA) are as follows:
  - a. It only provides a possibility of occurrence.
  - b. Reaction rates are ignored (ABA generally tests the fast reacting species; slow reacting neutralising species will usually not prevent acidification.)
  - c. Assume instant availability of reactive species.
  - d. Simple reaction stoichiometry is assumed.
  - e. Size effect is ignored.
  - f. Extrapolation to the field is uncertain when volumetric calculations cannot be made.

Despite all these limitations ABA is a very important component of the Acid Base Accounting, Techniques and Evaluation (ABATE). ABATE is also called prediction wheel and has the following components:

- a. Onside monitoring data
- b. Mineralogy
- c. Static Test (ABA)
- d. Total Metals & Whole Rock
- e. Geochemical Modelling
- f. Hydraulic tests
- g. Laboratory Kinetic Tests
- h. Field tests

The prediction wheel shows that many different aspects are required to arrive at the eventual answer, depending on the type of information required and which of the key questions need answering and to what level of accuracy (Usher, 2003). Predictive capability is best achieved by using a combination of data set and method rather than by relying on one procedure (Cravotta, 1997).

# 4.7 Screening criteria

The criteria to be used are:

- a. Net Acid Generation (NAG) test
- b. Net Neutralising Potential (NNP)
- c. Neutralising Potential Ratio (NPR)
- d. % S & NPR

#### 4.7.1 Net Acid Generation (NAG) Test pH

In this test, the value will be obtained from acid potential test as explained above. This can serve as a rough guideline but not as stand-alone criteria in categorising the sample (Price, 1997). The table below (Table 5) will be used to indicate the likelihood of net acid generation of the sample upon oxidation.

Table 5: Net Acid-Generating (NAG) Test pH guideline

Final pH NAG Test	Acid Generating Potential
>5.5	Non-acid generating
3.5 to 5.5	Low risk acid generating
<3.5	High risk acid generating

The pure deionised water in equilibrium with  $CO_2$  will have a pH value of around 5.69. Therefore, anything above this should be non-acid generating.

#### 4.7.2 Net Neutralising Potential (NNP)

Research has shown that there is a range from -20 to 20 kg/t  $CaCO_3$  where the sample can become acidic or remain neutral. Where Net Neutralising Potential (NNP) = Neutralising Potential (Kg/t  $CaCO_3$ ) - Acid Generating Potential (Kg/t  $CaCO_3$ )

The criteria are as follows:

If NNP < 20, the sample has the potential to generate acid

If NNP > 20, the sample has the potential to neutralise acid

# 4.7.3 Neutralising Potential Ratio (NPR)

The ratio of NP value to AP value, or Neutralisation Potential Ratio (NPR=NP/AP), and the acid generating is considered uncertain if the samples have a NPR of less than 4:1 (Usher, 2003).

Potential for Acid Rock Drainage (ARD)	Initial NPR Screening Criteria	Comments	
Likely	<1:1	Likely AMD	
Possibly	1:1 - 2:1	Possibly AMD generating if NP is insufficiently reactive or is depleted at a faster rate than sulphides.	
Low	2:1-4:1	Not potentially AMD generating unless significant preferential exposure of sulphides along fracture planes, or extremely reactive sulphides in combination with insufficiently reactive NP.	
None	>4:1	No further AMD testing required unless materials are to be used as a source of alkalinity.	

Table 6: Guidelines for screening criteria based on ABA (Price, 1997)

#### 4.7.4 % S & NPR

It has been shown that for sustainable long-term acid generation, at least 0.3 Sulphide-S is needed. Values below this can yield acidity, but this is likely to be only of short-term significance. Using this and the NPR values, another set of rules can be derived as follows (Price, 1997):

- Sample with less than 0.3% Sulphide-S are regarded as having insufficient oxidisable Sulphide-S to sustain acid generation,
- NPR ratio of >4:1 are considered to have enough neutralising capability,
- NPR ratio of 3:1 to 1:1 are considered inconclusive, and
- NPR ratios below 1:1 with sulphide-S above 0.3% are potentially generating.

# 4.8 Leach test

Leaching is the process by which soluble constituents are dissolved from a solid material (such as rock, soil, or waste) into a fluid by percolation or diffusion (Washington State Department of Ecology 2003). A leaching test can be conducted in the laboratory or field. Leach tests are either static or kinetic. Static leach tests are conducted over a short term (minutes) and are relatively less expensive than kinetic tests, which requires long term (weeks to years) testing. In this study, static laboratory tests were conducted to help identify the elements (some toxic) that go into solution in the overburden, coal seam and interburden formations.

The samples were leached with water, hydrogen peroxide and sulphuric acid. For water leach the samples were collected from the leachate prepared in section 3.5.1. The solution was filtered and analysed by inductively coupled plasma (ICP) for major and trace elements.

For hydrogen peroxide leach the samples were collected from the leachate prepared in section 3.5.2. The solution was filtered and analysed by ICP for major and trace elements.

For sulphuric acid leaching, samples were collected from the leachate prepared in section 3.5.3. Major and trace elements were determined (ICP) on these filtered samples.

#### CHAPTER 5 RESULTS AND DISCUSSIONS

#### 5.1 Mineralogical and whole rock analysis results

A total of 118 (Mine 1) and 71 (Mine 2) samples were collected respectively from the borehole core logs for mineralogy and whole rock analysis. Out of 118 samples (Mine 1), 13 were coal samples and out of 71 samples (Mine 2), 11 were coal samples. Minerals were detected by using XRD and classification of minerals was presented according to dominance as indicated in Table 7.

KEY		
XX	Dominant Mineral	>50%
Х	Major Mineral	20-50%
хх	Minor Mineral	10-20%
х	Accessory	2-10%
<x< th=""><th>Trace Mineral</th><th>&lt;2%</th></x<>	Trace Mineral	<2%

Table 7: Percentages for XRD interpretation

The whole rock analysis was performed using X-ray fluorescent spectrometry (XRF). The results of elements analysed are given in wt. %.

#### 5.1.1 Mine 1 mineralogical Analysis

The mineral components in the samples analysed are summarised in Appendix 3. Quartz is the dominant mineral in most samples with plagioclase dominating in few samples. Pyrite appeared as minor to trace mineral with a percentage of <2% to 20% only in sample T103 (W569001) which is a coal sample that appear as major mineral with a percentage of 20-50%. Most of the samples consisted of calcite which appeared as minor and trace mineral with a percentage of <2% to 20% and dolomite appeared only in 14 samples out of 118 with a percentage of 2% to 10%. Both calcite and dolomite were found to be the most common neutralising minerals. Siderite which is part of carbonate minerals existed in most of the samples with a percentage of <2% to 20% but appeared in one sample (T100) as major mineral. K-feldspar and koalinite existed in almost all samples from major to accessory. Magnesite appeared in 2 samples only with the concentration of 2 to10%. Analcine appeared only in one sample (T111), which is a coal sample, as major mineral. Most of the samples were found to originate from coal, sandstone and siltstone.

#### 5.1.2 Mine 2 mineralogical Analysis

Quartz was the most dominant mineral in all the samples (Appendix 4) with a concentration ranging from 10 to 50%. Plagioclase was dominant in only 4 samples (SF 11, SF 38, SF 50 and SF 57). Only 20 samples from 71 contained pyrite with concentrations ranging from 2 to 20% (accessory to minor). Pyrite was dominant only in sample F68 which was a coal sample and it appeared as trace mineral with a percentage of <2% in samples F6 and F52. The following minerals appeared only in one sample: alunite (SF 66), andalusite (SF 60), amphibole (SF 44) and ilmenite (SF 10) with concentration in the range of 2 to 10%. Anatase and magnetite existed in samples SF 34 and 38 and SF 54 and 57 with a concentration ranging from 2 to 20%. Kaolinite which is a clay material appear in most of the samples as major mineral with a concentration ranging from 2 to 20%. Kaolinite which is a clay material appear in most of the samples as major mineral with a concentration ranging from 2% to 20%. The mica mineral in sample SF 60 appeared as dominant and in sample SF 13, 28 & 59 as major.

#### 5.1.3 Mine 1 and Mine 2 whole rock analysis

Samples from mine 1 (Appendix 5) and mine 2 (Appendix 6) consisted mainly of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> which have the higher wt. % than the other elements. SiO<sub>2</sub> for mine 1 has an average of 57.5 wt.% and mine 2 has SiO<sub>2</sub> of 56.6 wt.%. Al<sub>2</sub>O<sub>3</sub> being the second highest with a concentration of 12.9 wt.% and 11.6 wt.% for mine 1 and mine 2 respectively. Fe<sub>2</sub>O<sub>3</sub> in both list have almost the same concentration, i.e. 5.2 wt.% for mine 1 and 5.1 wt.% for mine 2. The concentration of the rest of the elements was small.

#### 5.1.4 Conclusion

Quartz was found to be in all the samples ranging from 2 to 50% at mine 1 and mine 2. Eighty-five percent of the samples contained koalinite from trace to major concentration in mine 1 and only 18% in mine 2. Mine 1 consisted of calcite and dolomite with percentages of 55 and 44 respectively. Mine 2 consisted of calcite with percentage of 43 ranging from trace to accessory. For the whole rock analysis, SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> were present in both mines as major oxides which is similar to quarts and kaolinite and the presence of calcite and dolomite was supported by the availability of CaO and MgO. Only 22% of the samples contained pyrite ranging from trace to major concentration in mine 1 while at Mine 2, 28% ranged from trace to accessories. It can therefore be concluded that mine 1 has a higher concentration of minerals that can neutralise acid. Mine 2 has more samples that consisted of calcite compared to pyrite (trace to accessory).

# 5.2 Static test results

Acid-Base Accounting was used as a qualitative tool to determine the acid and neutralising potential of the samples. The release of metals from the spoil, rock and coal samples were also determined. The NNP, NPR and S% from the results obtained, are summarised below for each mine. The ABACUS programme (Usher 2003) was used to plot the data. The data used to plot the ABA results is presented in Appendix 7 and Appendix 8.

#### 5.2.1 Mine 1 analysis

13 borehole core samples were collected from different areas for Mine 1. 118 samples were selected and analysed. The samples collected consist of sandstone, siltstone, coal and other geological formations shown in Appendix 7.

# 5.2.1.1 Net Acid Generating Test (NAG) pH for Mine 1

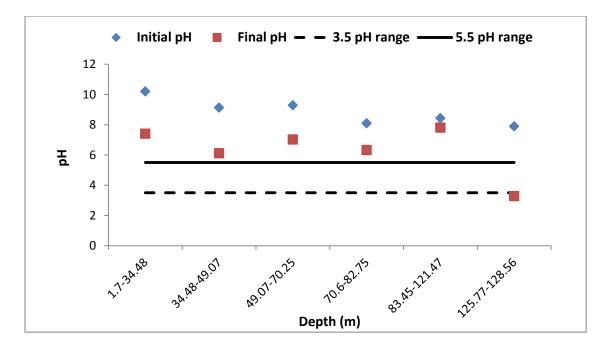
The results for the NAG test are presented in Table 8 to 20. The interpretation of the results helped to assess whether the samples were acid or non-acid generating. The graphs below each table of results were plotted to show the number of samples with the pH value of 5.5, between 5.5 and 3.5 and below 3.5. These graphs were used for further interpretation. The information from Table 5 above was used to indicate the likelihood of net acid generation of the sample upon oxidation.

#### 5.2.1.1.1 F142441 Borehole Core

Six samples were collected from F142441 borehole core log. The samples were taken from depths 1.7m to 128.56m. Table 8 shows the results of the NAG pH test with interpretations as from the ABACUS program. Only one sample (TC35) has a final pH's that is below 3.5 (Figure 6), which means it has a high risk of acid generation. The calculated NNP for this sample was -2.99 kgCaCO3/tonne (Appendix 9). Five samples (T18, T75, T38, T83 and T7) have low risk of generating an acid therefore it indicates that there are enough neutralising minerals to buffer the pH.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC18	1.7-34.48	DO	10.2	7.4	Lower Acid Risk
TC75	34.48-49.07	SST	9.13	6.11	Lower Acid Risk
TC38	49.07-70.25	SST,C	9.29	7.02	Lower Acid Risk
TC83	70.6-82.75	SST, SLT, SH	8.09	6.32	Lower Acid Risk
TC7	83.45-121.47	SST, SLT, SH	8.43	7.81	Lower Acid Risk
TC35	125.77-128.56	SST, SLT	7.89	3.28	Higher Risk Acid Generation

#### Table 8: Results of NAG pH test on F142441 core samples





#### 5.2.1.1.2 F142446 Borehole

Nine samples were collected from F142446 borehole core log. The samples were taken from depths 5.8m to 161.50m. Table 9 below shows the results of the NAG pH test with interpretations as from the ABACUS program. Samples TC51 and TC41 have final pH's that are below 3.5 (Figure 7), which means they have a high risk of acid generation. Only TC66 sample is between pH of 3.5 and 5.5 and is regarded as having medium risk of acid generation, whereas the remaining samples are non-acid generating (>5.5). Most of the samples in core F142446 show low acid risk, indicating there are enough neutralising minerals to buffer the pH. Therefore, there will be a low risk of acid generation for this core sample.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation		
TC66	5.8-9	DO, SST	8.46	4.94	Medium Risk Acid Generation		
TC57	9.15-34.95	SST, SLT	9.34	6.78	Lower Acid Risk		
TC 62	34.95-70.5	DO	9.99	7.20	Lower Acid Risk		
TC10	70.5-82.44	SST, SLT	9.79	7.11	Lower Acid Risk		
TC4	82.44-91.10	SST	9.70	6.11	Lower Acid Risk		
TC51	91.10-112.10	SST, SLT	8.78	3.34	Higher Risk Acid Generation		
TC41	112.10-113	C, SLT, CSH	8.84	1.63	Higher Risk Acid Generation		
TC33	113.12-155.3	SST, SLT, C	8.30	7.04	Lower Acid Risk		
TC87	155.3-161.50	SST, SLT, C	9.20	5.64	Lower Acid Risk		
DO - Dol	DO - Dolerite, SST - Sandstone, SLT - siltstone, C - coal, CSH -carbonaceous shale						

Table 9: Results of NAG test on F142446 core samples

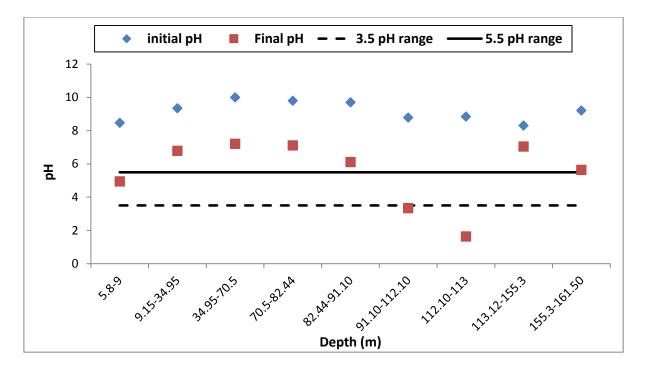


Figure 7: Initial and Final pH showing NAG pH test results of Borehole core F142446

# 5.2.1.1.3 F142471 Borehole Core

Ten samples from F142471 borehole core were taken from depths 4.68m to 109.90m (Table 10). Five samples (TC114, TC3, TC11, TC112 and TC105) have final pH's that are below 3.5 (Figure 8) and have a high risk of acid generation. Only TC107 sample is between pH of 3.5 and 5.5 and is regarded as having medium risk of acid generation, whereas the remaining samples are non-acid generating (>5.5). The NNP results (Appendix 9) indicates that 60% of samples need to be verified with other tests.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC65	4.68-12.9	SLT, SST, KV	8.3	6.69	Lower Acid Risk
TC55	12.9-18.82	SST	8.64	6.19	Lower Acid Risk
TC61	18.82-40.85	SST,SLT, KV	8.29	6.64	Lower Acid Risk
TC114 <sup>b</sup>	40.85-42.38	C5	8.45	1.65	Higher Risk Acid Generation
ТС3	42.38-75.68	SST, SLT, SH, KV	7.7	3.3	Higher Risk Acid Generation
TC107 <sup>b</sup>	76.09-80.80	C4H	7.67	4.22	Medium Risk Acid Generation
TC11	83.8-84.15	SLT	7.98	3.07	Higher Risk Acid Generation
TC112 <sup>b</sup>	84.15-84.80	C3	7.56	1.6	Higher Risk Acid Generation
TC84	84.8-109.4	SST, SLT, KV, GRT	8.08	6.38	Lower Acid Risk
TC105 <sup>b</sup>	109.4-109.90	C2	7.98	3.15	Higher Risk Acid Generation
		ne, SH-shale, GRT - coal seam & b = Coa		- Core loss, C	2 -No 2 coal seam, C3 -No 3

Table 10: Results of NAG pH test on F142471 core samples

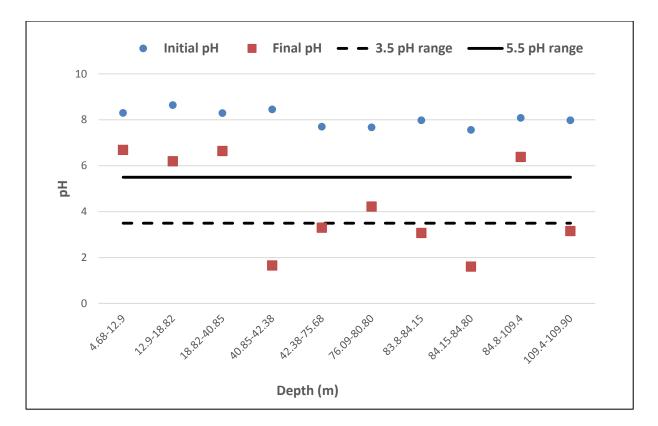


Figure 8: Initial and Final pH showing NAG pH test results of Borehole core F142471

# 5.2.1.1.4 O105016 Borehole Core

Four (TC113, TC48, TC24 & TC14) of eight samples collected from core O105016 indicates a low risk of acid generation (pH >5.5). Sample TC34 and TC79 indicate a high acid risk with a final pH less than 3.5 (Table 11 Figure 9). Samples TC43 and TC101 (coal sample) had a final pH between 3.5 and 5.5 (medium acid risk). Four samples indicated a low acid risk which means there is enough mineral to minimise the acid.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation		
TC113	20.26-33.5	SST, SLT	8.15	6.76	Lower Acid Risk		
TC48	33.5-41.65	SST, KV	8.98	7.14	Lower Acid Risk		
TC43	41.65-43.55	SST,SLT,SH	8.4	4.68	Medium Risk Acid Generation		
TC101 <sup>b</sup>	43.55-45.16	C5H & L	8.7	4.03	Medium Risk Acid Generation		
TC34	45.16-52.82	SST,SLT, CSH	7.57	2.92	Higher Risk Acid Generation		
TC24	52.82-58.99	DO, KV	9.64	7.89	Lower Acid Risk		
TC14	58.99-77.95	SST, SLT, KV, C	8.95	7.35	Lower Acid Risk		
тС79	80.3-85.56	SST, SLT, KV, GRT	7.32	2.31	Higher Risk Acid Generation		
	SST -Sandstone, SLT - Siltstone, DO - Dolerite, CSH -Carbonaceous shale, SH-Shale, GRT - Gritstone, KV - core loss, C - Coal, C5H - No 5 upper coal seam & lower coal seam, b = Coal sample						

Table 11: Results of NAG	pH test on O105016 core samples

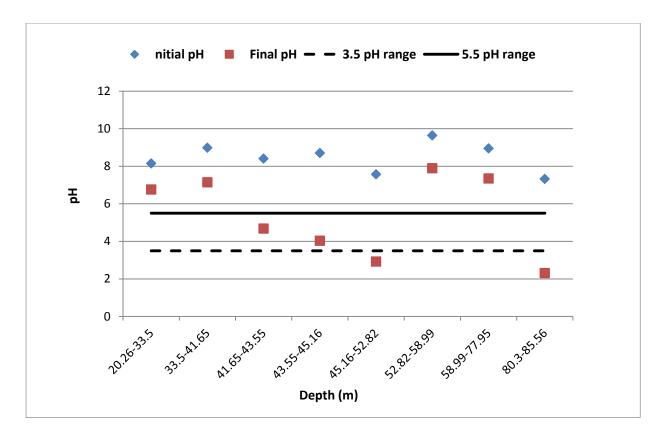


Figure 9: Initial and Final pH showing NAG pH test results of borehole core O105016

#### 5.2.1.1.5 P110030 Borehole Core

The results of the pH's for core P110030 are summarised in Table 12. Based on the interpretation of the pH's results for core P110030 (Table 12), It indicates that five samples have low risk, five samples have high risk and four have medium risk of acid generation (between 3.5 and 5.5). The medium risk samples fall within the grey NNP area (Appendix 9). For future work these samples (grey area) have to be submitted to kinetic testing to confirm the outcome of the samples under "field" weathering which was not part of this study.

Table 12: Results of NAG pH test on	P110030 core samples
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Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC49	3.8-16.23	MSB, KV, (OVB)	8.63	6.84	Lower Acid Risk
T52 <sup>ª</sup>	16.23-20.7	DO, KV	8.77	7.03	Lower Acid Risk
T99 <sup>a</sup>	16.23-20.7	DO, KV	8.97	8.12	Lower Acid Risk
TC1	20.7-30.23	SST, SSL	8.37	6.65	Lower Acid Risk
TC5	30.23-42.86	SST, KV	8.84	6.29	Lower Acid Risk
TC74	67.54-68.06	SH, C	8.76	4.30	Medium Risk Acid Generation
TC117 <sup>b</sup>	67.54-69.54	C5H	8.88	1.76	Higher Risk Acid Generation
TC104	69.54-106.77	SST, SLT, C	7.78	4.02	Medium Risk Acid Generation
TC109 <sup>b</sup>	106.38-110.22	C4	7.67	2.33	Higher Risk Acid Generation
TC82	106.77-109.99	C4L	7.69	3.65	Medium Risk Acid Generation
TC108 <sup>b</sup>	112.75-113.33	C3	7.67	1.47	Higher Risk Acid Generation

TC27	109.99-120.22	C, SST, SLT, CSH	7.64	4.03	Medium Risk Acid Generation	
TC98	120.22-141.42	SST, SLT, SH, GRT	8.26	3.12	Higher Risk Acid Generation	
TC106 <sup>b</sup>	141.42-141.55	C2	7.61	1.58	Higher Risk Acid Generation	
DO - Dolerite	, SST - Sandston	e, SLT - Siltstone, C	C - coal, CSH	H -Sarbonac	eous shale, MSB - Mudstone	
bands, KV -	bands, KV - Core loss, OVB - Overburden, SH - Shale, SSL - Sandstone lenses, C5H - No 5 Upper Coal					
Seam, C4 - N	Seam, C4 - No 4 coal seam, C4L - No 4 Lower Coal Seam, C3 - No 3 coal seam, C2 - No 2 coal seam, GRT -					
Gritstone, a = Same Sample , b = Coal Sample						

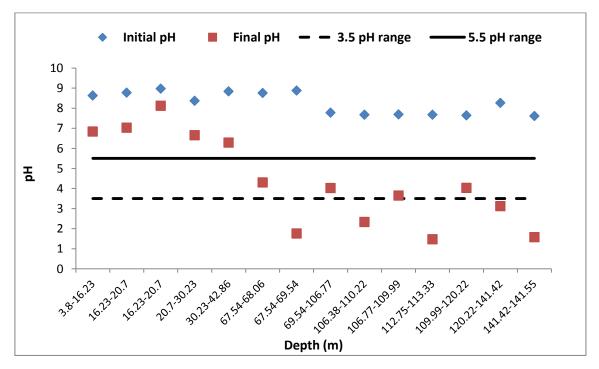
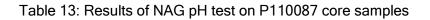


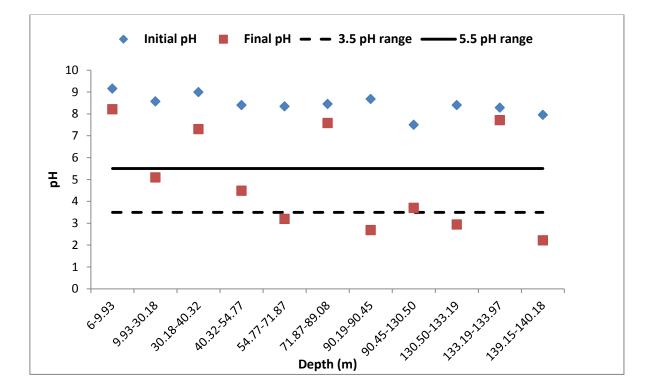
Figure 10: Initial and Final pH showing NAG pH test results of Borehole P110030

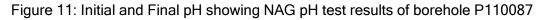
# 5.2.1.1.6 P110087 Borehole Core

The results of the pH's for core P110087 are summarised in Table 13. Based on the interpretation of the pH's results for core P110030 (Table 13), It indicates that four samples have low risk, four samples have high risk and three have medium risk of acid generation (between 3.5 and 5.5) (Figure 11). The medium risk samples fall within the grey NNP area (Appendix 9).

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC9	6-9.93	DO, KV	9.15	8.21	Lower Acid Risk
TC13	9.93-30.18	DO, SST, SLT, GRT	8.57	5.09	Medium Risk Acid Generation
TC97	30.18-40.32	SST, SLT	8.99	7.3	Lower Acid Risk
TC58	40.32-54.77	SST, SLT	8.4	4.48	Medium Risk Acid Generation
TC23	54.77-71.87	SST, GRT, KV	8.34	3.19	Higher Risk Acid Generation
TC94	71.87-89.08	SST, SLT	8.45	7.58	Lower Acid Risk
TC91 <sup>b</sup>	90.19-90.45	C5	8.68	2.68	Higher Risk Acid Generation
TC45	90.45-130.50	SST, SLT, GRT	7.5	3.7	Medium Risk Acid Generation
TC53	130.50-133.19	SST, SLT	8.4	2.94	Higher Risk Acid Generation
TC6	133.19-133.97	GRT, C, SST, SLT	8.28	7.71	Lower Acid Risk
TC73	139.15-140.18	SLT	7.95	2.21	Higher Risk Acid Generation







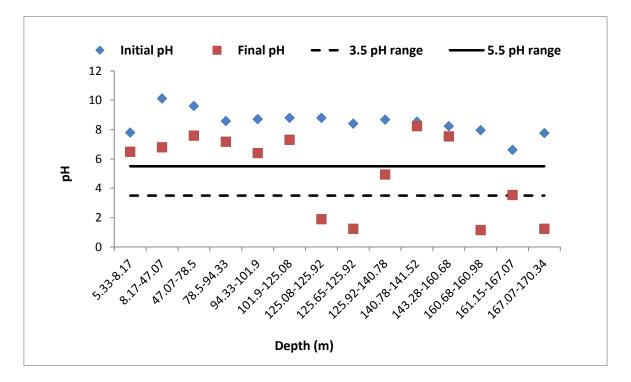
# 5.2.1.1.7 R146043 Borehole

Fourteen samples were collected from R146043 borehole core. The samples were taken from depths 5.33m to 170.34m. Four samples (TC28, TC110, 92 and TC118) have final pH's that are below 3.5 as indicated in Table 14 (high risk). TC40 and TC21 samples are between pH of 3.5 and 5.5 and are regarded as having medium risk of acid generation, whereas the remaining samples are non-acid generating (>5.5).

Most of the samples in core R146043 show low acid risk, indicating there are enough neutralising minerals to buffer the pH. Therefore, there will be a low risk of acid generation for this core sample

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC93	5.33-8.17	KLY, SST	7.80	6.49	Lower Acid Risk
TC116	8.17-47.07	DO	10.12	6.79	Lower Acid Risk
TC47	47.07-78.5	SST, SLT	9.61	7.59	Lower Acid Risk
TC59	78.5-94.33	SST, SLT	8.58	7.17	Lower Acid Risk
TC29	94.33-101.9	SST	8.71	6.40	Lower Acid Risk
TC16	101.9-125.08	SST, SLT	8.80	7.30	Lower Acid Risk
TC28	125.08-125.92	SLT, SH, C	8.80	1.89	Higher Risk Acid Generation
TC110 <sup>b</sup>	125.65-125.92	C5L	8.41	1.24	Higher Risk Acid Generation
TC40	125.92-140.78	SST, SLT, C, GRT	8.68	4.94	Medium Risk Acid Generation
TC15	140.78-141.52	SST	8.54	8.23	Lower Acid Risk
TC86	143.28-160.68	SST, SLT	8.23	7.53	Lower Acid Risk
TC92	160.68-160.98	CSH	7.96	1.15	Higher Risk Acid Generation
TC21	161.15-167.07	SST	6.62	3.54	Medium Risk Acid Generation
TC118 <sup>b</sup>	167.07-170.34	C4L	7.76	1.24	Higher Risk Acid Generation
KLY- Clay	ey, SST - Sandsto	one, DO - Dolerite, SH	I - Shale, CS	SH -carbonace	eous shale, SLT - siltstone, C5L -
No 5 Low	er coal seam ,C4L	- No 4 Lower coal se	eam, GRT-0	Gritstone, b =	Coal Sample

Table 14: Results of NAG pH test on R146043 core samples

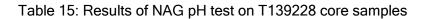




#### 5.2.1.1.8 T139228 Borehole Core

Eleven samples were collected from T139228 borehole core. Five samples (TC39, TC76, TC78, TC56 and TC60) have final pH's that are below 3.5 as indicated in Table 15 (high risk). TC70 and TC96 samples are between pH of 3.5 and 5.5 and are regarded as having medium risk of acid generation, whereas the remaining samples (four) are non-acid generating (>5.5).

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation		
TC70	2.2-7.35	SST, KV	8.24	5.4	Medium Risk Acid Generation		
TC44	7.35-39.45	DO	10.37	7.34	Lower Acid Risk		
TC36	39.45-42.87	SST, SLT	10.21	7.38	Lower Acid Risk		
TC63	42.87-43.5	DO	9.48	7.6	Lower Acid Risk		
TC30	43.5-82.2	SST,SLT, KV, GRT	9.08	7.06	Lower Acid Risk		
TC39	82.2-83.81	C, CSH, SH	9.43	1.75	Higher Risk Acid Generation		
TC76	83.81-115.08	SST,SLT,KV	7.96	3.36	Higher Risk Acid Generation		
TC96	115.08-117.67	C, SST, SLT, CSH	7.87	3.85	Medium Risk Acid Generation		
TC78	115.45-117.67	C, SST, SLT, CSH	6.12	2.24	Higher Risk Acid Generation		
TC56	121.35-141.39	SST,SLT,KV	7.95	3.23	Higher Risk Acid Generation		
TC60	141.39-145.24	SST, SLT, SH,C	8.52	2.89	Higher Risk Acid Generation		
	SST - Sandstone, SLT - Siltstone, DO - Dolerite, CSH -Carbonaceous shale, GRT - Gritstone, KV - Core loss, C - Coal, SH - Shale						



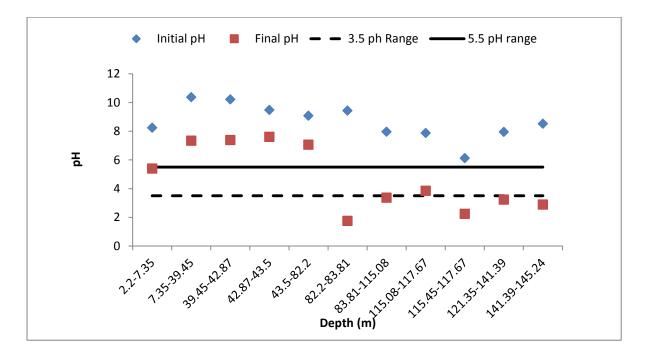


Figure 13: Initial and Final pH showing NAG pH test results of T139228

#### 5.2.1.1.9 W569001 Borehole

Three (TC111, TC102 & TC103) of six samples collected from core W569001 were coal samples. Two samples (T102 & T103) indicate a high risk of acid generation with final pH's less than 3.5. According to the XRD results the samples (T102 & T103) also contain pyrite mineral (Appendix 3). Sample T111 indicates a lower acid risk with a final pH above 5.5 (Table 16 and Figure 14) which means the sample has excess neutralising minerals (Appendix 9) and the calculated NNP for this sample was 47.31 kgCaCO3/tonne. Samples TC80 and TC89 had a final pH between 3.5 and 5.5 (medium acid risk). Four samples indicated a low acid risk which means there is enough mineral to minimise the acid.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation	
TC69	2.4-7.4	DO, KV	9.55	6.56	Lower Acid Risk	
TC80	18.7-19.5	SH, CSH	7.44	3.51	Medium Risk Acid Generation	
TC54	19.5-55.81	SST, SLT, KV	8.77	6.74	Lower Acid Risk	
TC111 <sup>b</sup>	55.81-59.57	C4L	7.64	6.01	Lower Acid Risk	
TC89	59.57-62.8	SST, SLT, KV	8	3.8	Medium Risk Acid Generation	
TC102 <sup>b</sup>	62.8-63.03	C3	7.97	1.78	Higher Risk Acid Generation	
TC22	63.03-90.4	SST, SLT, KV, SH, GRT, CSH	9.09	6.73	Lower Acid Risk	
TC103 <sup>b</sup>	90.4-90.8	C2	7.13	1.69	Higher Risk Acid Generation	
DO - Dolerite, KV - Core loss, SH - Shale, CSH -carbonaceous shale, SST - Sandstone, SLT - siltstone, C4L - No 4 Lower Coal Seam, C3 - No 3 coal seam, C2 - No 2 coal seam, GRT - Gritstone, b - Coal Sample						

Table 16: Results of NAG pH test on W569001 core samples

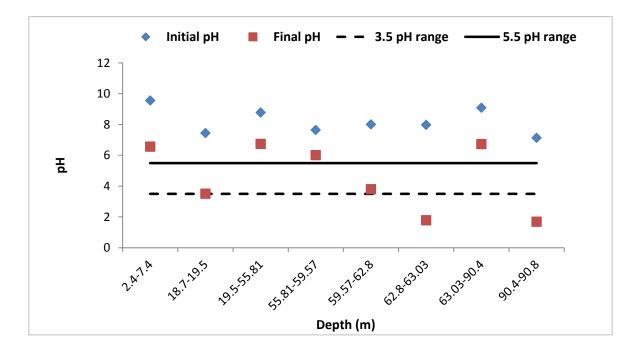


Figure 14: Initial and Final pH showing NAG pH test results of Borehole W569001

#### 5.2.1.1.10 Y106048 Borehole Core

The results of the pH's for core Y106048 are summarised in Table 17. Based on the interpretation of the pH's results for core Y106048 (Table 17), It indicates that two samples have high risk (<3.5) and three have medium risk of acid generation (between 3.5 and 5.5) (Figure 15). The medium risk samples fall within the grey NNP area (Appendix 9).

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation	
TC32	5.95-6.66	SST, KV	7.5	3.21	Higher Risk Acid Generation	
TC12	6.66-22.61	SST,SLT, GRT	8.18	5.1	Medium Risk Acid Generation	
TC8	22.61-45.06	SST, SLT, GRT	8.28	5.46	Medium Risk Acid Generation	
TC95	50.59-70.93	SST, SLT	7.97	4.53	Medium Risk Acid Generation	
TC19	70.93-75.27	SST, SLT,GRT	8.08	3	Higher Risk Acid Generation	
KV - Core l	KV - Core loss, SST - Sandstone, SLT - siltstone, GRT - Gritstone.					

Table 17: Results of NAG pH test on Y106048 core samples

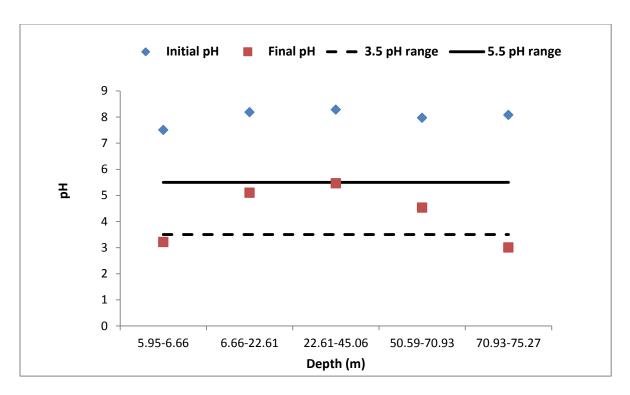


Figure 15: Initial and Final pH showing NAG pH test results of borehole core Y106048

#### 5.2.1.1.11 Z145168 Borehole Core

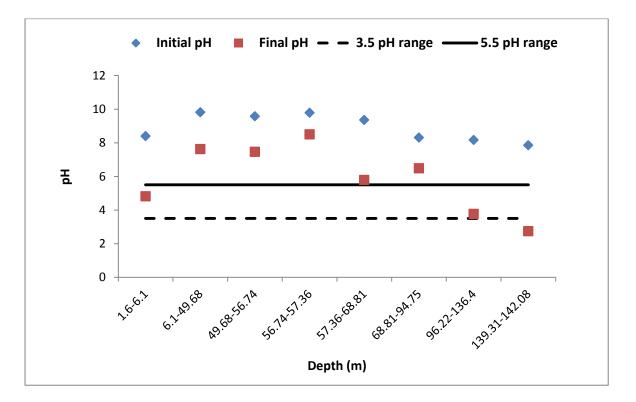
Eight samples were collected from Z145168 borehole core. Five samples indicate low acid risk (Table 18) Only one sample has a final pH that is below 3.5 as shown in Figure 16 (high risk). TC31 and TC67 samples are between pH of 3.5 and 5.5 and are regarded as having medium risk of acid generation.

Most of the samples in core Z145168 show low acid risk, indicating there are enough neutralising minerals to buffer the pH.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC31	1.6-6.1	SST, GKS, KV	8.39	4.81	Medium Risk Acid Generation
TC115	6.1-49.68	DO	9.81	7.62	Lower Acid Risk
TC100	49.68-56.74	SST, SLT, KV	9.57	7.46	Lower Acid Risk
TC37	56.74-57.36	DO	9.78	8.5	Lower Acid Risk
TC68	57.36-68.81	SST, SLT, MST	9.35	5.78	Lower Acid Risk
TC85	68.81-94.75	SST, SLT, GRT	8.31	6.48	Lower Acid Risk
TC67	96.22-136.4	SST, SLT, C, CSH, DO	8.16	3.77	Medium Risk Acid Generation
TC42	139.31-142.08	SST, SLT	7.85	2.74	Higher Risk Acid Generation

Table 18: Results of NAG pH test on Z145168 core samples

DO - Dolerite, KV - Core loss, , CSH -Carbonaceous shale, SST - Sandstone, SLT - Siltstone, GRT -Gritstone, C - Coal, GKS - Soil clayey mostly sandy, MST -Mudstone





# 5.2.1.1.12 Z145198 Borehole Core

Figure 17 show three samples (TC2, TC72 and TC88) with a final pH's below 3.5 (high risk) and Three samples (TC71, TC90 and TC26) with a final pH's above 5.5 (low acid risk). The XRD results (Appendix 3) shows that sample all samples with low acid risk contain calcite.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation		
TC71	3.33-45.04	DO	10.12	7.75	Lower Acid Risk		
TC90	45.04-90.02	SST,SLT,SH	8.78	6.83	Lower Acid Risk		
TC26	91.03-127.78	SST,SLT,SH	8.63	7.7	Lower Acid Risk		
TC2	132.14-134.94	SST, SLT, C	7.7	2.85	Higher Risk Acid Generation		
TC72	134.94-151.78	SST, SLT, SH	8.03	3.07	Higher Risk Acid Generation		
TC88	151.78-184	SST, SST,C	7.88	3.02	Higher Risk Acid Generation		
DO - Doler	DO - Dolerite, SH - Shale, SST - Sandstone, SLT - Siltstone, C - Coal						

Table 19: Results of NAG pH test on Z145198 core samples

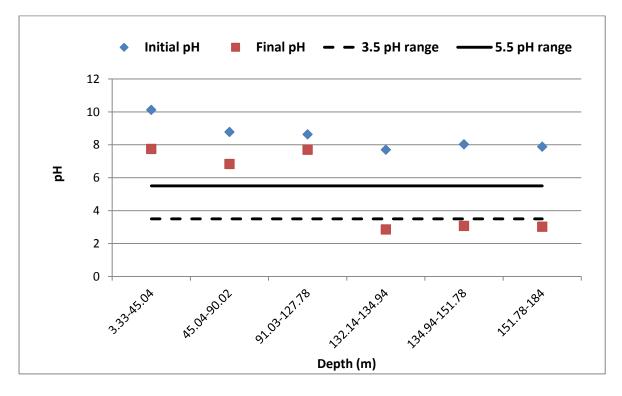


Figure 17: Initial and Final pH showing NAG pH test results of borehole core Z145198

#### 5.2.1.1.13 Z145199 Borehole Core

Figure 18 show two samples (TC81 & TC64) with a final pH below 3.5 (high risk of acid generation) (Table 20).The XRD results (Appendix 3) shows that sample TC64 contains pyrite which cause a high risk of acid generation. Sample TC46, TC17, TC50, TC77 and TC20 have a low acid risk with a final pH greater than 5.5. The XRF results for samples TC46 and TC50 (Appendix 5) indicates that it contains a high percentage of CaO. Samples TC77 and TC20 consist of calcite mineral which have the potential to neutralising the acid.

More samples have low risk of generating an acid therefore core samples from Borehole Z145199 will not be expected to generate acid as it has enough minerals to neutralise an acid.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
TC46	0-22.52	KLY,KV,DO	9.59	6.98	Lower Acid Risk
TC17	22.52-28.75	SST, SLT	8.79	6.82	Lower Acid Risk
TC50	28.75-52.14	DO,KV	10.18	7.64	Lower Acid Risk
TC77	52.14- 123.92	SST,SLT,KV,DO,GRT	8.53	7.76	Lower Acid Risk
TC81	123.92- 124.6	SH	8.77	1.79	Higher Risk Acid Generation
TC20	125.80- 159.6	SST, SLT, C	8.36	7.53	Lower Acid Risk
TC64	160.3-162.3	C, SST	8.15	3.03	Higher Risk Acid Generation
KLY- Clayey C - Coal	, KV - Core loss	, DO - Dolerite, SST - Sa	ndstone, S	SLT - siltston	e, SH - Shale, GRT - Gritstone,

Table 20: Results of NAG pH test on Z145199 core samples

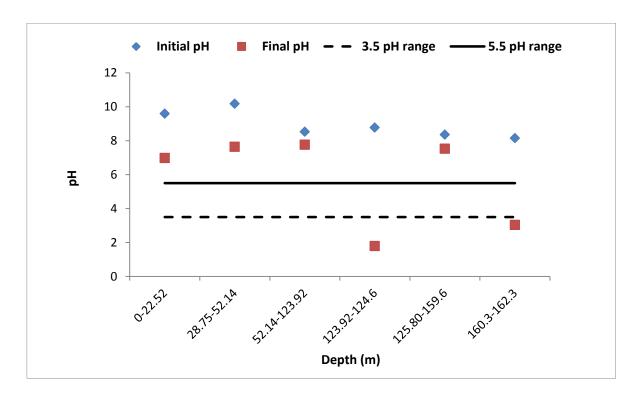


Figure 18: Initial and Final pH showing NAG pH test results of Borehole Z145199

#### 5.2.1.2 Net Neutralising Potential

Figure 19 shows the initial and final pH versus closed NNP results. All samples have initial pH that is greater than 6 which mean samples have not been oxidised in the field. NNP results (Appendix 9) indicates that 62 samples (52.5 %) are inconclusive as they fall within the area of uncertainty -20 to 20kg/t CaCO<sub>3</sub> (Figure 19). Samples in this area will need

further testing for further confirmation of the oxidisation outcome e.g. Kinetic/Humidity Tests. The NNP results also indicate that 36 samples (30.5 %) are classified as having a neutralising potential. Only 20 samples (17 %) have the potential to generate a net acid with a closed NNP of less than 20kg/t CaCO<sub>3</sub>.

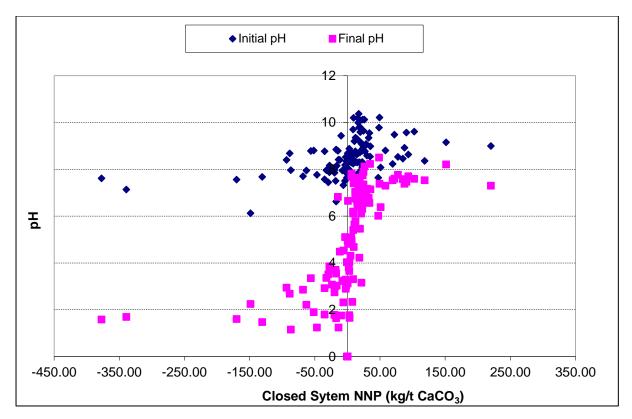


Figure 19: Initial and Final pH vs close NNP

# 5.2.1.3 Neutralising Potential Ratio (NPR)

The AP and NP data (Appendix 11) obtained for the mines is used to plot the NPR graph (Figure 20). Information from Table 6 was used as a guideline for the interpretation of Figure 20. From this NPR graph, the samples that plot below the green line of 4:1, acidification is unlikely. In samples that plot above the ratio of 1:1, acidification is likely to happen. For this specific mine, more (44.1%) samples plotted below the 4:1 ratio which are unlikely to generate an acid. Only a few (35.6%) samples plotted above 1:1 which are likely to acidify. The samples plotting between 1:1 and 4:1 ratio are inconclusive (20.3%).

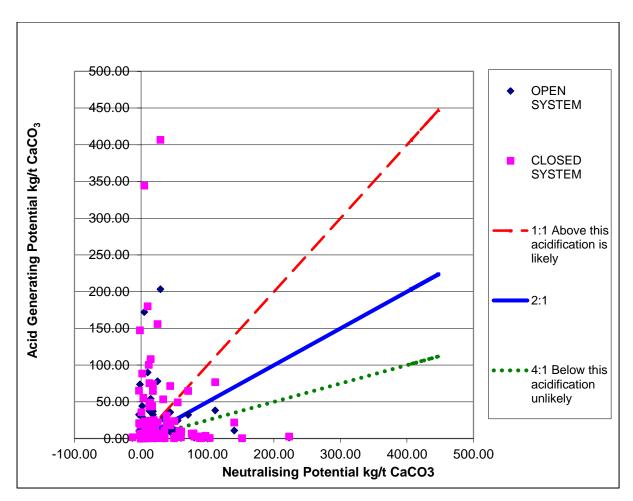


Figure 20: AP vs NP (NPR)

#### 5.2.1.4 Neutralising Potential Ratio (NPR) versus sulphide S

The data from Appendix 11 are used to create the NPR versus percentage S plot. The plots from this graph confirm the results as discussed in the previous headers. Overall there are more (44.1 %) samples plotting in the green area indicating that these samples contain a base potential and no potential to produce acid. The samples plotting in the red area (to the right of 0.3 %S) and below the NPR of 1 (red line) are the samples with the potential to generate acid. The rest of the samples are those plotting in the grey area.

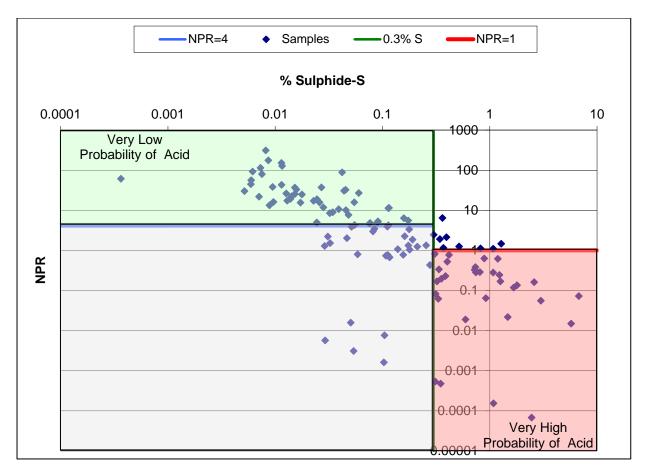


Figure 21: NPR vs Sulphide-S

# 5.2.1.5 Conclusion

Out of 118 samples analysed from Mine 1, 58 samples (49%) were non-acid generating with a pH's above 5.5, while 22 samples (19%) have a medium risk of generating an acid as their pH values were between 3.5 and 5.5 and the rest (32%) of the samples (38) have the pH's of less than 3.5 which have a high risk of generating an acid (Appendix 7). Most of the samples that have a high risk of generating an acid were coal samples and samples that have a mixture of coal combined with different rock formation. Samples that have a low risk of acid generation they contain calcite and dolomite which are the most common neutralising minerals. Therefore, the risk of acid generation will be on coal seam than on other rock formation.

# 5.2.2 Mine 2 analysis

5 borehole core samples were collected from for Mine 2. 71 samples were selected and analysed. The samples collected consist of sandstone, siltstone, coal and other geological formations shown in Appendix 8.

#### 5.2.2.1 Net Acid Generating (NAG) Test pH

Results for the NAG test are presented in Tables 21 to Table 25. Interpretation of the results helped to assess whether the samples are acid or non-acid generating. The graphs below each table of results are plotted to show the number of samples with a pH above 5.5, between 5.5 and 3.5 and below 3.5. These graphs are used for further interpretation. The information from Table 5 above was used to indicate the likelihood of net acid generation of the sample upon oxidation.

#### 5.2.2.1.1 R100001 Borehole Core

Thirteen samples were collected from R100001 borehole core. The samples were taken from depths 1.62m to 143.10m. Seven samples indicate a lower acid risk with a final pH above 5.5 (Table 21) which means the samples have excess neutralising minerals. Two samples (SF13 and SF2) have final pH's that are below 3.5 as indicated in Figure 22 (high risk). Samples SF1, SF15, SF16 and SF17 are between pH of 3.5 and 5.5 and are regarded as having medium risk of acid generation

Most of the samples in core R100001 show low acid risk, indicating there are sufficient neutralising minerals to buffer the pH.

Lab Number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
SF 11	1.62-12.46	DO	9.8	7.75	Lower Acid Risk
SF 12	12.46-24.48	SST, SLT, GRT	9.81	7.56	Lower Acid Risk
SF 13	24.48-25.26	C, CSH	9.85	2.37	Higher Risk Acid Generation
SF 61 <sup>b</sup>	24.48-25.80	C5H, C5M, C5L	8.89	5.61	Lower Acid Risk
SF 14	25.8-50.66	SLT, SST	8.93	7.3	Lower Acid Risk
SF 1	50.66-51.6	SLT, GRT	8.04	4.27	Medium Risk Acid Generation
SF 62 <sup>b</sup>	51.6-56.92	C4L	8.22	6.21	Lower Acid Risk
SF 2	56.92-57.14	SLT	7.73	3.44	Higher Risk Acid Generation
SF 15	57.14-85.28	SST, SLT	8.6	4.51	Medium Risk Acid Generation
SF 16	86.82-112.94	SST	7.63	3.64	Medium Risk Acid Generation
SF 17	112.94-116.23	GRT	7.55	3.51	Medium Risk Acid Generation
SF 18	116.23-130.1	SST, GRT, SLT, WSH	8.14	6.85	Lower Acid Risk
SF 19	130.1-143.1	TIL	9.24	7.83	Lower Acid Risk
					SH -Carbonaceous shale, , C5H - r Coal Seam, C4L - No 4 Lower

#### Table 21: Results of NAG pH test on R100001 core samples

Coal Seam, WSH - Varved Shale, TIL - Tillite, b = Coal Sample

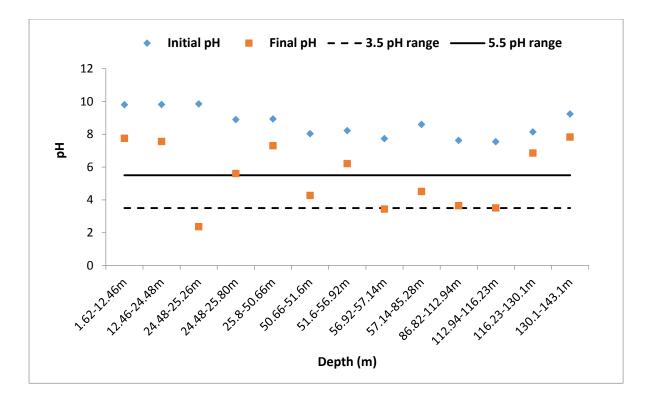


Figure 22: Initial and Final pH showing NAG pH test results of Borehole R100001

# 5.2.2.1.2 R100002 Borehole Core

Figure 23 shows one sample (SF4) with a final pH below 3.5 (high risk). The XRD results (Appendix 3) show that sample SF4 contains pyrite which causes a high risk of acid generation. Samples SF20, SF21, SF3 and SF24 have a low acid risk with a final pH greater than 5.5. The XRF results for samples SF20 and SF21 (Appendix 5) indicates that they contains a high percentage of CaO and they contain calcite which have the potential to neutralising the acid. The medium risk samples fall within the grey NNP area (Appendix 10).

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation		
SF 20	14.1-18	GRD, SST, DO	9.16	7.92	Lower Acid Risk		
SF 21	18.3-37.5	SST, SLT	8.7	7.62	Lower Acid Risk		
SF 3	37.5-37.82	GRT	7.86	5.91	Lower Acid Risk		
SF 63 <sup>b</sup>	37.82-43.08	C4L	7.95	5.49	Medium Risk Acid Generation		
SF 4	43.08-45.72	SST, SLT	7.92	1.84	Higher Risk Acid Generation		
SF 22	45.72-70.1	SST, SLT	8.4	4.25	Medium Risk Acid Generation		
SF 23	71.4-103.8	SST, GRT	7.77	3.72	Medium Risk Acid Generation		
SF 24	103.8-124.4	GRT, TIL, KV	8.25	6.57	Lower Acid Risk		
	DO - Dolerite, KV - Core loss, SH - Shale, CSH -Carbonaceous shale, SST - Sandstone, SLT - Siltstone, C4L - No 4 Lower Coal Seam, GRT - Gritstone, TIL - Tillite, KV - Core loss b = Coal Sample						

Table 22: Deculte of NAC	nH tast an aara samplas	from horoholo D100002
Table 22: Results of NAG	ph lest on core samples	

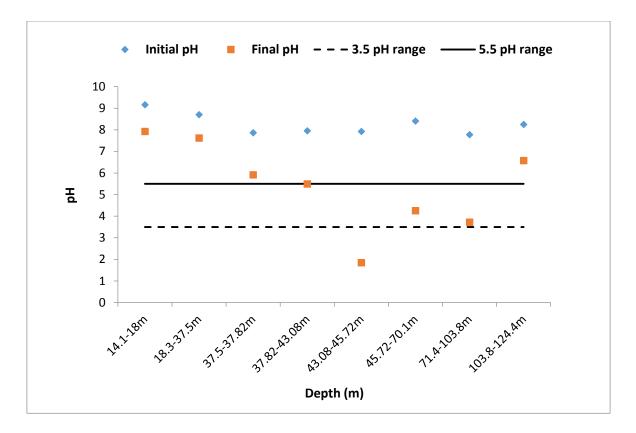


Figure 23: Initial and Final pH showing NAG pH test results of Borehole R100002

#### 5.2.2.1.3 Z124027 Borehole Core

The results of the pH's for core Z124027 are summarised in Table 23. Based on the interpretation of the pH's results for core Z124027 (Table 23), It indicates that six samples have high risk (<3.5). Two samples (SF59 and SF60) have medium risk of acid generation (between 3.5 and 5.5) (Figure 24). The medium risk samples fall within the grey NNP area (Appendix 10). Most (Thirteen) of the samples in core Z124027 show low acid risk, indicating there are enough neutralising minerals to buffer the pH.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
SF 45	4.13-14.04	DO, KV,	9.27	8.43	Lower Acid Risk
SF 46	14.04-48.99	SST, SLT	9.54	8.1	Lower Acid Risk
SF 47	48.99-69.26	SST, DO	8.9	5.86	Lower Acid Risk
SF 48	69.26-90.34	SST, SLT	9.05	7.22	Lower Acid Risk
SF 49	90.44-91.84	C, CSH	9.34	2.41	Higher Risk Acid Generation
SF 64 <sup>b</sup>	91.84-92.34	C5L	8.94	7.74	Lower Acid Risk
SF 50	92.34-93.61	DO	9.89	8.22	Lower Acid Risk
SF 51	93.61-123.04	SST, SLT	8.89	7.81	Lower Acid Risk
SF 65 <sup>b</sup>	123.08-123.68	C4H	8.31	2.29	Higher Risk Acid Generation

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
SF 52	123.68-125.28	SH, SST, GRT, SLT	8.07	3.21	Higher Risk Acid Generation
SF 7	125.28-125.88	GRT	5.57	2.33	Higher Risk Acid Generation
SF 66 <sup>b</sup>	125.88-129.68	C4L	8.18	5.51	Lower Acid Risk
SF 8	129.78-134.51	SLT, SST, C	7.37	2.87	Higher Risk Acid Generation
SF 53	134.48-162.76	SLT, SST, C, DO	9.21	6.67	Lower Acid Risk
SF 54	162.54-164.8	DO, SST, C	10.33	8.5	Lower Acid Risk
SF 55	167.2-172.87	GRT, SLT	8.72	2.52	Higher Risk Acid Generation
SF 56	172.87-180.67	SST, DO	8.92	6.83	Lower Acid Risk
SF 57	180.67-183.7	DO	9.91	8.25	Lower Acid Risk
SF 58	183.7-186.86	SST, SLT	8.88	5.88	Lower Acid Risk
SF 59	186.86-189.02	TIO, SST	8.32	4.37	Medium Risk Acid Generation
SF 60	189.02-200.16	QZT, EOH	8.65	4.32	Medium Risk Acid Generation

DO - Dolerite, SST - Sandstone, SLT - Siltstone, C - Soal, CSH - Carbonaceous shale, MSB - Mudstone bands, KV - Core loss, SH - Shale, C5L - No 5 Lower Coal Seam, C4L - No 4 Lower Coal Seam, C4H - No 4 Upper Coal Seam, GRT - Gritstone, TIO - Tilloid, QZT - Quartzite, EOH - End of hole, b = Coal Sample

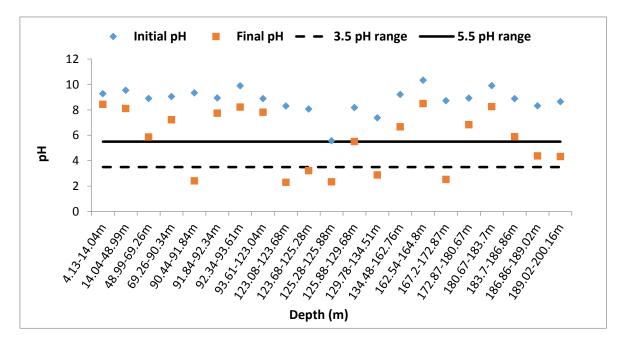


Figure 24: Initial and Final pH showing NAG pH test results of Borehole Z124027

#### 5.2.2.1.4 Z124029 Borehole Core

Figure 25 shows four samples (SF5, SF30, SF31 and SF32) with a final pH below 3.5 (high risk). The XRD results (Appendix 3), indicates that sample SF5 and SF32 contains pyrite which cause a high risk of acid generation. Eight samples have a low acid risk with a final pH greater than 5.5. Only one sample has medium risk. Borehole core Z124029 has more samples with potential of neutralising an acid than potential of generating an acid.

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
SF 25	6.9-8.1	KV, SST	8.73	5.65	Lower Acid Risk
SF 26	8.1-61.46	DO	10.41	7.05	Lower Acid Risk
SF 27	61.46-73.68	SST, SLT, GRT	10.38	6.46	Lower Acid Risk
SF 28	73.68-74.72	C, CSH	10.09	5.91	Lower Acid Risk
SF 29	74.72-98.23	SST, SLT	9.9	6.89	Lower Acid Risk
SF 5	98.23-98.88	SST, SLT	8.2	3.36	Higher Risk Acid Generation
SF 67 <sup>b</sup>	98.88-102.83	C4L	8.19	5.41	Medium Risk Acid Generation
SF 6	102.83-104.6	GRT	8.43	6.75	Lower Acid Risk
SF 30	104.6-132.74	SST, SLT, GRT	8.54	3.45	Higher Risk Acid Generation
SF 31	133.63-136.28	SST, SLT, GRT	8.57	2.59	Higher Risk Acid Generation
SF 32	138.64-158.45	SST, SLT, GRT	8.58	2.97	Higher Risk Acid Generation
SF 33	158.45-165.43	SST, SLT, TIL	8.55	5.65	Lower Acid Risk
SF 34	165.43-174.53	GR, EOH	10.26	7.9	Lower Acid Risk

Table 24: Results of NAG pH test on core samples from borehole Z124029

KV - Core loss DO - Dolerite, SST - Sandstone, SLT - Siltstone, GRT - Gritstone , C - Coal, CSH - carbonaceous shale, C4L - No 4 Lower Coal Seam, GR - Granite , EOH - End of hole, b = Coal Sample

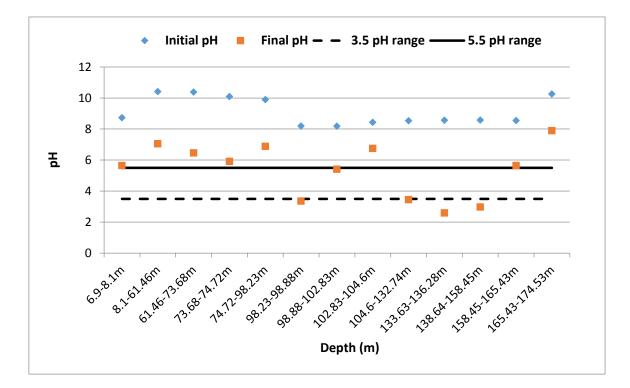


Figure 25: Initial and Final pH showing NAG pH test results of Borehole Z124029

#### 5.2.2.1.5 Z124030 Borehole Core

Four (SF68, SF69, SF70 and SF71) of sixteen samples collected from core Z124030 were coal samples. Three samples (SF68, SF69 and SF70) indicate a high risk of acid generation with final pH's less than 3.5. According to the XRD results the samples (SF68, SF40 and 70)

also contain pyrite mineral (Appendix 4). Eight samples indicate a lower acid risk with a final pH's above 5.5 (Table 25 and Figure 26) which means the sample has excess neutralising minerals (Appendix 10). Samples SF35 and SF41 had a final pH between 3.5 and 5.5 (medium acid risk).

Lab number	Depth (m)	Geology	Initial pH	Final pH	Interpretation
SF 35	11.6-12.52	SST	9.15	5.29	Medium Risk Acid Generation
SF 36	12.52-20.8	DO, KV	9.11	8.84	Lower Acid Risk
SF 37	20.8-41.35	SST, SLT	8.62	6.43	Lower Acid Risk
SF 38	41.35-49	DO	10.21	6.42	Lower Acid Risk
SF 39	49-98.3	SST, SLT	9.31	8.3	Lower Acid Risk
SF 68 <sup>b</sup>	99.2-99.42	C5H	8.1	2	Higher Risk Acid Generation
SF 40	98.3-100.45	C, SLT	9.63	2.34	Higher Risk Acid Generation
SF 69 <sup>b</sup>	100.45-100.8	C5L	9.25	2.51	Higher Risk Acid Generation
SF 41	100.8-133.37	SST, SLT	8.91	4.4	Medium Risk Acid Generation
SF 70 <sup>b</sup>	133.37-133.92	C4H	8.73	1.97	Higher Risk Acid Generation
SF 9	133.92-135.56	GRT	6.6	2.81	Higher Risk Acid Generation
SF 71 <sup>b</sup>	135.56-139.83	C4L	8.17	6.82	Lower Acid Risk
SF 10	139.83-140.6	SST	6.96	3.34	Higher Risk Acid Generation
SF 42	140.6-164.48	C, SST, SLT, KV	8.91	5.84	Lower Acid Risk
SF 43	167.1-168.2	SLT	10.11	6.66	Lower Acid Risk
SF 44	168.2-185.66	SST, TIO, LAV	10.25	7.09	Lower Acid Risk

Table 25: Results of NAG pH test on core samples from borehole Z124030

KV - Core loss DO - Dolerite, SST - Sandstone, SLT - Siltstone, GRT - Gritstone, C - Coal, C5H - No 5 Upper Coal Seam, C5L - No 5 Lower Coal Seam, C4H - No 4 Upper Coal Seam, TIO - Tilloid , LAV- Lava, b = Coal Sample

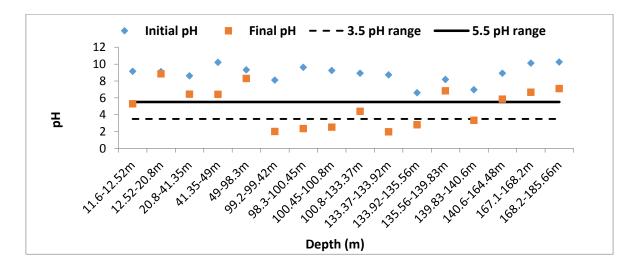


Figure 26: Initial and Final pH showing NAG pH test results of Borehole Z124030

#### 5.2.2.2 Net Neutralising Potential (NNP)

Figure 27 shows the initial and final pH versus closed NNP results. All samples have initial pH that is greater than 6 which mean samples have not been oxidised in the field. NNP results (Appendix 10) indicates that 30 samples (42.3 %) are inconclusive as they fall within the area of uncertainty -20 to 20kg/t CaCO<sub>3</sub> (Figure 27). Samples in this area will need further testing for further confirmation of the oxidisation outcome e.g. Kinetic/Humidity Tests. The NNP results also indicate that 31 samples (43.7 %) are classified as having a neutralising potential. Only 10 samples (14 %) have the potential to generate a net acid with a closed NNP of less than 20kg/t CaCO<sub>3</sub>.

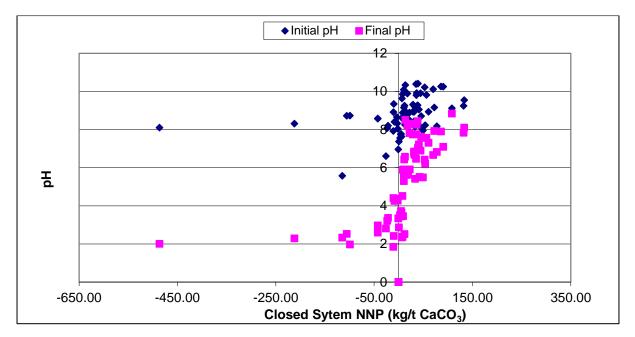


Figure 27 : Initial and Final vs Closed NNP

#### 5.2.2.3 Neutralising Potential Ratio (NPR)

The AP and NP data (Appendix 12) obtained for the mines is used to plot the NPR graph (Figure 28). Information from Table 6 was used as a guideline for the interpretation of Figure 20. From this NPR graph, the samples that plot below the green line of 4:1, acidification is unlikely. Samples that plot above the ratio of 1:1, acidification is likely to occur. For this specific mine, more (69%) samples plotted below the 4:1 ratio which are unlikely to generate acid. Only a few (12.7%) samples plotted above 1:1 which are likely to acidify. The samples plotting between 1:1 and 4:1 ratio are inconclusive (18.3%).

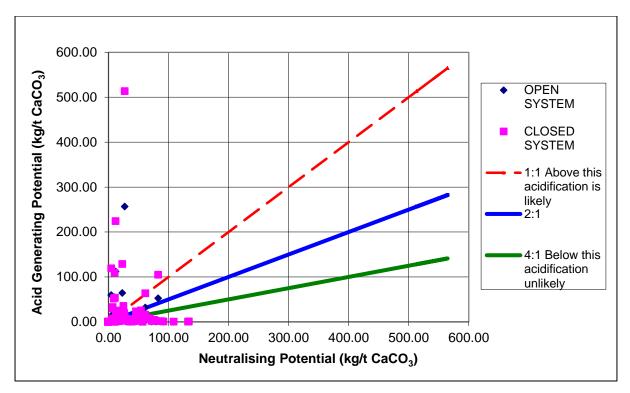


Figure 28: AP vs NP (NPR)

# 5.2.2.4 Neutralising Potential Ratio (NPR) versus sulphide S

The NPR and percentage S data (Appendix 12) are plotted in the following graph. The plots from this graph confirm the results as discussed in the previous headers. Most (69%) samples plot in the green area indicating that these samples contain a base potential and no potential to produce acid. The samples plotting in the red area (to the right of 0.3 %S) and below the NPR of 1 (red line) are the samples with the potential to generate acid. The rest of the samples are those plotting in the grey area.

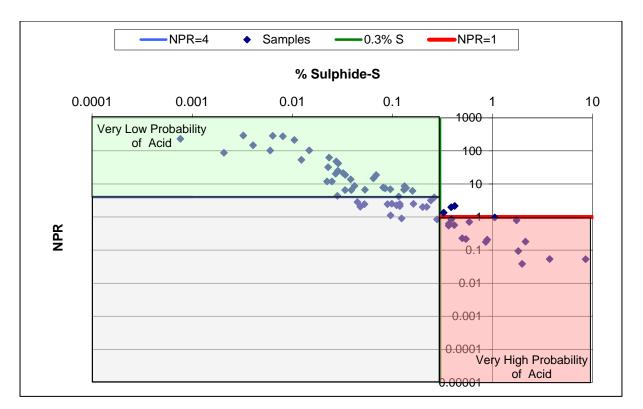


Figure 29: NPR vs Sulphide-S

# 5.2.2.5 Conclusion

71 Samples from Mine 2 were analysed for NAG pH test and out of 71 samples analysed, 40 samples (56%) were non-acid generating with pH values above 5.5, while 12 samples (17%) have a medium risk of generating an acid as their pH was between 3.5 and 5.5 and the rest (27%) of the samples have the pH of less than 3.5 which have a high risk of generating an acid. The initial and final pH results are shown in Appendix 8. The samples that showed high risk of acid generation contained pyrite and most of them were coal samples. Therefore coal samples have the potential to generate an acid as compared to other geological formations.

# 5.3 Leach Test

Groundwater near mines is often heavily polluted due to oxidation of sulphide minerals like pyrite (FeS<sub>2</sub>), galena (PbS), sphalerite (ZnS) or arsenopyrite (FeAsS). The pH has a major influence on the mobility of metals and it differs for various metals/elements (Appelo and Postma 2005). In conditions where AMD occurs, metals are leached out of the surrounding lithology and these can contaminate groundwater resources (Blowes and Ptacek, 1994).

Metal and trace element content of the samples tested were analysed using Inductively Coupled Plasma Optical Emission Spectrometer (ICP-OES). The following elements were determined using ICP-OES: Ag, Al, As, Ba, Be, Ca, Cd, Cr, Co, Cu, Fe, K, Li, Mg, Mn, Mo, Na, Ni, Sb, Se, Sn, Si, Pb, V, Zn, and SO<sub>4</sub>. The leach results are summarised in the following part.

#### 5.3.1 Mine 1 leach test results

The solubility of some of the elements leached in different media (water, hydrogen peroxide and acid leach) for Mine 1 are plotted against the pH for overburden (Figure 30), interburden (Figure 31) and the coal formations (Figure 32).

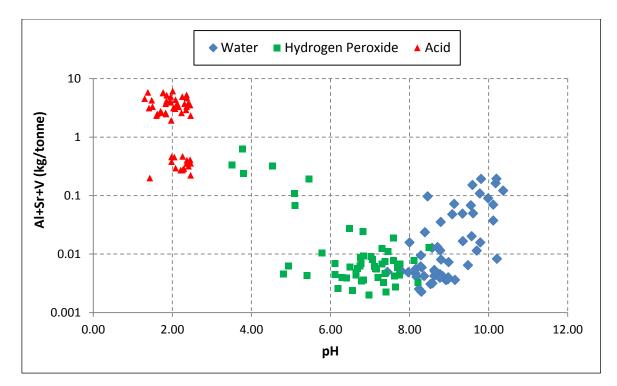


Figure 30: Plot of metal solubility (AI+Sr+V) vs pH (overburden)

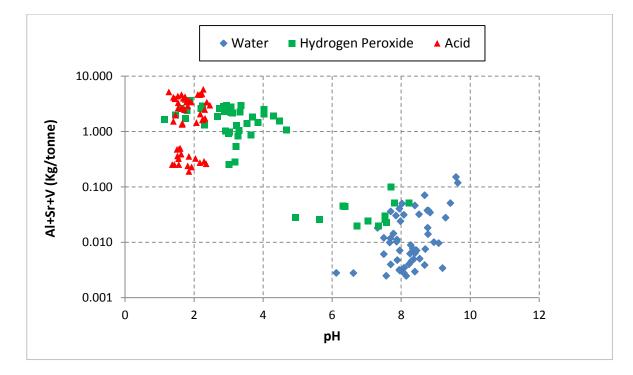


Figure 31: Plot of metal solubility (AI+Sr+V) vs pH (interburden)

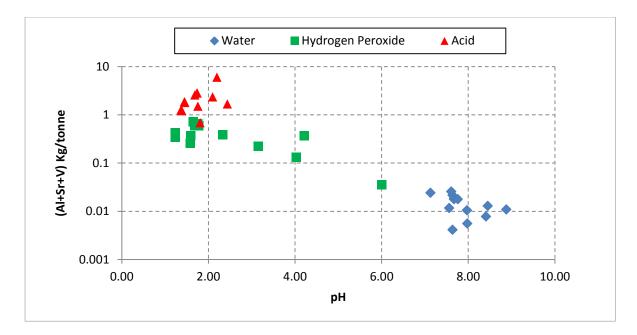


Figure 32: Plot of metal solubility (AI+Sr+V) vs pH (coal formations)

The solubility of the metals varies with the pH.

From these graphs (Figure 30 to Figure 32) it can be seen that the solubility of the metals varies with the pH. The presence of more sulphide minerals causing acidification upon oxidation can be seen in the interburden and coal formations. The peroxide pH results for the overburden demonstrates a minimum pH of 3.51 while for the interburden and coal formations the minimum pH values were 1.15 and 1.24.

Plots of other metals indicate the same trend (Figure 33-36)

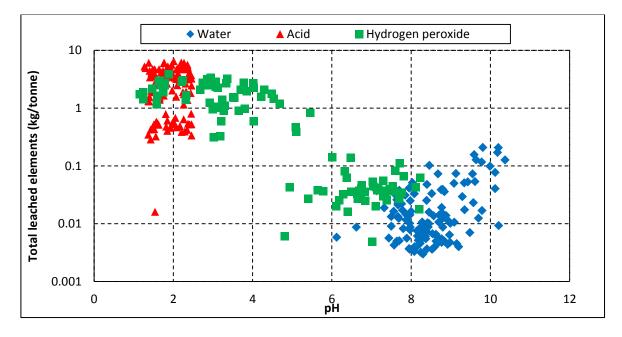


Figure 33: Plot of metal solubility (Total leached elements) vs pH

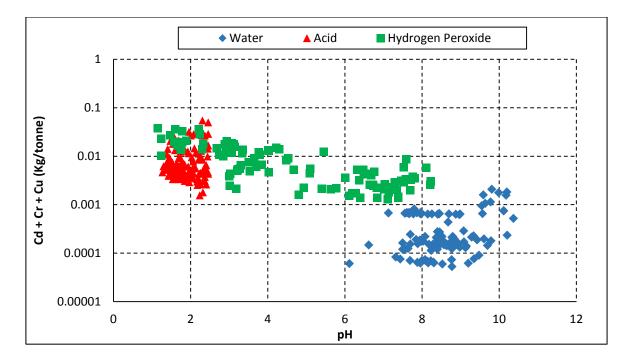


Figure 34: Plot of metal solubility (Cd + Cr + Cu) vs pH

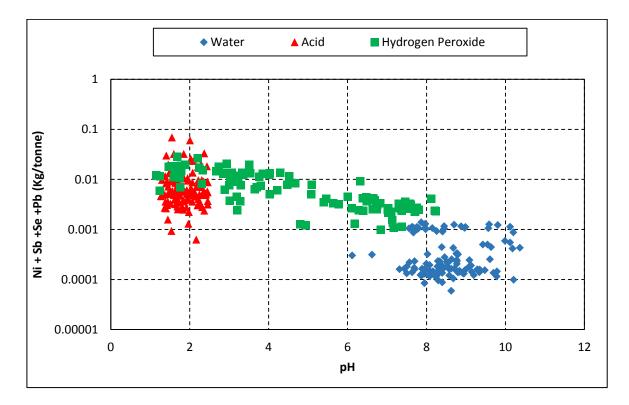


Figure 35: Plot of metal solubility (Ni + Sb + Se + Pb) vs pH

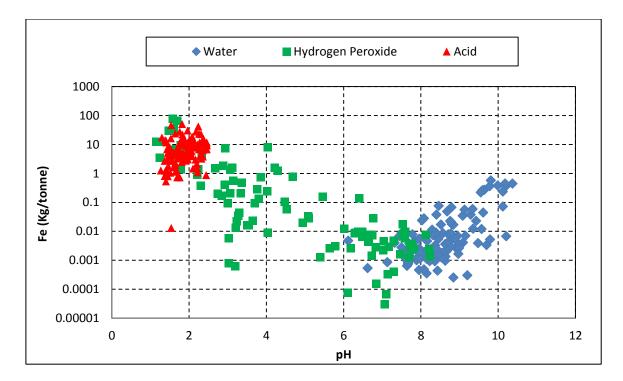


Figure 36: Plot of metal solubility (Fe) vs pH

# 5.3.2 Mine 2 leach test results

The solubility of some of the elements leached in different media (water, hydrogen peroxide and acid leach) for Mine 2 are plotted against the pH for overburden (Figure 37), interburden (Figure 38) and the coal formations (Figure 39).

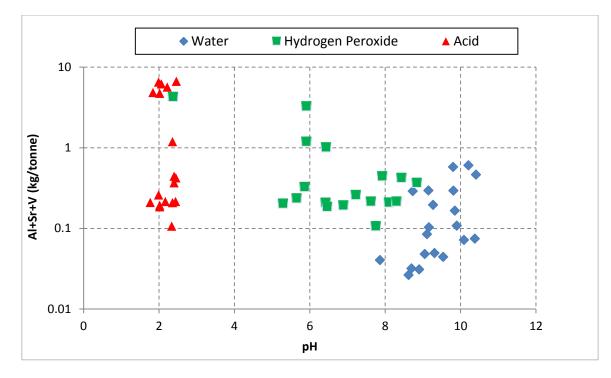


Figure 37: Plot of metal solubility (Al+Sr+V) vs pH (Overburden)

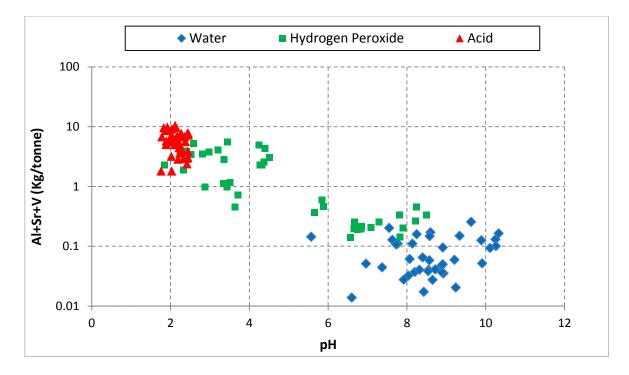


Figure 38: Plot of metal solubility (AI+Sr+V) vs pH (Interburden)

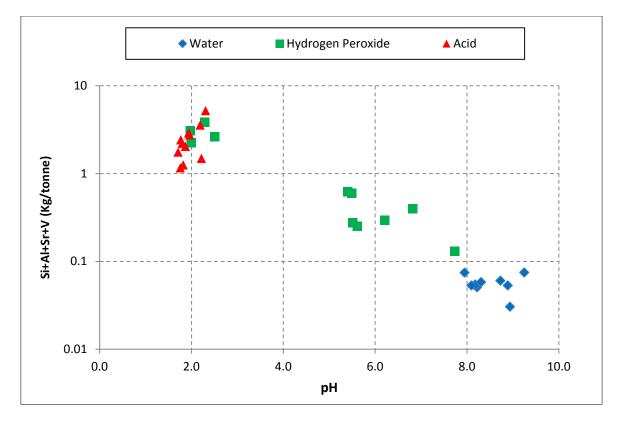
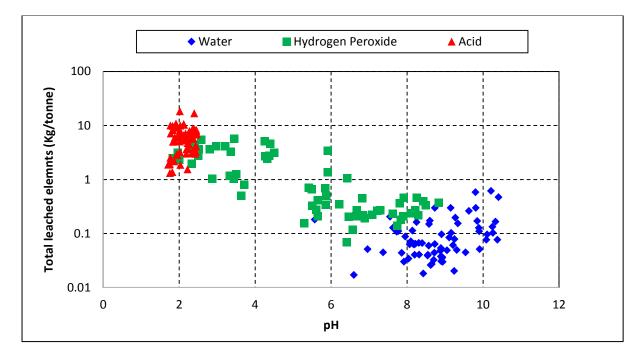


Figure 39: Plot of metal solubility (AI+Sr+V) vs pH (coal formations)

The solubility of the metals in the abovementioned graphs also indicates the influence of more metals in solution at a lower pH. From these graphs (Figure 37-39) the presence of more sulphide minerals in the interburden and coal formation are clear. The peroxide pH

results for the overburden demonstrates a minimum pH of 2.37 while for the interburden and coal formations the minimum pH values were 1.76 and 1.96.



Plots of other metals indicates the same trend (Figure 40-43)

Figure 40: Plot of metal solubility (Total leached elements) vs pH

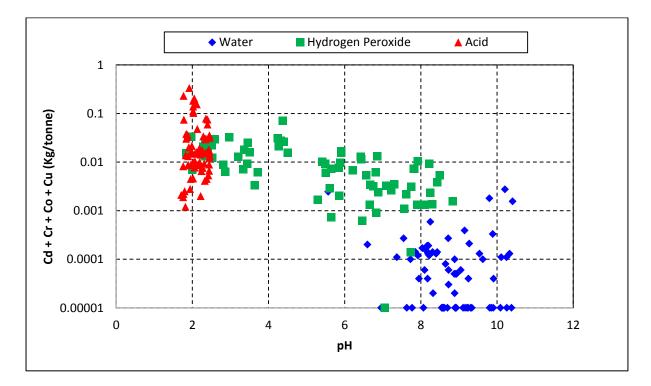


Figure 41: Plot of metal solubility (Cd + Cr +Co + Cu) vs pH

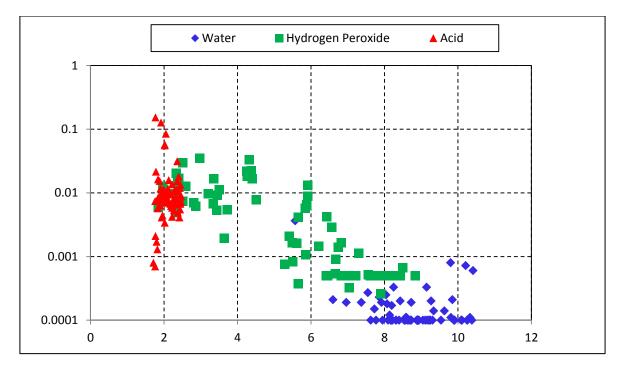


Figure 42: Plot of metal solubility (Ni + Sb + Se + Pb) vs pH

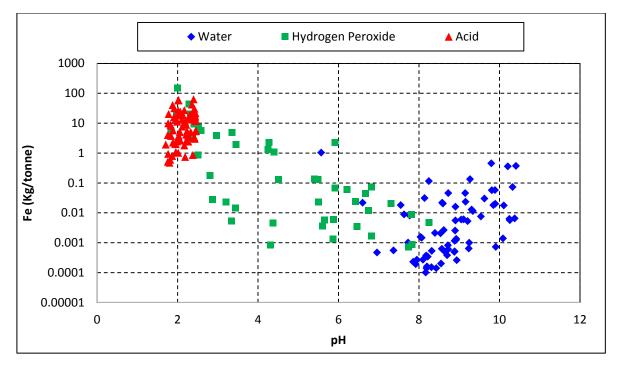


Figure 43: Plot of metal solubility (Fe) vs pH

# CHAPTER 6 CONCLUSIONS AND RECOMMENDATIONS

This study was successfully conducted to assess the acid base potential at Mine 1 and Mine 2 collieries in the Southern Highveld coalfield in Mpumalanga province of South Africa. In total, 118 and 71 borehole core samples were collected from Mine 1 and Mine 2 sites. The samples were subjected to mineralogical analysis, leach test and acid base counting tests. Site specific findings are now going to be presented. The general conclusions across the two mines are that:

- Both Mines indicated that AMD will be produced from the interburden and coal seam, therefore more care is needed when these layers are exposed during mining.
- All coal samples have a higher risk of acid generation which is attributed to the presence of pyrite in the coal formation. It can therefore be expected that most of the acid generation would be mainly confined to the coal seams. The coal will be removed in the mining process and the pyrite is usually discarded.
- The acid generation risk varies with from one borehole to another but also with depth within the boreholes as influenced by subsurface heterogeneity.
- The results indicate a higher release of elements in an acidic environment. Aluminium, iron, calcium and magnesium are released in the highest concentration in the overburden. (See summary in Table 26 and 27)

Parameter	Average kg/tonne leach water	Average kg/tonne leached acid	Ratio Acid:Water
pН	8.97	2.03	
Mn	0.10	21.50	208.76
Fe	7.28	1001.05	137.57
AI	3.28	277.82	84.73
Zn	0.04	2.44	64.90
Ni	0.01	0.86	60.85
Li	0.01	0.39	46.91
Sr	0.07	3.28	46.70
Са	10.51	392.27	37.31
Mg	5.09	165.82	32.60
Cr	0.01	0.21	32.58
Cu	0.03	0.91	26.87
Pb	0.01	0.11	9.37
K	5.66	51.38	9.07
V	0.08	0.37	4.35
Sb	0.01	0.02	3.44
Na	12.88	42.87	3.33

Table 26: Comparison of parameters released in different media (overburden, interburden and coal) at Mine 1

Parameter	Average kg/tonne leach water	Average kg/tonne leached acid	Ratio Acid:Water
Ва	0.02	0.03	1.42
As	0.01	0.01	0.56
Мо	0.02	0.00	0.11
Se	0.01	0.00	0.03

Table 27: Comparison of parameters released in different media (overburden, interburden and coal) at Mine 2

Parameter	Average kg/tonne leach water	Average kg/tonne leached acid	Ratio Acid:Water
pН	9.393	2.181	
Ni	0.000	0.021	312.92
Mn	0.001	0.320	304.37
Cr	0.000	0.013	240.54
Fe	0.078	15.164	194.04
AI	0.048	4.710	97.30
Cu	0.000	0.011	82.55
Sr	0.001	0.043	59.64
Mg	0.070	3.481	49.47
Са	0.106	4.403	41.68
V	0.001	0.017	24.38
К	0.065	0.668	10.20
Li	0.000	0.001	8.46
Na	0.308	0.948	3.08

# 6.1 Site conclusions

# 6.1.1 Mine 1

Important findings at Mine 1 site are that:

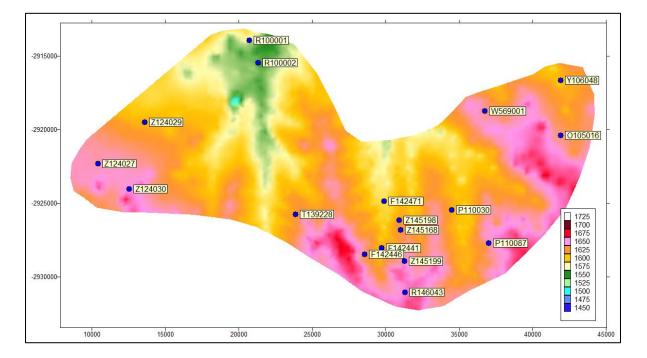
- The mineralogical results indicated that 20% of the samples contained pyrite indicating the potential to generate acid. A total of 53% of the samples contained calcite and dolomite was detected only in 12% of the samples. Both of these minerals have the potential to neutralise acid.
- Out of 118 samples analysed for Mine 1, 49% were non-acid generating, 19% had a medium risk of generating acid and the rest of the samples (32%) had a final pH of less than 3.5, which indicates a high risk of acid generation.

# 6.1.2 Mine 2

Important findings at Mine 2 site are that:

- A total of 28% of the samples contained pyrite which is the main mineral of concern with respect to acid generation. The pyrite is mainly associated with coal formation,
- Nearly half of the samples (44%) contained calcite which offers a neutralising potential to acid generation,
- Out of the 71 samples analysed from Mine 2 site, 56% were non-acid generating, while 17% samples had a medium risk of generating acid and 27% samples had a high risk of acid generation.
- From the leach test it can be concluded that Fe is the main trace element as it has the highest concentrations leached in comparison to all the other trace elements. Pyrite is the main source of Fe.

# 6.1.3 A summary/model of the acid generation risk findings for each sample collected from the Mine 1 and Mine 2 site is presented in Figure 45 - Figure 49.



• Graph 44 indicates the topographic distribution of Mine 1 and Mine 2 sampling points.

Figure 44: surface contours with borehole positions for Mine1 and Mine 2

• The NAG pH results are graphically presented (Figures 45-49) for Mines 1 and 2. The layers that had a potential for producing acid are clearly demonstrated. Layers are divided into three categories: Low, Medium and High risk.

- Eleven borehole cores at Mine 1 indicated the high risk of acid generation and three at Mine 2 (Figure 45). Samples with high risk (32%) were found to be in the interburden between 40 m and 190 m for Mine 1 and they have low NNP values.
- Mine 2 samples indicated high risk (27%) between 20 m and 60 m, and between 80 m and 180 m. These 27% samples with high risk were coal (low NNP) and sandstone samples that contain pyrite. Most of the acid generation would be mainly found at the coal seams (Figure 46-49). Therefore, more attention is needed to take precautionary measures when these layers are discarded.

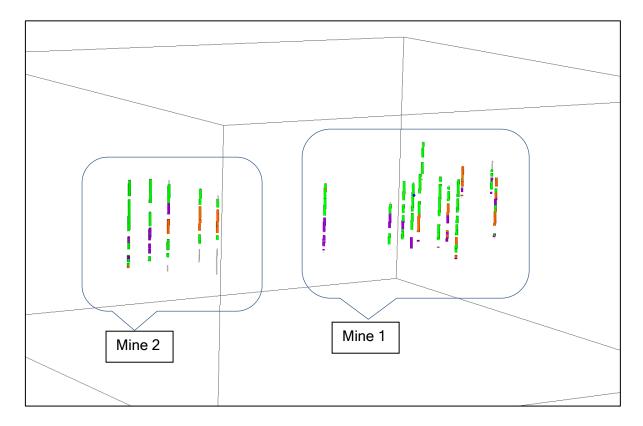


Figure 45: NAG values for each borehole (green = low risk), (Orange - medium risk) & (Purple = high risk)

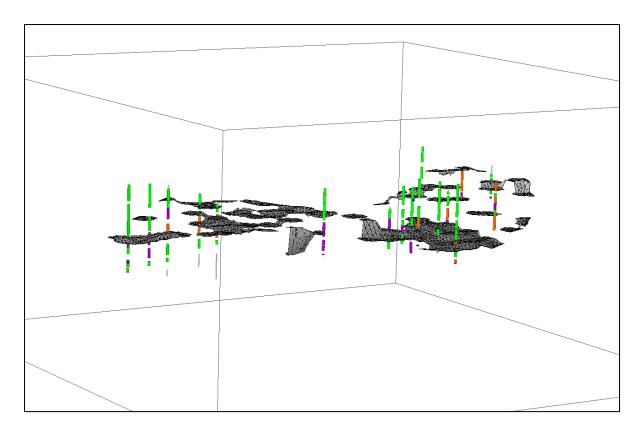


Figure 46: NAG values for each borehole with 5 seam top contours in 3D view

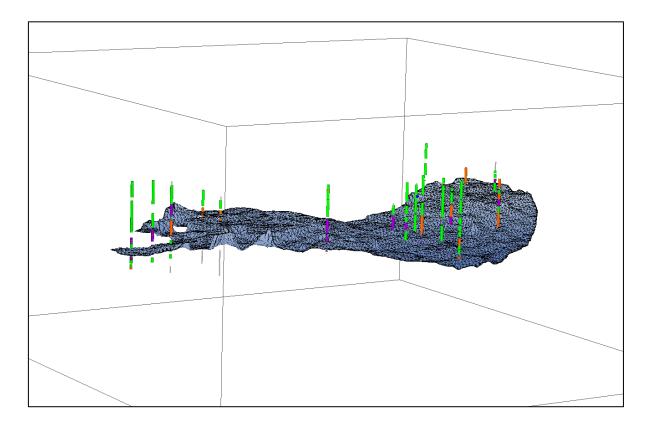


Figure 47: NAG values for each borehole with the 4 seam top contour in 3D view

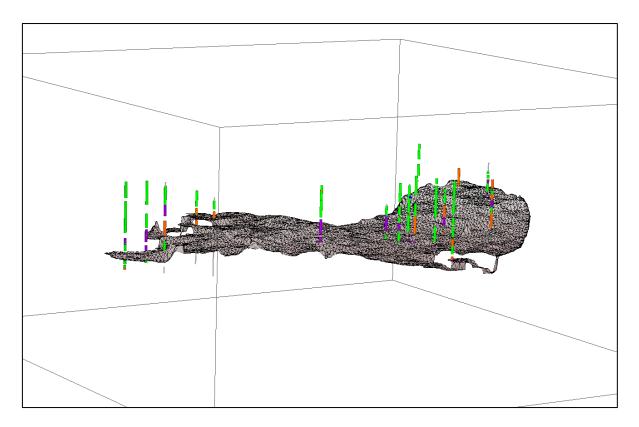


Figure 48: NAG values for each borehole with the 3 seam top contours in 3D view

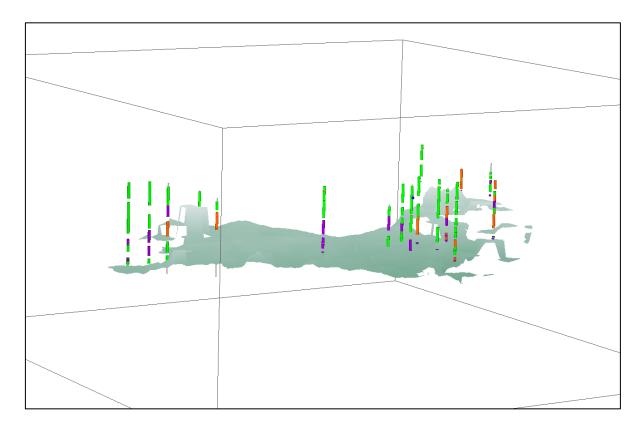


Figure 49: NAG values for each borehole with the 2 seam top contours in 3D view

Samples that contained layers of sandstone, siltstone and shale had the highest NNP values and these layers can be used as a buffer to neutralise the AMD. These layers consist of dolomite and calcite, CaO and MgO indicating that they will have a potential to neutralise the acid.

### 6.2 Recommendations

Based on lessons learned and experience gained during this study, the following recommendations can be made:

- All results that are inconclusive should be tested with other methods e.g. Kinetic test
- For all samples with high risk for acid generation, control measures need to be put in place.
- For better comparison of samples, it is important to collect samples per each meter depth across all the sampling boreholes.
- For improved decision making, samples tested with the static methods can also be subjected to Kinetic tests assessment.

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#### ABSTRACT

Pyrite, iron disulphide is the most common mineral in the metal sulphite and coal deposit. The oxidation of Pyrite and other metal-sulphide minerals by oxygen has a large environmental impact and plays a key role in Acid Mine Drainage (AMD). Environmental impact shows that soil acidity, toxic metal concentrations and vegetation damage are the predominant negative impact of AMD. AMD is a major concern for the mining industry because mining activities tend to increase the amount of rock surface exposed to air and water. Mining companies are increasingly required to evaluate the AMD potential at future mine sites and provide detailed plans to prevent or minimise AMD at all phases of mine operation as part of the environmental Impact assessment (EIA) process.

An investigation was conducted in two mine (Mine 1 and Mine 2) areas in the Southern Highveld Coalfield of Mpumalanga. Mine 1 is an underground coal mine and is situated 10 km outside Trichardt on the road to Bethel. It was established in May 2012 and its shaft supply coal to Sasol Synfuels. Mine 2 is an open-cast coal mine and is situated between 2 towns, amely Trichardt and Kriel. Construction activities started in 1990 and the mine reached full production in September 1992. Both Mines fall in the Karoo Supergroup which comprises of Ecca group formation and consist dominantly of sandstone, siltstone, shale and coal.

The aim of this study was to investigate the acid-base potential of these two Mines. 118 samples were collected from Mine 1 borehole core and 71 samples were collected from Mine 2 borehole core to conduct mineralogical and Acid Base Accounting (ABA) analysis. Acid-base potential leachate were further analysed for major and trace elements using Inductively Coupled Plasma-Optical Emission Spectrometer (ICP-OES).

Most of the samples analysed for Mine 1 have been found to have High AMD risk at the interburden and these samples have low NNP values, all these samples consisted of coal and they contained pyrite mineral as indicated in the mineralogical analysis. These samples must be taken into consideration to minimise oxidation.

Samples (Mine 1) that contained layers of sandstone, siltstone and shale have the highest NNP values and these layers can be used as a buffer to neutralise the AMD. Therefore Mine 1 will produce an AMD at the coal seam once exposed during mining.

Mine 2 shows that 3 samples have High AMD risk at the interburden while 2 boreholes show the risk between 20m and 60m, and these samples were coal samples and others were sandstone samples and they contained pyrite. These samples that have coal showed low NNP. Therefore most of the acid generation would be mainly found at the coal seams. Layers that show high NNP values consisted of sandstone, siltstone and shale. They showed no indication of acid generation, therefore they will work as buffers to neutralise the any AMD that will be produced. Mine 2 also indicate that AMD will be produced at the interburden at layers that contains coal, therefore more consideration is needed when these layers are exposed during mining.

With such condition it is possible for the mines to predict the types of situations that might arise concerning groundwater quality, and implement proper prevention or remediation programs.

## APPENDICES

# Appendix 1: List of samples for Mine 1

Sample Name	Lab Number	Depth (m)	
F142441	TC18	1.7-34.48	
	TC75	34.48-49.07	
	TC38	49.07-70.25	
	TC83	70.6-82.75	
	TC7	83.45-121.47	
	TC35	125.77-128.56	
F142446			
1 172770	TC66	5.8-9	
	TC57	9.15-34.95	
	TC62	34.95-70.5	
	TC10	70.5-82.44	
	TC4	82.44-91.10	
	TC51	91.10-112.10	
	TC41	112.10-113	
	TC33	113.12-155.3	
	TC87	155.3-161.50	
F142471	TC65	4.68-12.9	
	TC55	12.9-18.82	
	TC61		
	TC3	18.82-40.85 42.38-75.68	
	TC107 <sup>b</sup>	76.09-80.80	
	TC11	83.8-84.15	
	TC112 <sup>b</sup>	84.15-84.80	
	TC84	84.8-109.4	
	TC105 <sup>b</sup>	109.4-109.90	
	TC114 <sup>b</sup>	C5	
	10114		
O105016	TC25	7.3-20.26	
	TC113	20.26-33.5	
	TC48	33.5-41.65	
	TC43	41.65-43.55	
	TC101	43.55-45.16	
	TC34	45.16-52.82	
	TC24	52.82-58.99	
	TC14	58.99-77.95	
	TC79	80.3-85.56	

a = same sample

b = coal sample

Sample Name	Lab Number	Meters
P110030	TC49	3.8-16.23
	TC52 <sup>a</sup>	16.23-20.7
	TC99 <sup>a</sup>	16.23-20.7
	TC1	20.7-30.23
	TC5	30.23-42.86
	TC74	67.54-68.06
	TC117 <sup>b</sup>	67.54-69.54
	TC104	69.54-106.77
	TC109 <sup>b</sup>	106.38-110.22
	TC82	106.77-109.99
	TC108 <sup>b</sup>	112.75-113.33
	TC27	109.99-120.22
	TC98	120.22-141.42
	TC106 <sup>b</sup>	141.42-141.55
P110087	TC9	6-9.93
	TC13	9.93-30.18
	TC97	30.18-40.32
	TC58	40.32-54.77
	TC23	54.77-71.87
	TC94	71.87-89.08
	TC91	90.19-90.45
	TC45	90.45-130.50
	TC53	130.50-133.19
	TC6	133.19-133.97
	TC73	139.15-140.18
R146043	TC93	5.33-8.17
	TC116	8.17-47.07
	TC47	47.07-78.5
	TC59	78.5-94.33
	TC29	94.33-101.9
	TC16	101.9-125.08
	TC28	125.08-125.92
	TC110 <sup>b</sup>	125.65-125.92
	TC40	125.92-140.78
	TC15	140.78-141.52
	TC86	143.28-160.68
	TC92	160.68-160.98
	TC21	161.15-167.07
	TC118 <sup>b</sup>	167.07-170.34

Sample Name	Lab Nunber	Meters
T139228	TC70	2.2-7.35
	TC44	7.35-39.45
	TC36	39.45-42.87
	TC63	42.87-43.5
	TC30	43.5-82.2
	TC39	82.2-83.81
	TC76	83.81-115.08
	TC96	115.08-117.67
	TC78	115.45-117.67
	TC56	121.35-141.39
	TC60	141.39-145.24
W569001	TC69	2.4-7.4
	TC80	18.7-19.5
	TC54	19.5-55.81
	TC111 <sup>b</sup>	55.81-59.57
	TC89	59.57-62.8
	TC102 <sup>b</sup>	62.8-63.03
	TC22	63.03-90.4
	TC103 <sup>b</sup>	90.4-90.8
Y106048	TC32	5.95-6.66
	TC12	6.66-22.61
	TC8	22.61-45.06
	TC95	50.59-70.93
	TC19	70.93-75.27
Z145168	TC31	1.6-6.1
	TC115	6.1-49.68
	TC100	49.68-56.74
	TC37	56.74-57.36
	TC68	57.36-68.81
	TC85	68.81-94.75
	TC67	96.22-136.4
	TC42	139.31-142.08
Z145198	TC71	3.33-45.04
	ТС90	45.04-90.02
	TC26	91.03-127.78
	TC2	132.14-134.94
	TC72	134.94-151.78
	TC88	151.78-184

Sample Name	Lab Number	Meters
Z145199	TC46	0-22.52
	TC17	22.52-28.75
	TC50	28.75-52.14
	TC77	52.14-123.92
	TC81	123.92-124.6
	TC20	125.80-159.6
	TC64	160.3-162.3

# Appendix 2: List of samples for Mine 2

## b = coal sample

Sample name	Lab number	Depth (m)
R 100 001	SF 11	1.62-12.46
	SF 12	12.46-24.48
	SF 13	24.48-25.26
	SF 61 <sup>b</sup>	24.48-25.80
	SF 14	25.80-52.66
	SF 1	50.66-51.6
	SF 62 <sup>b</sup>	51.6-56.92
	SF 2	56.92-57.14
	SF 15	57.14-85.28
	SF 16	86.82-112.94
	SF 17	112.94-116.23
	SF 18	116.23-130.1
	SF 19	130.1-143.1
R 100 002	SF 20	14.1-18
	SF 21	18.3-37.5
	SF 3	37.5-37.82
	SF 63 <sup>b</sup>	37.82-43.08
	SF 4	43.08-45.72
	SF 22	45.72-70.1
	SF 23	71.4-103.8
	SF 24	103.8-124.4
Z124027	SF 45	4.13-14.04
	SF 46	14.04-48.99
	SF 47	48.99-69.26
	SF 48	69.26-90.34
	SF 49	90.44-91.84
	SF 64 <sup>b</sup>	91.84-92.34m
	SF 50	92.34-93.61
	SF 51	93.61-123.04
	SF 65 <sup>b</sup>	123.08-123.68
	SF 52	123.68-125.28
	SF 7	125.28-125.88
	SF 66 <sup>b</sup>	125.88-129.68
	SF 8	129.78-134.51
	SF 53	134.48-162.76
	SF 54	162.54-164.8
	SF 55	167.2-172.87
	SF 56	172.87-180.67

Sample name	Lab no.	Meters	
Z124027	SF 57	180.67-183.7	
	SF 58	183.7-186.86	
	SF 59	186.86-189.02	
	SF 60	189.02-200.16	
Z124029	SF 25	6.9-8.1	
	SF 26	8.1-61.46	
	SF 27	61.46-73.68	
	SF 28	73.68-74.72	
	SF 29	74.72-98.23	
	SF 5	98.23-98.88	
	SF 67 <sup>b</sup>	98.88-102.83	
	SF 6	102.83-104.6	
	SF 30	104.6-132.74	
	SF 31	133.63-136.28	
	SF 32	138.64-158.45	
	SF 33	158.45-165.43	
	SF 34	165.43-174.53	
Z124030	SF 35	11.6-12.52	
	SF 36	12.52-20.8	
	SF 37	20.8-41.35	
	SF 38	41.35-49	
	SF 39	49-98.3	
	SF 68 <sup>b</sup>	99.20-99.42	
	SF 40	98.3-100.45	
	SF 69 <sup>b</sup>	100.45-100.80	
	SF 41	100.80-133.37	
	SF 70 <sup>b</sup>	133.37-133.92	
	SF 9	133.92-135.56	
	SF 71 <sup>b</sup>	135.56-139.83	
	SF 10	139.83-140.6	
	SF 42	140.60-164.48	
	SF 43	167.1-168.2	
	SF 44	168.2-185.66	

# Appendix 3: Mine 1 XRD Mineralogical analysis

Sample	Lab no.	Quartz	Plagio clase	K- feldspar and/or rutile	Pyro xene	Kaolinite and/ or chlorite	Mica	Side	Anke rite	Mag netite	Goethite and/or hematite and/or pyrite	Cal cite	Sme	Anal cime	Mag nesite	Dolo mite
F142441	TC18			Tuule		Chlonte	Mica	nte		neute	pyrite	Cite		cine	Tiesite	Tinte
	TC75	xx XX	XX		xx				x				x			
	TC38		x	XX		x	x		x							
	TC83	x	XX	Х		XX	x	x				x	x			
	TC7	X	x	x		ХХ	xx	x	x			x				
	TC35	X	хх	XX		хх	x	XX					x			
5140440	1035	XX		хх	1	x			<x< td=""><td>I</td><td></td><td>1</td><td>I</td><td></td><td>   </td><td><u> </u></td></x<>	I		1	I		 	<u> </u>
F142446	TC66	хх	x	xx			x					<x< td=""><td></td><td></td><td></td><td></td></x<>				
	TC57	хх	x	хх		x	x		x			x	<x< td=""><td></td><td></td><td></td></x<>			
	TC62	хх	хх		х										x	
	TC10	хх	хх	xx		x	x			<x< td=""><td></td><td></td><td></td><td></td><td></td><td></td></x<>						
	TC4	хх	хх	x		<x< td=""><td>x</td><td></td><td>x</td><td></td><td></td><td><x< td=""><td></td><td></td><td></td><td></td></x<></td></x<>	x		x			<x< td=""><td></td><td></td><td></td><td></td></x<>				
	TC51	х	хх	xx		xx	x	x			<x< td=""><td>x</td><td><x< td=""><td></td><td></td><td></td></x<></td></x<>	x	<x< td=""><td></td><td></td><td></td></x<>			
	TC41	xx		x		x	x						x			
	TC33	xx	xx	xx		xx	x	x				x	<x< td=""><td></td><td></td><td></td></x<>			
	TC87	х		хх		x	xx		x							
F142471	TOFF					1										
	TC65	XX	x	x		x	x	x	x			<x< td=""><td></td><td></td><td></td><td></td></x<>				
	TC55	XX	x	XX			x									
	TC61	Х	x	xx		XX	x	x				<x< td=""><td>x</td><td></td><td></td><td></td></x<>	x			
	TC3 TC107 <sup>b</sup>	Х	XX	xx		XX	x	x			<x< td=""><td>x</td><td><x< td=""><td></td><td></td><td></td></x<></td></x<>	x	<x< td=""><td></td><td></td><td></td></x<>			
	-	Х				Х						x				x
	TC11	XX		XX		Х	x									
	TC112 <sup>b</sup>	Х				Х					xx	x				x
	TC84 TC105 <sup>b</sup>	Х	x	Х		XX	xx	X				x				
	TC105 <sup>a</sup> TC114 <sup>b</sup>	x				X			x		x	x				
0105010	10114	х	xx	xx		х		I	x	I	I	I	I		I	x
O105016	TC25	хх	xx			<x< td=""><td>x</td><td></td><td>x</td><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td></x<>	x		x			x				
	TC113	х	хх	x		хх	xx	x					хх			
	TC48	xx	хх	хх			<x< td=""><td>x</td><td></td><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td></x<>	x				x				
	TC43	x	xx	x		xx	x	xx	x							
	TC101	xx	x	хх		хх	xx	x				x	xx			
	TC34	xx	x	xx		xx	x									
	TC24	хх	х	x	xx	x	x		x				x			
	TC14	xx	xx	хх		x	x	x				xx	<x< td=""><td> </td><td></td><td></td></x<>			
	TC79	х	x	x		х	xx		<x< td=""><td></td><td></td><td><x< td=""><td></td><td></td><td></td><td></td></x<></td></x<>			<x< td=""><td></td><td></td><td></td><td></td></x<>				

				K- feldspar		Kaolinite					Goethite and/or hematite					
Sample Name	Lab number	Quartz	Plagio clase	and/or rutile	Pyro xene	and/or chlorite	Mica	Side rite	Anke rite	Mag netite	and/or pyrite	Cal cite	Sme ctite	Anal cime	Mag nesite	Dolo mite
P110030	TC49	xx	x	xx			x		x			<x< td=""><td></td><td></td><td></td><td></td></x<>				
	TC52 <sup>a</sup>	хх	x	xx		xx	x	x			<x< td=""><td>x</td><td>x</td><td></td><td></td><td></td></x<>	x	x			
	TC99 <sup>ª</sup>	xx	х		xx			x	x				xx			
	TC1	x	x	xx		xx	x	x				<x< td=""><td>x</td><td></td><td></td><td></td></x<>	x			
	TC5	xx	хх	x		x	x		<x< td=""><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td></x<>			x				
	TC74	х	xx	xx		x	х	xx								
	TC117 <sup>b</sup>	х	x	xx		x	xx	x	x			x				
	TC104	хх		хх		x	xx		x							
	TC109 <sup>b</sup>	хх		х		xx	x		<x< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td></x<>							
	TC82	х	x	хх		x	xx	x	x							
	TC108 <sup>b</sup>	х				x					х	x				x
	TC27	х		x		x					х	x				x
	TC98	х		x		x	xx				xx	x				x
	TC106 <sup>b</sup>	хх	x	x		xx	xx	x				x	xx			x
P110087	-					1		1				ł	1	1	r	
1 110007	TC9	хх	х	<x< td=""><td>x</td><td>x</td><td>x</td><td></td><td>x</td><td></td><td></td><td>хх</td><td>x</td><td></td><td></td><td></td></x<>	x	x	x		x			хх	x			
	TC13	хх	хх	хх		x	x	x					<x< td=""><td></td><td></td><td></td></x<>			
	TC97	х	x	хх		xx	хх	x	x				xx			
	TC58	ХХ	x	хх		xx	x	x								
	TC23	ХХ	x	хх		x	<x< td=""><td></td><td>x</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></x<>		x							
	TC94	х	x	хх		xx	хх	x				x				
	TC91	Х	хх	хх	x	xx	хх	xx			x					
	TC45	хх	хх	хх		x	x	x								
	TC53	ХХ	x	x		хх	x	x			x		<x< td=""><td></td><td></td><td></td></x<>			
	TC6	Х		<x< td=""><td></td><td>x</td><td>хх</td><td>xx</td><td></td><td></td><td>x</td><td>x</td><td></td><td></td><td></td><td></td></x<>		x	хх	xx			x	x				
	TC73	Х		x		х	хх		x		x	x				
R146043	TC93															
	TC116	XX	x	XX			x		x			x				
	TC47	x	х		XX	~							XX			XX
	TC59	X	x	x		Х	xx				x		<x< td=""><td></td><td></td><td></td></x<>			
	TC29	XX	xx	X		x	<x< td=""><td>x</td><td>x</td><td></td><td><x< td=""><td>x</td><td></td><td></td><td></td><td></td></x<></td></x<>	x	x		<x< td=""><td>x</td><td></td><td></td><td></td><td></td></x<>	x				
	TC29	XX	x	x		x	x	x				xx				
	TC16	XX	x	xx		x	x	x				x				
	TC28	XX	XX	xx		x	<x< td=""><td></td><td>x</td><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td></x<>		x			x				
	TC40	XX		x		x					x					
	TC40 TC15	XX	x	X		x	x		<x< td=""><td></td><td></td><td><x< td=""><td></td><td></td><td></td><td></td></x<></td></x<>			<x< td=""><td></td><td></td><td></td><td></td></x<>				
	TC86	X	x	xx		XX	XX	x	x			x				
	TC80	X	xx	x		X	XX				x					
	TC92 TC21	X	x	XX		XX	xx		x			x	x			
	TC21 TC118 <sup>b</sup>	X		x		X	xx	x			x	x				x
	IUIIÕ	Х	1		1	Х	XX	I	1	1	х	х	1			

											Goethite					
	Lab		Plagio	K- feldspar and/or	Pyro	Kaolinite and/or		Side	Anke	Mag	and/or hematite and/or	Cal	Sme	Anal	Mag	Dolo
Sample Name T139228	number	Quartz	clase	rutile	xene	chlorite	Mica	rite	rite	netite	pyrite	cite	ctite	cime	nesite	mite
	TC70	XX	xx	xx			хх									<u> </u>
	TC44	xx	х		xx	x			xx				x			
	TC36	ХХ	x	x			хх				x	xx	x			
	TC63	x	ХХ		xx				x	x		x				
	TC30	хх	x	xx		x	x		x			x				<u> </u>
	TC39	xx	x	xx		xx	x	x								
	TC76	х	x	xx		xx	хх	x	x		x					ļ
	TC96	х		x		xx	хх	xx			x	x				ļ
	TC78	xx		x		xx	x		x		x					
	TC56	хх		xx		xx	x	x								
	TC60	х	x	xx		x	хх		x			<x< td=""><td></td><td></td><td></td><td></td></x<>				
W569001	TC69	~	~		×	~										
	TC80	×	X		Х	x	xx		x							
	TC54	X	XX			X	X									
	TC111 <sup>b</sup>	Х	XX	XX		x	x	x				x				
	TC89	ХХ				ХХ			x			x	X	X		x
	TC102 <sup>b</sup>	Х		XX		X	XX		x			<x< td=""><td></td><td></td><td></td><td></td></x<>				
		Х		x		Х	XX				XX	x				x
	TC22	XX	x	xx		XX	x	x				x	<x< td=""><td></td><td></td><td></td></x<>			
	TC103 <sup>b</sup>	Х		x		XX					Х	x	Х			x
Y106148	TC32	ХХ	x	xx		xx	x		x							
	TC12	XX	xx	х		x	<x< td=""><td>x</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td><u> </u></td></x<>	x								<u> </u>
	TC8	х	xx	x		xx	x	x					<x< td=""><td></td><td></td><td><u> </u></td></x<>			<u> </u>
	TC95	х	x	x		х	xx	x	x			x				x
	TC19	ХХ		xx							x					
Z145168	TC31	xx	x	x	[	x	×	[	<x< td=""><td>[</td><td></td><td>1</td><td>[</td><td>[</td><td>[</td><td></td></x<>	[		1	[	[	[	
	TC115	xx	xx	^		^	^		~^	xx		x	xx		x	
	TC100	x	x	x		~~	x								^	
	TC37	xx	xx	~ <x< td=""><td>x</td><td>XX</td><td>xx</td><td>x</td><td>~</td><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td></x<>	x	XX	xx	x	~			x				
	TC68	x			^			^	x							
	TC85	x	x	XX		xx	xx	v				x				x
	TC67		xx	XX		xx	xx	x								
	TC42	x	x	XX	x	xx	xx	x	x		x	x	x			
Z145198		_ X	xx	I	I	X	x	I	I	I	I	I	I	I	I	L
2140190	TC71	хх	х		х							x	хх			<u> </u>
	TC90	х	x	хх		хх	xx	x				x				<u> </u>
	TC26	XX	xx	хх		x	x	x				<x< td=""><td></td><td></td><td></td><td><u> </u></td></x<>				<u> </u>
	TC2	хх		хх		х	x	<x< td=""><td>ļ</td><td>ļ</td><td></td><td></td><td>ļ</td><td><b> </b></td><td>ļ</td><td><b> </b></td></x<>	ļ	ļ			ļ	<b> </b>	ļ	<b> </b>
	TC72	х	x	х		х	xx		x			<x< td=""><td> </td><td> </td><td> </td><td><u> </u></td></x<>				<u> </u>
	TC88	xx		xx		x	x	<x< td=""><td>x</td><td></td><td></td><td><x< td=""><td></td><td></td><td></td><td></td></x<></td></x<>	x			<x< td=""><td></td><td></td><td></td><td></td></x<>				

Sample Name	Lab number	Quartz	Plagio clase	K- feldspar and/or rutile	Pyro xene	Kaolinite and/or chlorite	Mica	Side rite	Anke rite	Mag netite	Goethite and/or hematite and/or pyrite	Cal cite	Sme ctite	Anal cime	Mag nesite	Dolo mite
Z145199	TC46	x	xx		xx				x				x			
	TC17	xx	x	x	701		x		X	x			x			
	TC50	xx	хх	<x< td=""><td>xx</td><td></td><td></td><td></td><td>x</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></x<>	xx				x							
	TC77	x	x	xx		xx	хх	x				x				
	TC81	х	xx	xx		х	xx	x								
	TC20	х	xx	xx		x	x	x	x			x				
	TC64	xx		x		x	x		x		<x< td=""><td></td><td></td><td></td><td></td><td></td></x<>					

Sample	Lab number	QUARTZ	PLAGIOCLASE	<b>K-FELDSPAR/RUTILE</b>	CALCITE	SIDERITE	ANATASE	ANKERITE	PYRITE	ILMENITE	KAOLINITE	CLINOCHLORE	MICA	ILLITE/SMECTITE INTERSTRATIFICATION	SMECTITE	PYROXENE	MAGNETITE	PYROPHYLLITE	SERPENTINE	AMPHIBOLE	ANDALUSITE	ALUNITE
R 100 001	SF 11	xx	xx									хх				xx						
	SF 12	x	x	x	x							хх	xx	xx								
	SF 13	x	x	x								хх	х		х							
	SF 61 <sup>b</sup>	x	х		x						х											
	SF 14	x	x	xx	x	x			x			xx	xx	xx								
	SF 1	x	x	xx		x		x				х	xx	xx								
	SF 62 <sup>b</sup>	x			x				x		х											
	SF 2	xx		x				x			х		x									
	SF 15	x	x	xx								х	xx	x								
	SF 16	xx	xx								xx		x	x								
	SF 17	xx		x	<x< td=""><td></td><td></td><td></td><td></td><td></td><td>xx</td><td></td><td>x</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></x<>						xx		x									
	SF 18	x	x	xx		xx		x			xx		xx									
	SF 19	х	xx	x	x	xx		x			x		xx									
	•		1				-	-	1	-	-	-	1	•		1	•			-		
R 100 002	SF 20	х	xx		x						xx		xx		х	xx						
	SF 21	х	x	xx	x	x						х	хх	xx	x							
	SF 3	х	x	xx	<x< td=""><td>x</td><td></td><td></td><td></td><td></td><td></td><td>х</td><td>хх</td><td>xx</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></x<>	x						х	хх	xx								
	SF 63 <sup>b</sup>	х		xx					x		х											
	SF 4	х		xx					x		хх		хх	x		x						
	SF 22	х	x	xx		x		x				хх	xx	x								
	SF 23	xx		xx							xx		x									
	SF 24	xx	x	x		x					хх		x									
Z124027	1	r	1	1	1	1	1	1	1	1	1	1	1	1		1	1	1	1	1	1	
2124021	SF 45	хх	х									х				xx						
	SF 46	х	x	x	хх	x						хх	xx	xx					xx			
	SF 47	х	x	xx								хх	xx	x								
	SF 48	х	x	x	x	x		хх				хх	xx	xx								
	SF 49	х	x	x					хх		xx		xx		х							
	SF 64 <sup>b</sup>	х	хх		хх				x		х											
	SF 50	хх	хх			хх										xx						
	SF 51	х	x	xx	x	x						хх	xx	xx								
	SF 65 <sup>b</sup>	х	xx	xx					x		х											
	SF 52	xx		x	x			хх	<x< td=""><td></td><td>xx</td><td></td><td>x</td><td>x</td><td> </td><td></td><td> </td><td></td><td></td><td></td><td></td><td></td></x<>		xx		x	x								
	SF 7	хх		x	<χ				x		x		x									

Sample	Lab number	QUARTZ	PLAGIOCLASE	FELDSPAR/RUTIL E	CALCITE	SIDERITE	ANATASE	ANKERITE	PYRITE	ILMENITE	KAOLINITE	CLINOCHLORE	MICA	ILLITE/SMECTITE INTERSTRATIFIC ATION	SMECTITE	PYROXENE	MAGNETITE	PYROPHYLLITE	SERPENTINE	AMPHIBOLE	ANDALUSITE	ALUNITE
Z124030	SF 71 <sup>b</sup>	x			xx	x					х											
	SF 10	х		xx						x	х		xx	x		x						
	SF 42	x	x	x		x		xx				х	xx	xx								
	SF 43	х	x	x	x							xx	xx	xx	xx							
	SF 44	xx	x		x	x						xx	xx		xx				xx	хх		

# Appendix 5: Mine 1 XRF Mineral Analysis

Sample name		SiO2	TiO2	AI2O3	Fe2O3	MgO	MnO	CaO	Na2O	к20	P2O5	TOTAL
F142441	TC18	52.1	1.4	13.3	12.4	6.5	0.2	8.9	2.3	0.8	0.2	98.1
	TC75	82.0	0.2	9.9	1.9	0.4	0.0	0.9	0.7	3.6	0.1	99.9
	TC38	68.2	0.7	14.1	4.5	1.0	0.1	0.9	0.8	3.3	0.1	93.7
	TC83	63.7	0.9	14.8	5.5	1.0	0.1	1.1	1.1	3.0	0.1	91.1
	TC7	54.0	0.6	12.2	8.1	2.0	0.1	3.4	0.4	2.9	0.4	84.1
	TC35	76.4	0.7	13.4	0.6	0.2	0.0	0.2	0.0	2.4	0.0	93.9
		•					•					
F142446	TC66	82.4	0.3	9.3	2.0	0.5	0.0	0.3	1.8	3.0	0.0	99.7
	TC57	72.7	0.6	12.4	3.1	1.0	0.1	1.9	1.8	3.5	0.1	97.2
	TC62	53.1	1.5	14.0	12.5	5.8	0.2	8.9	2.5	1.0	0.2	99.6
	TC10	70.4	0.6	11.9	3.7	0.9	0.1	1.1	0.6	2.7	0.1	92.0
	TC4	77.7	0.4	9.3	1.2	0.2	0.0	0.7	0.8	3.5	0.0	93.9
	TC51	58.6	0.8	15.6	7.7	1.5	0.1	0.9	0.7	3.1	0.1	89.1
	TC41	39.7	0.7	17.8	3.8	1.3	0.0	0.2	0.4	2.0	0.0	66.0
	TC33	69.7	0.7	13.5	4.2	1.6	0.1	1.1	1.0	3.2	0.1	95.1
	TC87	70.0	0.8	18.4	1.0	0.4	0.0	0.5	0.9	3.3	0.0	95.2
F142471	TC65	73.6	0.5	10.9	6.4	0.8	0.1	0.9	0.4	2.7	0.2	96.0
	TC55	85.6	0.5	8.5	1.6	0.2	0.0	0.5	0.4	3.4	0.0	100.9
	TC61	59.8	0.7	16.3	7.3	1.7	0.1	0.7	1.3	3.5	0.1	91.
	TC3	60.4	0.8	15.3	7.5	1.5	0.1	0.5	1.1	3.2	0.1	90.
	TC107 <sup>b</sup>	20.2	0.5	8.5	0.6	0.5	0.0	1.3	0.3	0.2	0.1	32.
	TC11	66.7	0.8	20.5	1.0	0.1	0.0	0.1	0.9	2.5	0.0	92.
	TC112 <sup>b</sup>	5.4	0.2	3.8	3.5	0.2	0.0	0.4	0.1	0.0	0.0	13.4
	TC84	67.0	0.6	14.0	2.6	1.2	0.1	2.8	0.1	2.8	0.1	91.:
	TC105 <sup>b</sup>	4.2	0.1	2.9	0.1	0.2	0.0	0.6	0.1	0.1	0.0	8.
	TC114 <sup>b</sup>	22.3	0.4	9.8	2.2	0.7	0.0	0.4	0.6	1.0	0.0	37.4
O105016	TC25	72.4	0.4	10.9	2.5	0.6	0.0	4.4	0.8	3.5	0.0	95.
	TC113	54.4	0.9	17.4	7.9	1.5	0.1	1.4	1.3	3.0	0.2	88.
	TC48	77.2	0.3	10.7	2.1	0.4	0.0	2.0	1.5	3.3	0.0	97.
	TC43	46.9	0.7	15.4	12.0	1.8	0.1	0.9	0.8	2.5	0.1	81.
	TC101	39.4	0.6	15.0	9.4	1.5	0.1	1.0	0.8	1.7	0.1	69.
	TC34	72.2	0.7	15.9	4.0	1.2	0.0	0.5	0.4	3.3	0.1	98.
	TC24	52.0	1.2	14.3	11.8	5.1	0.2	9.4	1.9	0.9	0.1	96.9
	TC14	64.0	0.8	15.0	5.0	1.1	0.1	1.7	1.3	3.1	0.1	92.
	TC79	71.9	0.7	16.6	1.0	0.3	0.0	0.1	0.0	3.0	0.0	93.3

Sample name		SiO2	TiO2	AI2O3	Fe2O3	MgO	MnO	CaO	Na2O	к20	P2O5	TOTAL
P110030	TC49	80.3	0.4	9.8	3.0	0.7	0.0	1.1	0.3	2.4	0.1	98.1
	TC52 <sup>a</sup>	65.8	0.6	13.4	4.4	0.8	0.1	0.7	0.8	2.8	0.1	89.6
	TC99ª	51.3	1.6	13.2	13.7	5.0	0.2	8.7	2.3	0.9	0.2	97.1
	TC1	56.2	0.9	17.5	8.0	2.0	0.1	1.0	0.8	3.3	0.2	89.9
	TC5	83.3	0.3	9.2	1.2	0.4	0.0	1.3	0.5	3.4	0.0	99.6
	TC74	38.6	0.6	15.7	9.0	1.5	0.1	0.5	0.3	2.0	0.0	68.5
	TC117 <sup>b</sup>	21.1	0.3	8.4	1.8	0.6	0.0	0.5	0.4	0.9	0.0	34.1
	TC104	56.3	0.7	17.6	4.3	0.9	0.0	0.4	1.3	3.2	0.1	84.9
	TC109 <sup>b</sup>	23.9	0.4	9.6	1.0	0.6	0.0	1.3	0.3	0.6	0.2	37.9
	TC82	70.3	0.5	18.7	0.6	0.2	0.0	0.1	0.0	3.5	0.0	93.9
	TC108 <sup>b</sup>	6.8	0.1	3.9	2.3	0.3	0.0	0.6	0.1	0.1	0.0	14.3
	TC27	75.9	0.7	14.8	0.9	0.2	0.0	0.2	0.0	3.4	0.0	96.1
	TC98	65.1	0.7	16.1	2.7	0.7	0.0	0.3	1.2	2.6	0.1	89.7
	TC106 <sup>b</sup>	2.4	0.1	1.5	5.0	0.3	0.0	0.8	0.0	0.0	0.0	10.1
P110087		1										
1 110007	ТС9	45.2	0.8	13.1	10.6	6.4	0.2	13.7	1.6	1.2	0.1	92.8
	TC13	67.3	0.8	14.9	5.2	1.2	0.1	0.6	0.9	3.4	0.1	94.5
	TC97	49.1	0.5	12.7	5.4	1.3	0.1	11.8	0.7	2.2	0.8	84.5
	TC58	69.7	0.7	14.3	4.7	1.1	0.1	0.7	0.9	2.9	0.1	95.1
	TC23	84.1	0.2	9.4	1.0	0.2	0.0	0.2	0.6	4.1	0.0	99.7
	TC94	58.9	0.6	13.8	6.5	1.6	0.1	4.5	1.6	3.0	0.3	90.8
	TC91	31.7	0.5	12.8	5.6	1.0	0.0	0.4	0.4	1.4	0.0	53.8
	TC45	72.6	0.5	12.8	4.2	0.7	0.0	0.4	0.9	3.5	0.0	95.6
	TC53	50.4	0.9	18.3	8.1	1.4	0.1	0.5	0.7	2.7	0.1	83.1
	TC6	47.0	0.5	4.0	22.2	1.4	0.1	2.2	0.0	1.6	0.0	79.0
	TC73	57.7	0.9	18.0	2.0	0.8	0.0	0.5	0.0	2.2	0.0	82.1
R146043		1			[		[			[	1	
	TC93	78.0	0.6	9.6	2.3	0.8	0.0	1.0	1.0	3.3	0.1	96.8
	TC116	51.6	0.9	15.4	10.6	7.7	0.2	10.6	2.6	0.7	0.1	100.5
	TC47	70.0	0.4	10.9	2.8	0.7	0.2	5.2	0.9	3.4	0.2	94.7
	TC59	70.1	0.5	12.6	4.0	0.8	0.1	1.5	0.9	3.2	0.1	93.9
	TC29	82.1	0.3	9.7	1.1	0.4	0.0	1.1	0.9	3.9	0.0	99.5
	TC16	55.9	0.8	15.8	6.4	1.4	0.1	3.0	0.9	3.0	0.3	87.6
	TC28	35.3	0.6	15.9	3.8	1.2	0.0	0.2	0.4	1.8	0.0	59.2
	TC110 <sup>b</sup>	7.4	0.1	4.1	0.5	0.1	0.0	0.2	0.2	0.1	0.0	12.7
	TC40	84.1	0.3	10.2	1.3	0.5	0.0	0.2	1.1	3.9	0.0	101.7
	TC15	60.3	0.3	7.8	10.2	1.7	0.2	4.8	0.6	3.0	0.1	88.8
	TC86	61.5	0.7	13.5	6.5	1.6	0.1	2.8	0.8	2.8	0.1	90.3
	TC92	20.4	0.5	7.9	2.0	0.4	0.0	0.1	0.2	0.7	0.0	32.3
	TC21	84.4	0.4	8.4	1.7	0.0	0.0	0.1	0.0	3.2	0.0	98.1
	TC118 <sup>b</sup>	17.0	0.4	6.6	1.4	0.2	0.0	0.7	0.2	0.4	0.1	26.9

Sample name		SiO2	TiO2	AI2O3	Fe2O3	MgO	MnO	CaO	Na2O	к20	P2O5	TOTAL
T139228	ТС70	72.5	0.4	14.2	3.7	0.8	0.0	0.8	1.4	4.0	0.0	97.9
	TC44	51.5	1.4	13.2	12.6	6.5	0.2	8.6	2.5	0.9	0.2	97.5
	TC36	66.6	0.7	14.2	6.0	1.7	0.1	2.6	1.1	3.2	0.1	96.1
	TC63	44.9	2.4	14.8	14.3	3.2	0.3	9.9	2.8	0.2	0.2	93.1
	TC30	74.0	0.6	11.9	2.8	0.6	0.0	1.7	0.8	3.4	0.2	96.0
	TC39	41.2	0.7	17.1	3.7	1.2	0.0	0.3	0.5	2.0	0.0	66.6
	TC76	67.7	0.7	15.1	4.5	0.8	0.0	0.4	0.8	3.3	0.0	93.4
	тс96	50.3	0.7	9.5	15.4	1.6	0.1	1.1	0.2	2.2	0.0	80.9
	TC78	72.5	0.5	11.9	3.4	0.1	0.0	0.1	0.0	2.5	0.0	90.9
	TC56	72.3	0.8	17.7	1.0	0.1	0.0	0.1	0.0	2.3	0.0	93.5
	TC60	62.8	1.0	17.9	2.9	0.9	0.0	0.3	0.4	3.3	0.0	89.5
	1000	02.0	1.0	17.5	2.5	0.9	0.0	0.5	0.4	5.5	0.1	69.5
W569001	ТС69	49.1	1.1	14.2	13.8	8.0	0.2	11.5	2.7	0.3	0.2	101.1
	TC80	42.2	0.6	14.5	20.2	2.1	0.1	1.0	0.9	1.6	0.1	83.3
	TC54	66.4	0.7	13.4	4.1	0.8	0.0	1.0	1.0	3.1	0.1	90.7
	TC111 <sup>b</sup>	9.3	0.2	4.1	0.3	0.5	0.0	2.0	0.3	0.1	0.2	17.2
	TC89	56.9	0.9	27.7	1.1	0.1	0.0	0.2	0.0	2.5	0.1	89.5
	TC102 <sup>b</sup>	11.0	0.3	6.3	1.3	0.4	0.0	1.2	0.3	0.2	0.0	20.9
	TC22	62.3	0.9	17.0	3.4	1.0	0.0	1.2	0.5	3.2	0.1	89.6
	TC103 <sup>b</sup>	19.2	0.3	7.5	7.5	0.3	0.0	0.4	0.3	0.8	0.0	36.4
	10100	10.2	0.0	1.0	1.0	0.0	0.0	011	0.0	0.0	0.0	
Y106148	TC32	75.5	0.8	15.6	2.4	0.4	0.0	0.3	0.1	3.9	0.0	99.0
	TC12	74.1	0.5	11.4	3.1	0.4	0.0	0.9	1.4	3.8	0.0	95.7
	TC8	56.3	0.8	16.8	6.2	1.3	0.1	1.2	0.9	2.8	0.1	86.5
	TC95	60.9	0.9	18.7	4.3	1.0	0.1	0.6	1.0	3.4	0.1	90.9
	TC19	93.7	0.4	4.2	1.2	0.1	0.0	0.0	0.0	2.2	0.0	101.7
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Z145168	TC31	79.2	0.5	12.1	2.5	0.7	0.0	0.5	1.2	3.0	0.1	99.8
	TC115	52.8	1.5	13.5	12.7	5.1	0.2	8.5	2.7	1.0	0.2	98.2
	TC100	58.9	0.7	15.2	5.8	1.8	0.1	4.6	2.0	3.3	0.1	92.5
	TC37	47.2	2.0	14.5	13.9	3.3	0.2	8.1	2.3	0.8	0.2	92.6
	TC68	73.6	0.6	14.2	3.9	1.3	0.0	0.6	1.0	3.4	0.1	98.6
	TC85	64.8	0.7	13.9	7.3	1.4	0.1	0.8	0.9	3.1	0.1	93.0
	TC67	60.9	0.7	15.5	5.4	1.1	0.1	0.6	1.6	3.2	0.1	89.2
	TC42	64.5	0.8	22.9	1.1	0.0	0.0	0.1	0.0	2.3	0.0	91.6
Z145198	TC71	51.6	1.3	13.6	12.3	6.8	0.2	9.3	2.3	0.8	0.2	98.4
	ТС90	66.1	0.8	15.2	6.0	1.1	0.1	1.2	0.8	3.2	0.1	94.6
	TC26	65.6	0.7	12.8	3.6	1.4	0.1	4.1	1.6	3.3	0.2	93.5
	TC2	69.4	0.6	17.7	0.6	0.1	0.0	0.1	0.0	2.6	0.0	91.1
	TC72	63.9	1.0	17.4	2.3	0.5	0.0	0.1	0.1	3.1	0.1	88.5
	T88	85.4	0.5	10.1	1.0	0.1	0.0	0.1	0.0	3.5	0.0	100.7

Sample name		SiO2	TiO2	AI2O3	Fe2O3	MgO	MnO	CaO	Na2O	к20	P2O5	TOTAL
Z145199	TC46	51.0	1.0	14.8	10.4	7.6	0.2	10.4	1.9	0.6	0.1	98.0
	TC17	72.3	0.6	12.4	5.8	1.1	0.1	0.5	1.2	3.2	0.1	97.2
	TC50	52.4	1.6	13.0	13.5	5.4	0.2	8.5	2.7	0.9	0.2	98.4
	TC77	60.9	0.6	12.8	8.0	1.1	0.1	3.5	0.7	2.7	0.2	90.6
	TC81	28.0	0.5	11.5	3.7	0.8	0.0	0.4	0.2	1.4	0.0	46.5
	TC20	56.9	0.6	13.2	5.2	1.9	0.1	5.6	1.0	3.0	0.1	87.6
	TC64	88.0	0.4	8.2	0.9	0.0	0.0	0.1	0.0	3.1	0.0	100.7
	Average	57.5	0.7	12.9	5.2	1.4	0.1	2.2	0.9	2.4	0.1	83.3

# Appendix 6: Mine 2 XRF Mineral Analysis

Sample name	Lab no.	AI2O3	CaO	Fe2O3	к20	MgO	MnO	P2O5	SiO2	TiO2	Na2O	TOTAL
R 100 001	SF 11	14.7	10.4	11.0	0.5	7.2	0.1	0.1	51.7	0.9	2.1	98.6
	SF 12	12.2	2.7	3.6	3.2	0.9	0.0	0.0	71.2	0.5	1.6	95.9
	SF 13	18.2	0.4	4.0	1.9	1.6	0.0	0.0	46.6	0.7	0.9	74.3
	SF 61 <sup>B</sup>	5.65	0.67	1.27	0.40	0.69	0.01	0.02	13.26	0.15	0.93	23.1
	SF 14	15.4	3.1	4.3	3.1	1.0	0.0	0.1	61.6	0.6	0.9	90.2
	SF 1	16.4	1.3	11.3	2.2	1.7	0.0	0.2	46.9	0.8	0.4	81.2
	SF 62 <sup>B</sup>	4.71	2.17	0.36	0.16	0.64	0.01	0.14	11.79	0.20	0.41	20.6
	SF 2	19.4	0.1	0.8	2.1	0.2	0.0	0.0	70.0	1.1	1.0	94.7
	SF 15	16.1	0.2	2.3	3.2	0.5	0.0	0.1	67.4	0.6	0.7	91.3
	SF 16	12.3	0.1	1.0	3.3	0.3	0.0	0.0	79.7	0.4	0.5	97.6
	SF 17	8.5	0.1	0.7	1.9	0.1	0.0	0.0	83.1	0.6	0.7	95.8
	SF 18	12.9	0.4	10.5	2.3	0.9	0.1	0.1	59.6	1.7	0.5	89.1
	SF 19	8.9	4.1	7.3	3.2	1.8	0.1	0.1	63.3	0.3	2.3	91.4
			-		-	-				-		
R 100 002	SF 20	14.4	6.6	8.1	1.7	3.9	0.1	0.1	57.8	0.8	1.2	94.7
	SF 21	14.0	2.4	7.3	2.8	1.2	0.1	0.1	61.3	0.7	0.9	90.6
	SF 3	15.4	0.8	9.5	2.5	1.5	0.0	0.1	54.7	0.7	0.4	85.4
	SF 63 <sup>B</sup>	2.68	1.23	0.21	0.08	0.37	0.00	0.08	6.81	0.11	0.58	12.2
	SF 4	10.4	0.3	1.6	2.5	0.2	0.0	0.0	67.3	0.9	1.2	84.5
	SF 22	17.9	0.5	2.7	2.8	0.6	0.0	0.2	59.8	0.7	0.6	85.8
	SF 23	11.2	0.1	0.9	2.9	0.2	0.0	0.0	80.1	0.4	0.5	96.3
	SF 24	10.2	0.3	4.6	3.0	0.4	0.0	0.0	75.9	0.7	1.9	97.0
	T	•	1		1	1		1		1		Γ
Z124027	SF 45	10.2	8.0	13.6	0.8	5.5	0.1	0.2	52.1	1.4	2.4	94.4
	SF 46	8.4	12.4	5.0	2.5	0.9	0.1	1.5	54.8	0.3	0.9	86.7
	SF 47	10.9	0.4	3.2	3.5	0.8	0.0	0.1	71.3	0.6	1.7	92.6
	SF 48	12.4	1.8	5.0	3.0	1.1	0.0	0.1	64.2	0.6	1.0	89.1
	SF 49	13.7	0.4	4.2	1.8	1.3	0.0	0.0	41.4	0.5	0.6	64.0
	SF 64 <sup>B</sup>	3.52	5.33	5.24	0.20	1.22	0.03	0.00	20.36	0.10	1.26	37.3
	SF 50	12.2	8.6	15.6	0.4	3.8	0.1	0.2	47.1	2.0	2.1	92.1
	SF 51	12.4	0.7	5.7	3.1	1.0	0.0	0.1	64.2	0.5	1.0	88.6
	SF 65 <sup>B</sup>	7.88	0.44	5.28	0.65	0.23	0.01	0.01	22.32	0.24	0.67	37.7
	SF 52	11.6	0.2	1.4	1.5	0.1	0.0	0.0	77.2	0.6	1.0	93.7
	SF 7	3.0	0.3	4.1	0.9	0.0	0.0	0.0	89.3	0.6	0.7	98.9
	SF 66 <sup>B</sup>	5.08	1.56	0.13	0.09	0.58	0.01	0.03	10.52	0.16	0.48	18.6
	SF 8	15.5	0.1	0.7	3.4	0.1	0.0	0.0	73.1	0.5	0.7	94.2
	SF 53	13.2	1.5	2.6	3.1	0.9	0.0	0.0	66.5	0.7	0.9	89.5
	SF 54	12.6	6.6	13.6	0.5	3.8	0.1	0.1	49.2	1.8	1.7	90.0
	SF 55	4.5	0.8	3.9	1.0	0.3	0.0	0.0	83.2	0.5	0.9	95.2

Sample name	Lab no.	AI2O3	CaO	Fe2O3	к20	MgO	MnO	P2O5	SiO2	TiO2	Na2O	TOTAL
Z124027	SF 56	9.3	0.8	1.5	3.7	0.5	0.0	0.0	80.0	0.2	0.8	96.8
	SF 57	12.2	9.8	16.7	0.4	4.9	0.1	0.2	48.9	2.0	2.6	97.7
	SF 58	11.0	0.4	0.9	3.0	0.3	0.0	0.0	76.7	0.4	0.5	93.1
	SF 59	14.0	0.3	5.2	2.2	1.2	0.0	0.2	69.3	0.7	0.3	93.4
	SF 60	19.2	0.6	11.0	2.9	1.5	0.0	0.2	57.3	0.7	0.4	93.8
Z124029	SF 25	10.4	0.3	1.3	4.0	0.4	0.0	0.0	80.4	0.1	2.5	99.6
	SF 26	15.1	10.8	10.8	0.5	7.4	0.1	0.1	54.1	0.8	2.5	102.2
	SF 27	14.0	1.6	4.3	3.1	1.0	0.0	0.1	69.4	0.5	1.3	95.3
	SF 28	16.5	0.6	3.4	1.6	1.3	0.0	0.0	43.7	0.6	1.0	68.9
	SF 29	15.8	2.7	6.4	3.0	1.4	0.0	0.5	56.4	0.7	0.9	87.8
	SF 5	12.2	1.3	9.6	2.3	1.3	0.0	0.1	52.7	0.6	1.0	81.0
	SF 67 <sup>B</sup>	6.81	1.55	0.33	0.18	0.55	0.01	0.16	15.27	0.31	0.51	25.7
	SF 6	2.9	2.2	1.2	0.5	0.1	0.0	0.0	90.0	0.2	0.1	97.0
	SF 30	14.0	0.3	2.1	3.2	0.6	0.0	0.0	70.6	0.6	0.8	92.2
	SF 31	13.9	0.2	1.4	1.6	0.2	0.0	0.0	71.4	0.9	1.2	90.8
	SF 32	14.8	0.2	4.4	2.2	0.4	0.0	0.0	68.0	0.8	0.5	91.3
	SF 33	10.2	0.4	5.3	2.1	0.8	0.0	0.0	75.2	1.5	0.8	96.4
	SF 34	15.4	4.5	15.5	1.0	2.3	0.1	0.2	44.8	2.7	1.4	87.9
Z124030	1	-	-		-	-	-	1	[	[	[	
2124030	SF 35	10.2	0.3	1.3	4.0	0.4	0.0	0.0	81.1	0.2	2.0	99.6
	SF 36	9.0	11.8	14.8	0.5	4.3	0.3	0.1	45.7	1.3	0.6	88.4
	SF 37	13.2	0.3	0.8	2.0	0.3	0.0	0.0	75.9	0.5	0.8	93.8
	SF 38	15.6	10.6	10.3	0.5	6.4	0.1	0.1	51.4	0.9	2.3	98.1
	SF 39	13.7	1.1	5.2	3.4	1.2	0.0	0.1	68.3	0.5	1.3	94.9
	SF 68 <sup>B</sup>	1.15	0.69	7.61	0.04	0.27	0.00	0.00	3.80	0.02	0.41	14.0
	SF 40	12.9	0.5	5.4	1.4	1.1	0.0	0.0	32.1	0.4	0.4	54.3
	SF 69 <sup>B</sup>	9.78	0.78	0.92	0.77	0.92	0.01	0.01	23.51	0.23	0.66	37.6
	SF 41	15.3	0.6	5.4	3.1	1.1	0.0	0.1	65.4	0.6	0.7	92.3
	SF 70 <sup>B</sup>	9.66	0.24	2.36	0.44	0.23	0.01	0.02	27.12	0.37	1.08	41.5
	SF 9	3.7	0.0	1.5	1.1	0.1	0.0	0.0	88.8	0.3	0.9	96.4
	SF 71 <sup>B</sup>	3.06	1.69	0.23	0.05	0.65	0.01	0.09	6.64	0.13	0.85	13.4
	SF 10	13.6	0.1	0.7	2.6	0.1	0.0	0.0	75.3	0.5	0.6	93.5
	SF 42	17.8	0.6	4.9	3.2	1.3	0.0	0.1	60.9	0.8	0.5	90.1
	SF 43	15.8	3.7	5.7	2.6	1.7	0.0	0.2	61.0	0.6	1.0	92.5
	SF 44	15.3	5.0	9.1	0.9	4.0	0.1	0.1	53.9	0.6	0.6	89.6
	Average	11.6	2.3	5.1	1.9	1.4	0.0	0.1	56.6	0.7	1.0	80.8

Sample						
Name F142441	Lab no.	Depth (m)	Geology	Initial pH	Final pH	Interpretation
F 14244 I	TC18	1.7-34.48	DO	10.2	7.4	Lower Acid Risk
	TC75	34.48-49.07	SST	9.13	6.11	Lower Acid Risk
	TC38	49.07-70.25	SST,C	9.29	7.02	Lower Acid Risk
	TC83	70.6-82.75	SST, SLT, SH	8.09	6.32	Lower Acid Risk
	TC7	83.45-121.47	SST, SLT, SH	8.43	7.81	Lower Acid Risk
	TC35	125.77-128.56	SST, SLT	7.89	3.28	Higher Risk Acid Generation
	1		1	1	1	
F142446	TC66	5.8-9	DO, SST	8.46	4.94	Medium Risk Acid Generation
	TC57	9.15-34.95	SST, SLT	9.34	6.78	Lower Acid Risk
	TC 62	34.95-70.5	DO	9.99	7.2	Lower Acid Risk
	TC10	70.5-82.44	SST, SLT	9.79	7.11	Lower Acid Risk
	TC4	82.44-91.10	SST	9.7	6.11	Lower Acid Risk
	TC51	91.10-112.10	SST, SLT	8.78	3.34	Higher Risk Acid Generation
	TC41	112.10-113	C, SLT, CSH	8.84	1.63	Higher Risk Acid Generation
	TC33	113.12-155.3	SST, SLT, C	8.3	7.04	Lower Acid Risk
	TC87	155.3-161.50	SST, SLT, C	9.2	5.64	Lower Acid Risk
F142471	TC65	4.68-12.9	SLT, SST, KV	8.3	6.69	Lower Acid Risk
	TC55	12.9-18.82	SST	8.64	6.19	Lower Acid Risk
	TC61	18.82-40.85	SST,SLT, KV	8.29	6.64	Lower Acid Risk
	ТС3	42.38-75.68	SST, SLT, SH, KV	7.7	3.3	Higher Risk Acid Generation
	TC107 <sup>b</sup>	76.09-80.80	CH4	7.67	4.22	Medium Risk Acid Generation
	TC11	83.8-84.15	SLT	7.98	3.07	Higher Risk Acid Generation
	TC112 <sup>b</sup>	84.15-84.80	C3	7.56	1.6	Higher Risk Acid Generation
	TC84	84.8-109.4	SST, SLT, KV, DRT	8.08	6.38	Lower Acid Risk
	TC105 <sup>b</sup>	109.4-109.90	C2	7.98	3.15	Higher Risk Acid Generation
	TC114 <sup>b</sup>	No depth	C5	8.45	1.65	Higher Risk Acid Generation
	•				I	
O105016	TC25	7.3-20.26	SST, QZT, KV	8.92	7.38	Lower Acid Risk
	TC113	20.26-33.5	SST, SLT	8.15	6.76	Lower Acid Risk
	TC48	33.5-41.65	SST, KV	8.98	7.14	Lower Acid Risk
	TC43	41.65-43.55	SST,SLT,SH	8.4	4.68	Medium Risk Acid Generation
	TC101	43.55-45.16	C5H & L	8.7	4.03	Medium Risk Acid Generation
	TC34	45.16-52.82	SST,SLT, CSH	7.57	2.92	Higher Risk Acid Generation
	TC24	52.82-58.99	DO, KV	9.64	7.89	Lower Acid Risk
	TC14	58.99-77.95	SST, SLT, KV, C	8.95	7.35	Lower Acid Risk
	TC79	80.3-85.56	SST, SLT, KV, GRT	7.32	2.31	Higher Risk Acid Generation

# Appendix 7: Interpretation of ABA pH results (Mine 1)

Sample	l ah ra	Donth (m)	Coology		Final all	Interpretation
Name P110030	Lab no.	Depth (m)		Initial pH	Final pH	
	TC49	3.8-16.23	MSB, KV, (OVB)	8.63	6.84	Lower Acid Risk
	TC52 <sup>a</sup>	16.23-20.7	DO, KV	8.77	7.03	Lower Acid Risk
	TC99 <sup>a</sup>	16.23-20.7	DO, KV	8.97	8.12	Lower Acid Risk
	TC1	20.7-30.23	SST, SSL	8.37	6.65	Lower Acid Risk
	TC5	30.23-42.86	SST, KV	8.84	6.29	Lower Acid Risk
	TC74	67.54-68.06	SH, C	8.76	4.3	Medium Risk Acid Generation
	TC117 <sup>b</sup>	67.54-69.54	C5H	8.88	1.76	Higher Risk Acid Generation
	TC104	69.54-106.77	SST, SLT, C	7.78	4.02	Medium Risk Acid Generation
	TC109 <sup>b</sup>	106.38-110.22	C4	7.67	2.33	Higher Risk Acid Generation
	TC82	106.77-109.99	C4L	7.69	3.65	Medium Risk Acid Generation
	TC27	109.99-120.22	C, SST, SLT, CSH	7.64	4.03	Medium Risk Acid Generation
	TC108 <sup>b</sup>	112.75-113.33	C3	7.67	1.47	Higher Risk Acid Generation
	TC98	120.22-141.42	SST, SLT, SH, GRT	8.26	3.12	Higher Risk Acid Generation
	TC106 <sup>b</sup>	141.42-141.55	C2	7.61	1.58	Higher Risk Acid Generation
	1	ſ	ſ	1		
P110087	TC9	6-9.93	DO, KV	9.15	8.21	Lower Acid Risk
	TC13	9.93-30.18	DO, SST, SLT, GRT	8.57	5.09	Medium Risk Acid Generation
	TC97	30.18-40.32	SST, SLT	8.99	7.3	Lower Acid Risk
	TC58	40.32-54.77	SST, SLT	8.4	4.48	Medium Risk Acid Generation
	TC23	54.77-71.87	SST, GRT, KV	8.34	3.19	Higher Risk Acid Generation
	TC94	71.87-89.08	SST, SLT	8.45	7.58	Lower Acid Risk
	TC91	90.19-90.45	C5	8.68	2.68	Higher Risk Acid Generation
	TC45	90.45-130.50	SST, SLT, GRT	7.5	3.7	Medium Risk Acid Generation
	TC53	130.50-133.19	SST, SLT	8.4	2.94	Higher Risk Acid Generation
	TC6	133.19-133.97	GRT, C, SST, SLT	8.28	7.71	Lower Acid Risk
	TC73	139.15-140.18	SLT	7.95	2.21	Higher Risk Acid Generation
R146043	TC93	5.33-8.17	KLY, SST	7.8	6.49	Lower Acid Risk
	TC116	8.17-47.07	DO	10.12	6.79	Lower Acid Risk
	TC47	47.07-78.5	SST, SLT	9.61	7.59	Lower Acid Risk
	TC59	78.5-94.33	SST, SLT	8.58	7.17	Lower Acid Risk
	TC29	94.33-101.9	SST	8.71	6.4	Lower Acid Risk
	TC16	101.9-125.08	SST, SLT	8.8	7.3	Lower Acid Risk
	TC28	125.08-125.92	SLT, SH, C	8.8	1.89	Higher Risk Acid Generation
	TC110 <sup>b</sup>	125.65-125.92	C5L	8.41	1.24	Higher Risk Acid Generation
	TC40	125.92-140.78	SST, SLT, C, GRT	8.68	4.94	Medium Risk Acid Generation
	TC15	140.78-141.52	SST	8.54	8.23	Lower Acid Risk
	TC86	143.28-160.68	SST, SLT	8.23	7.53	Lower Acid Risk
	TC92	160.68-160.98	CSH	7.96	1.15	Higher Risk Acid Generation
	TC21	161.15-167.07	SST	6.62	3.54	Medium Risk Acid Generation
	TC118 <sup>b</sup>	167.07-170.34	C4L	7.76	1.24	Higher Risk Acid Generation

Sample Name	Lab no.	Depth (m)	Geology	Initial pH	Final pH	Interpretation
T139228	TC70	2.2-7.35	SST, KV	8.24	5.4	Medium Risk Acid Generation
	TC44	7.35-39.45	DO	10.37	7.34	Lower Acid Risk
	TC36	39.45-42.87	SST, SLT	10.21	7.38	Lower Acid Risk
	TC63	42.87-43.5	DO	9.48	7.6	Lower Acid Risk
	TC30	43.5-82.2	SST,SLT, KV, GRT	9.08	7.06	Lower Acid Risk
	TC39	82.2-83.81	C, CSH, SH	9.43	1.75	Higher Risk Acid Generation
	TC76	83.81-115.08	SST,SLT,KV	7.96	3.36	Higher Risk Acid Generation
	TC96	115.08-117.67	C, SST, SLT, CSH	7.87	3.85	Medium Risk Acid Generation
	TC78	115.45-117.67	C, SST, SLT, CSH	6.12	2.24	Higher Risk Acid Generation
	TC56	121.35-141.39	SST,SLT,KV	7.95	3.23	Higher Risk Acid Generation
	TC60	141.39-145.24	SST, SLT, SH,C	8.52	2.89	Higher Risk Acid Generation
N569001	TC69	2.4-7.4	DO, KV	9.55	6.56	Lower Acid Risk
	TC80	18.7-19.5	SH, CSH	7.44	3.51	Medium Risk Acid Generation
	TC54	19.5-55.81	SST, SLT, KV	8.77	6.74	Lower Acid Risk
	TC111 <sup>b</sup>	55.81-59.57	C4L	7.64	6.01	Lower Acid Risk
	TC89	59.57-62.8	SST, SLT, KV	8	3.8	Medium Risk Acid Generation
	TC102 <sup>b</sup>	62.8-63.03	C3	7.97	1.78	Higher Risk Acid Generation
	TC22	63.03-90.4	SST, SLT, KV, SH, GRT, CSH	9.09	6.73	Lower Acid Risk
	TC103 <sup>b</sup>	90.4-90.8	C2	7.13	1.69	Higher Risk Acid Generation
Y106048	TC32	5.95-6.66	SST, KV	7.5	3.21	Higher Risk Acid Generation
	TC12	6.66-22.61	SST,SLT, GRT	8.18	5.1	Medium Risk Acid Generation
	TC8	22.61-45.06	SST, SLT, GRT	8.28	5.46	Medium Risk Acid Generation
	TC95	50.59-70.93	SST, SLT	7.97	4.53	Medium Risk Acid Generation
	TC19	70.93-75.27	SST, SLT,GRT	8.08	3	Higher Risk Acid Generation
2145168	TC31	1.6-6.1	SST, GKS, KV	8.39	4.81	Medium Risk Acid Generation
	TC115	6.1-49.68	DO	9.81	7.62	Lower Acid Risk
	TC100	49.68-56.74	SST, SLT, KV	9.57	7.46	Lower Acid Risk
	TC37	56.74-57.36	DO	9.78	8.5	Lower Acid Risk
	TC68	57.36-68.81	SST, SLT, MST	9.35	5.78	Lower Acid Risk
	TC85	68.81-94.75	SST, SLT, GRT	8.31	6.48	Lower Acid Risk
	TC67	96.22-136.4	SST, SLT, C, CSH,	0.16	2 77	Madium Diak Asid Constation
	TC67 TC42		DO	8.16 7.85	3.77	Medium Risk Acid Generation
	1042	139.31-142.08	SST, SLT	7.65	2.74	Higher Risk Acid Generation
Z145198	TC71	3.33-45.04	DO	10.12	7.75	Lower Acid Risk
	TC90	45.04-90.02	SST,SLT,SH	8.78	6.83	Lower Acid Risk
	TC90	91.03-127.78	SST,SLT,SH	8.63	7.7	
			SST, SLT, SH	7.7	2.85	Lower Acid Risk Higher Risk Acid Generation
	TC2				200	
	TC2 TC72	132.14-134.94 134.94-151.78	SST, SLT, C	8.03	3.07	Higher Risk Acid Generation

Sample Name	Lab no.	Depth (m)	Geology	Initial pH	Final pH	Interpretation
Z145199	TC46	0-22.52	KLY,KV,DO	9.59	6.98	Lower Acid Risk
	TC17	22.52-28.75	SST, SLT	8.79	6.82	Lower Acid Risk
	TC50	28.75-52.14	DO,KV	10.18	7.64	Lower Acid Risk
	TC77	52.14-123.92	SST,SLT,KV,DO,GRT	8.53	7.76	Lower Acid Risk
	TC81	123.92-124.6	SH	8.77	1.79	Higher Risk Acid Generation
	TC20	125.80-159.6	SST, SLT, C	8.36	7.53	Lower Acid Risk
	TC64	160.3-162.3	C, SST	8.15	3.03	Higher Risk Acid Generation

Sample						
Name	Lab no.	Depth (m)	Geology	Initial pH	Final pH	Interpretation
R100001	05 11	1 00 10 40	DO	0.0	7 75	Lewer Asid Disk
	SF 11	1.62-12.46	SST, SLT, GRT	9.8	7.75	Lower Acid Risk
	SF 12	12.46-24.48		9.81	7.56	Lower Acid Risk
	07.40		C, CSH			
	SF 13	24.48-25.26	C5H, C5M, C5L	9.85	2.37	Higher Risk Acid Generation
	SF61 <sup>b</sup>	24.48-25.80	,,	8.89	5.61	Lower Acid Risk
			SLT, SST			
	SF 14	25.80-52.66	SLT, GRT	8.93	7.3	Lower Acid Risk
	SF 1	50.66-51.6	SET, GIVI	8.04	4.27	Medium Risk Acid Generation
		00.00 01.0	C4L	0.01	1.27	
	SF62 <sup>b</sup>	51.6-56.92		8.22	6.21	Lower Acid Risk
	SF 2	56.92-57.14	SLT	7.73	3.44	Higher Risk Acid Generation
	512	30.92-37.14	SST, SLT	7.73	5.44	Tigher Risk Acid Generation
	SF 15	57.14-85.28		8.6	4.51	Medium Risk Acid Generation
	05.40		SST	7.00	0.04	
	SF 16	86.82-112.94	GRT	7.63	3.64	Medium Risk Acid Generation
	SF 17	112.94-116.23		7.55	3.51	Medium Risk Acid Generation
			SST, GRT, SLT,			
			WSH			
	SF 18	116.23-130.1	TIL	8.14	6.85	Lower Acid Risk
	SF 19	130.1-143.1		9.24	7.83	Lower Acid Risk
R100002			GRD, SST, DO			
	SF 20	14.1-18	SST, SLT	9.16	7.92	Lower Acid Risk
	SF 21	18.3-37.5	001,021	8.7	7.62	Lower Acid Risk
			GRT			
	SF 3	37.5-37.82	C4L	7.86	5.91	Lower Acid Risk
	SF63 <sup>b</sup>	37.82-43.08	C4L	7.95	5.49	Medium Risk Acid Generation
	5-05	37.82-43.08	SST, SLT	7.95	5.49	
	SF 4	43.08-45.72		7.92	1.84	Higher Risk Acid Generation
	05.00	45 70 70 1	SST, SLT	0.1	4.05	Madium Dials A 11 C
	SF 22	45.72-70.1	SST, GRT	8.4	4.25	Medium Risk Acid Generation
	SF 23	71.4-103.8		7.77	3.72	Medium Risk Acid Generation
			GRT, TIL, KV			
	SF 24	103.8-124.4		8.25	6.57	Lower Acid Risk

# Appendix 8: Interpretation of ABA pH results (Mine 2)

Sample						
Name	Lab no.	Depth (m)	Geology	Initial pH	Final pH	Interpretation
Z124027			DO, KV,			
	SF 45	4.13-14.04	SST, SLT	9.27	8.43	Lower Acid Risk
	SF 46	14.04-48.99	001, 021	9.54	8.1	Lower Acid Risk
			SST, DO	0.01	0.1	
	SF 47	48.99-69.26		8.9	5.86	Lower Acid Risk
			SST, SLT			
	SF 48	69.26-90.34	C, CSH	9.05	7.22	Lower Acid Risk
	SF 49	90.44-91.84	0,001	9.34	2.41	Higher Risk Acid Generation
	51 45	30.44-31.04	C5L	3.54	2.41	Higher Misk Acid Generation
	SF64 <sup>b</sup>	91.84-92.34		8.94	7.74	Lower Acid Risk
			DO			
	SF 50	92.34-93.61		9.89	8.22	Lower Acid Risk
	SF 51	02 61 122 04	SST, SLT	8.89	7.81	Lower Acid Pick
	35 31	93.61-123.04	C4H	0.09	7.01	Lower Acid Risk
	SF65 <sup>b</sup>	123.08-123.68		8.31	2.29	Higher Risk Acid Generation
			SH, SST, GRT, SLT			
	SF 52	123.68-125.28	0.07	8.07	3.21	Higher Risk Acid Generation
	05.7		GRT	F F7	0.00	
	SF 7	125.28-125.88	C4L	5.57	2.33	Higher Risk Acid Generation
	SF66 <sup>b</sup>	125.88-129.68		8.18	5.51	Lower Acid Risk
			SLT, SST, C			
	SF 8	129.78-134.51		7.37	2.87	Higher Risk Acid Generation
	0 0		SLT, SST, C, DO			
	SF 53	134.48-162.76	DO, SST, C	9.21	6.67	Lower Acid Risk
	SF 54	162.54-164.8	20,001,0	10.33	8.5	Lower Acid Risk
			GRT, SLT		0.0	
	SF 55	167.2-172.87		8.72	2.52	Higher Risk Acid Generation
			SST, DO			
	SF 56	172.87-180.67	DO	8.92	6.83	Lower Acid Risk
	SF 57	180.67-183.7	20	9.91	8.25	Lower Acid Risk
	0.07		SST, SLT	0.01	0.20	
	SF 58	183.7-186.86		8.88	5.88	Lower Acid Risk
			TIO, SST			
	SF 59	186.86-189.02	QZT, EOH	8.32	4.37	Medium Risk Acid Generation
	SF 60	189.02-200.16		8.65	4.32	Medium Risk Acid Generation
		100.02-200.10	1	0.00	7.02	modulin Hisk / Icia Generation
Z124029			KV, SST			
	SF 25	6.9-8.1		8.73	5.65	Lower Acid Risk
			DO			
	SF 26	8.1-61.46		10.41	7.05	Lower Acid Risk
	05.07	61 46 70 60	SST, SLT, GRT	10.00	6.40	Lower Asid Disk
	SF 27	61.46-73.68	C, CSH	10.38	6.46	Lower Acid Risk
	SF 28	73.68-74.72	-,	10.09	5.91	Lower Acid Risk
			SST, SLT		2.01	
	SF 29	74.72-98.23		9.9	6.89	Lower Acid Risk

Sample						
Name Z124029	Lab no.	Depth (m)	Geology SST, SLT	Initial pH	Final pH	Interpretation
Z124029		00 00 00 00	551, 5L1	0.0	2.20	Llinker Diel: Asid Consertion
	SF 5	98.23-98.88	C4L	8.2	3.36	Higher Risk Acid Generation
	SF67 <sup>b</sup>	98.88-102.83		8.19	5.41	Medium Risk Acid Generation
	0107	00.00 102.00	GRT	0.10	0.11	
	SF 6	102.83-104.6		8.43	6.75	Lower Acid Risk
			SST, SLT, GRT			
	SF 30	104.6-132.74		8.54	3.45	Higher Risk Acid Generation
			SST, SLT, GRT			
	SF 31	133.63-136.28	SST, SLT, GRT	8.57	2.59	Higher Risk Acid Generation
	05.00	100 04 150 45	551, 5L1, GR1	0 5 0	2.07	Llinker Dick Acid Concretion
	SF 32	138.64-158.45	SST, SLT, TIL	8.58	2.97	Higher Risk Acid Generation
	SF 33	158.45-165.43		8.55	5.65	Lower Acid Risk
	01 00	100.40-100.40	GR, EOH	0.00	0.00	Lower Acid Hisk
	SF 34	165.43-174.53		10.26	7.9	Lower Acid Risk
Z124030			SST			
	SF 35	11.6-12.52		9.15	5.29	Medium Risk Acid Generation
			DO, KV			
	SF 36	12.52-20.8		9.11	8.84	Lower Acid Risk
			SST, SLT			
	SF 37	20.8-41.35		8.62	6.43	Lower Acid Risk
			DO			
	SF 38	41.35-49		10.21	6.42	Lower Acid Risk
			SST, SLT			
	SF 39	49-98.3		9.31	8.3	Lower Acid Risk
			C5H			
	SF68 <sup>b</sup>	99.20-99.42		8.1	2	Higher Risk Acid Generation
			C, SLT			
	SF 40	98.3-100.45		9.63	2.34	Higher Risk Acid Generation
	h		C5L			
	SF69 <sup>b</sup>	100.45-100.80		9.25	2.51	Higher Risk Acid Generation
	0 - 11		SST, SLT			
	SF 41	100.80-133.37	C4H	8.91	4.4	Medium Risk Acid Generation
	orzob	100.07.100.00		0.70	1.07	
	SF70 <sup>b</sup>	133.37-133.92	GRT	8.73	1.97	Higher Risk Acid Generation
	05.0	100 00 105 50	GIT	0.0	0.01	Llink on Disk Asid Osmanation
	SF 9	133.92-135.56	C4L	6.6	2.81	Higher Risk Acid Generation
	SF71 <sup>♭</sup>	135.56-139.83		8.17	6.82	Lower Acid Risk
	3F71	133.30-133.63	SST	0.17	0.02	
	SF 10	139.83-140.6		6.96	3.34	Higher Risk Acid Generation
		100.00-1-0.0	C, SST, SLT, KV	0.00	0.04	
	SF 42	140.60-164.48	. , , .	8.91	5.84	Lower Acid Risk
	<u> </u>		SLT	0.01	0.01	
	SF 43	167.1-168.2		10.11	6.66	Lower Acid Risk
			SST, TIO, LAV			
	SF 44	168.2-185.66		10.25	7.09	Lower Acid Risk

## Appendix 9: Interpretation of ABA Net Neutralising Potential Results (Mine 1)

Side name	Lab No.	Depth	Thickness (m)	NNP	NNP x Thickness	Geology	NNP - Interpretation
F142441	TC18	1.7-34.48	32.78	9.60	314.67	DO	Verify with other tests
	TC75	34.48-49.07	14.59	21.09	307.75	SST	Probably Excess Neutralising minerals
	TC38	49.07-70.25	21.18	15.13	320.54	SST,C	Verify with other tests
	TC83	70.6-82.75	12.15	22.74	276.31	SST, SLT, SH	Probably Excess Neutralising minerals
	TC7	83.45-121.47	38.02	6.33	240.48	SST, SLT, SH	Verify with other tests
	TC35	125.77-128.56	2.79	-2.99	-8.35	SST, SLT	Verify with other tests
F142446	TC66	5.8-9	3.2	6.54	20.92	DO, SST	Verify with other tests
	TC57	9.15-34.95	25.8	32.77	845.57	SST, SLT	Probably Excess Neutralising minerals
	TC 62	34.95-70.5	35.55	16.65	591.93	DO	Verify with other tests
	TC10	70.5-82.44	11.94	18.17	216.98	SST, SLT	Verify with other tests
	TC4	82.44-91.10	8.66	8.80	76.20	SST	Verify with other tests
	TC51	91.10-112.10	21	-55.95	-1174.97	SST, SLT	Potential Acid Generator
	TC41	112.10-113	0.9	-17.13	-15.42	C, SLT, CSH	Verify with other tests
	TC33	113.12-155.3	42.18	12.04	507.67	SST, SLT, C	Verify with other tests
	TC87	155.3-161.50	6.2	10.86	67.30	SST, SLT, C	Verify with other tests
F142471	TC65	4.68-12.9	8.22	18.16	149.29	SLT, SST, KV	Verify with other tests
	TC55	12.9-18.82	5.92	8.78	52.00	SST	Verify with other tests
	TC61	18.82-40.85	22.03	14.95	329.38	SST,SLT, KV	Verify with other tests
	ТС3	42.38-75.68	33.3	8.91	296.69	SST, SLT, SH, KV	Verify with other tests
	TC107 <sup>b</sup>	76.09-80.80	4.71	18.50	87.11	C4H	Verify with other tests
	TC11	83.8-84.15	0.35	-23.69	-8.29	SLT	Potential Acid Generator
	TC112 <sup>b</sup>	84.15-84.80	0.65	-169.96	-110.48	C3	Potential Acid Generator
	TC84	84.8-109.4	24.6	51.16	1258.45	SST, SLT, KV, DRT	Probably Excess Neutralising minerals
	TC105 <sup>ь</sup>	109.4-109.90	0.5	21.70	10.85	C2	Probably Excess Neutralising minerals
	TC114 <sup>b</sup>	40.85-42.38	1.53	3.37	5.16	C5	Verify with other tests
O105016	TC25	7.3-20.26	12.96	87.72	1136.89	SST, QZT, KV	Probably Excess Neutralising minerals
	TC113	20.26-33.5	13.24	25.50	337.57	SST, SLT	Probably Excess Neutralising minerals
	TC48	33.5-41.65	8.15	35.39	288.46	SST, KV	Probably Excess Neutralising minerals
	TC43	41.65-43.55	1.9	9.82	18.65	SST,SLT,SH	Verify with other tests
	TC101	43.55-45.16	1.61	3.25	5.22	C5H & L	Verify with other tests
	TC34	45.16-52.82	7.66	-34.98	-267.92	SST,SLT, CSH	Potential Acid Generator
							Probably Excess Neutralising
	TC24	52.82-58.99	6.17	24.76 24.75	152.76 469.27	DO, KV SST, SLT, KV, C	minerals Probably Excess Neutralising minerals
	TC14	58.99-77.95	18.96				

Side name	Lab No.	Depth (m)	Thickness (m)	NNP	NNP x Thickness	Geology	NNP - Interpretation
P110030	TC49	3.8-16.23	12.43	14.45	179.60	MSB, KV, (OVB)	Verify with other tests
	T52	16.23-20.7	4.47	26.91	120.28	DO, KV	Probably Excess Neutralising minerals Probably Excess Neutralising
	Т99	16.23-20.7	4.47	26.17	117.00	DO, KV	minerals
	TC1	20.7-30.23	9.53	1.01	9.64	SST, SSL	Verify with other tests
	TC5	30.23-42.86	12.63	22.24	280.93	SST, KV	Probably Excess Neutralising minerals
	TC74	67.54-68.06	0.52	5.02	2.61	SH, C	Verify with other tests
	TC117 <sup>b</sup>	67.54-69.54	2	2.97	5.95	C5H	Verify with other tests
	TC104	69.54-106.77	37.23	0.47	17.40	SST, SLT, C	Verify with other tests
	TC109 <sup>b</sup>	106.38-110.22	3.84	7.68	29.51	C4	Verify with other tests
	TC82	106.77-109.99	3.22	2.84	9.16	C4L	Verify with other tests
	TC27	109.99-120.22	10.23	-0.73	-7.47	C, SST, SLT, CSH	Verify with other tests
	TC108 <sup>b</sup>	112.75-113.33	0.58	-130.70	-75.81	C3	Potential Acid Generator
	TC98	120.22-141.42	21.2	0.57	12.02	SST, SLT, SH, GRT	Verify with other tests
	TC106 <sup>ь</sup>	141.42-141.55	0.13	-377.21	-49.04	C2	Potential Acid Generator
P110087	TC9	6-9.93	3.93	151.52	595.47	DO, KV	Probably Excess Neutralising minerals
	TC13	9.93-30.18	20.25	5.81	117.63	DO, SST, SLT, GRT	Verify with other tests
	TC97	30.18-40.32	10.14	220.34	2234.30	SST, SLT	Probably Excess Neutralising minerals
	TC58	40.32-54.77	14.45	-11.61	-167.79	SST, SLT	Verify with other tests
	TC23	54.77-71.87	17.1	-1.56	-26.73	SST, GRT, KV	Verify with other tests
	TC94	71.87-89.08	17.21	85.11	1464.70	SST, SLT	Probably Excess Neutralising minerals
	TC91	90.19-90.45	0.26	-88.51	-23.01	C5	Potential Acid Generator
	TC45	90.45-130.50	40.05	-17.91	-717.36	SST, SLT, GRT	Verify with other tests
	TC53	130.50-133.19	2.69	-93.38	-251.20	SST, SLT	Potential Acid Generator
	TC6	133.19-133.97	0.78	6.06	4.73	GRT, C, SST, SLT	Verify with other tests
	TC73	139.15-140.18	1.03	-62.80	-64.68	SLT	Potential Acid Generator
R146043	TC93	5.33-8.17	2.84	15.54	44.14	KLY, SST	Verify with other tests
	TC116	8.17-47.07	38.9	26.13	1016.63	DO	Probably Excess Neutralising minerals
	TC47	47.07-78.5	31.43	102.78	3230.38	SST, SLT	Probably Excess Neutralising minerals
	TC59	78.5-94.33	15.83	30.14	477.05	SST, SLT	Probably Excess Neutralising minerals
	TC29	94.33-101.9	7.57	19.24	145.63	SST	Verify with other tests
	TC16	101.9-125.08	23.18	58.44	1354.74	SST, SLT	Probably Excess Neutralising minerals
	TC28	125.08-125.92	0.84	-51.68	-43.41	SLT, SH, C	Potential Acid Generator
	TC110 <sup>b</sup>	125.65-125.92	0.27	-13.44	-3.63	C5L	Verify with other tests
	TC40	125.92-140.78	14.86	0.50	7.40	SST, SLT, C, GRT	Verify with other tests
	TC15	140.78-141.52	0.74	34.99	25.89	SST	Probably Excess Neutralising minerals
	TC86	143.28-160.68	17.4	69.46	1208.69	SST, SLT	Probably Excess Neutralising minerals

Side name	Lab No.	Depth (m)	Thickness (m)	NNP	NNP x Thickness	Geology	NNP - Interpretation
	TC92	160.68-160.98	0.3	-86.60	-25.98	CSH	Potential Acid Generator
	TC21	161.15-167.07	5.92	-17.23	-102.01	SST	Verify with other tests
	TC118 <sup>b</sup>	167.07-170.34	3.27	-46.60	-152.38	C4L	Potential Acid Generator
T139228	TC70	2.2-7.35	5.15	8.98	46.26	SST, KV	Verify with other tests
	TC44	7.35-39.45	32.1	17.55	563.24	DO	Verify with other tests
	TC36	39.45-42.87	3.42	49.48	169.24	SST, SLT	Probably Excess Neutralising minerals
	TC63	42.87-43.5	0.63	72.18	45.47	DO	Probably Excess Neutralising minerals
	TC30	43.5-82.2	38.7	28.83	1115.68	SST,SLT, KV, GRT	Probably Excess Neutralising minerals
	TC39	82.2-83.81	1.61	-9.49	-15.28	C, CSH, SH	Verify with other tests
	TC76	83.81-115.08	31.27	-32.06	-1002.54	SST,SLT,KV	Potential Acid Generator
	TC96	115.08-117.67	2.59	-27.48	-71.16	C, SST, SLT, CSH	Potential Acid Generator
	TC78	115.45-117.67	2.22	-148.69	-330.09	C, SST, SLT, CSH	Potential Acid Generator
	TC56	121.35-141.39	20.04	-7.60	-152.30	SST,SLT,KV	Verify with other tests
	TC60	141.39-145.24	3.85	-2.33	-8.97	SST, SLT, SH,C	Verify with other tests
			_	<u> </u>	400 70		Probably Excess Neutralising
W569001	TC69	2.4-7.4	5	33.94	169.70	DO, KV	minerals
	TC80	18.7-19.5	0.8	-29.21	-23.37	SH, CSH	Potential Acid Generator
	TC54 TC111 <sup>b</sup>	19.5-55.81 55.81-59.57	36.31 3.76	12.34 47.31	447.99 177.89	SST, SLT, KV	Verify with other tests Probably Excess Neutralising minerals
	TC89	59.57-62.8	3.23	2.26	7.29	SST, SLT, KV	Verify with other tests
	TC102 <sup>b</sup>	62.8-63.03	0.23	-19.99	-4.60	C3	Verify with other tests
	TC22	63.03-90.4	27.37	21.53	589.27	SST, SLT, KV, SH, GRT, CSH	Probably Excess Neutralising minerals
	TC103 <sup>b</sup>		0.4			C2	Potential Acid Generator
	10103	90.4-90.8	0.4	-339.12	-135.65	02	Potential Acid Generator
Y106048	TC32	5.95-6.66	0.71	-3.71	-2.64	SST, KV	Verify with other tests
	TC12	6.66-22.61	15.95	-3.45	-55.09	SST,SLT, GRT	Verify with other tests
	TC8	22.61-45.06	22.45	19.33	433.88	SST, SLT, GRT	Verify with other tests
	TC95	50.59-70.93	20.34	-5.84	-118.73	SST, SLT	Verify with other tests
	TC19	70.93-75.27	4.34	-18.71	-81.18	SST, SLT,GRT	Verify with other tests
Z145168	TC31	1.6-6.1	4.5	1.31	5.90	SST, GKS, KV	Verify with other tests
	TC115	6.1-49.68	43.58	19.18	835.84	DO	Verify with other tests
	TC100	49.68-56.74	7.06	90.54	639.24	SST, SLT, KV	Probably Excess Neutralising minerals
	TC37	56.74-57.36	0.62	48.54	30.10	DO	Probably Excess Neutralising minerals
	TC68	57.36-68.81	11.45	12.62	144.49	SST, SLT, MST	Verify with other tests
	TC85	68.81-94.75	25.94	22.74	590.00	SST, SLT, GRT	Probably Excess Neutralising minerals
	TC67	96.22-136.4	40.18	-27.34	-1098.63	SST, SLT, C, CSH, DO	Potential Acid Generator
	TC42	139.31-142.08	2.77	-19.66	-54.44	SST, SLT	Verify with other tests

Side name	Lab No.	Depth (m)	Thickness (m)	NNP	NNP x Thickness	Geology	NNP - Interpretation
Z145198	TC71	3.33-45.04	41.71	23.48	979.17	DO	Probably Excess Neutralising minerals
	TC90	45.04-90.02	44.98	22.55	1014.42	SST,SLT,SH	Probably Excess Neutralising minerals
	TC26	91.03-127.78	36.75	93.60	3439.70	SST,SLT,SH	Probably Excess Neutralising minerals
	TC2	132.14-134.94	2.8	-67.92	-190.16	SST, SLT, C	Potential Acid Generator
	TC72	134.94-151.78	16.84	-2.10	-35.42	SST, SLT, SH	Verify with other tests
	TC88	151.78-184	32.22	-1.74	-56.05	SST, SST,C	Verify with other tests
Z145199	TC46	0-22.52	22.52	19.78	445.45	KLY,KV,DO	Verify with other tests
	TC17	22.52-28.75	6.23	-14.57	-90.78	SST, SLT	Verify with other tests
	TC50	28.75-52.14	23.39	16.50	385.86	DO,KV	Verify with other tests
	TC77	52.14-123.92	71.78	77.50	5563.16	SST,SLT,KV,DO,GR T	Probably Excess Neutralising minerals
	TC81	123.92-124.6	0.68	-34.77	-23.65	SH	Potential Acid Generator
	TC20	125.80-159.6	33.8	118.51	4005.59	SST, SLT, C	Probably Excess Neutralising minerals
	TC64	160.3-162.3	2	-16.12	-32.24	C, SST	Verify with other tests

## NNP x Thickness Side name Lab No. Depth(m) NNP Geology **NNP** - Interpretation (m) Thickness R10001 SF 11 1.62-12.46 10.84 36.96 400.69 DO Probably Excess Neutralising Minerals SF 12 56.73 12.46-24.48 12.02 681.94 SST, SLT, GRT Probably Excess Neutralising Minerals SF 13 24.48-25.26 0.78 11.68 9.11 C, CSH Verify with other tests SF61<sup>b</sup> 24.48-25.80 1.32 17.02 22.47 C5H, C5M, C5L Verify with other tests SF 14 25.8-52.66 26.86 64.65 1736.49 SLT, SST Probably Excess Neutralising Minerals SF 1 0.94 50.66-51.6 29.87 28.08 SLT, GRT Verify with other tests SF62<sup>b</sup> 51.6-56.92 5.32 58.16 309.39 C4L **Probably Excess Neutralising Minerals** SF 2 56.92-57.14 0.22 6.10 1.34 SLT Verify with other tests SF 15 57.14-85.28 28.14 10.38 292.17 SST, SLT Verify with other tests SF 16 6.54 170.85 SST 86.82-112.94 26.12 Verify with other tests SF 17 112.94-116.23 3.29 4.41 14.52 GRT Verify with other tests SST, GRT, SLT, SF 18 116.23-130.1 13.87 35.59 493.57 WSH Probably Excess Neutralising Minerals SF 19 130.1-143.1 13 132.44 1721.74 TIL Probably Excess Neutralising Minerals R100002 SF 20 14.1-18 74.45 290.35 GRD, SST, DO Probably Excess Neutralising Minerals 3.9 SF 21 18.3-37.5 19.2 53.63 1029.70 SST, SLT Probably Excess Neutralising Minerals 37.5-37.82 0.32 34.45 SF 3 11.03 GRT Probably Excess Neutralising Minerals C4L SF63<sup>b</sup> 285.19 37.82-43.08 5.26 54.22 Probably Excess Neutralising Minerals SF 4 2.64 1.62 4.28 SST, SLT 43.08-45.72 Verify with other tests SF 22 45.72-70.1 24.38 2.56 62.41 SST, SLT Verify with other tests SF 23 71.4-103.8 32.4 6.25 202.41 Verify with other tests SST, GRT SF 24 103.8-124.4 20.6 14.11 290.56 GRT, TIL, KV Verify with other tests Z124027 SF 45 4.13-14.04 9.91 39.16 388.08 DO. KV. Probably Excess Neutralising Minerals SF 46 14.04-48.99 34.95 133.61 4669.84 SST, SLT Probably Excess Neutralising Minerals SF 47 48.99-69.26 20.27 15.19 307.85 SST, DO Verify with other tests SF 48 69.26-90.34 21.08 42.20 889.53 SST, SLT Probably Excess Neutralising Minerals SF 49 90.44-91.84 1.4 1.09 1.53 C, CSH Verify with other tests SF64<sup>b</sup> 41.52 91.84-92.34 0.5 20.76 C5L **Probably Excess Neutralising Minerals** Verify with other tests SF 50 92.34-93.61 1.27 19.56 24.85 DO SF 51 93.61-123.04 29.43 25.56 752.29 SST, SLT **Probably Excess Neutralising Minerals** SF65<sup>b</sup> -100.00 -60.00 123.08-123.68 0.6 C4H Potential Acid Generator SH, SST, GRT, SF 52 123.68-125.28 1.6 -8.23 -13.16 SLT Potential Acid Generator SF 7 125.28-125.88 0.6 -55.04 -33.03 GRT Potential Acid Generator SF66<sup>b</sup> 125.88-129.68 3.8 43.18 164.10 C4L **Probably Excess Neutralising Minerals**

## Appendix 10: Interpretation of ABA Net Neutralising Potential Results (Mine 2)

Side name	Lab No.	Depth(m)	Thickness (m)	NNP	NNP x Thickness	Geology	NNP - Interpretation
	SF 8	129.78-134.51	4.73	3.58	16.94	SLT, SST, C SLT, SST, C,	Verify with other tests
	SF 53	134.48-162.76	28.28	35.06	991.54	DO	Probably Excess Neutralising Minerals
	SF 54	162.54-164.8	2.26	20.24	45.75	DO, SST, C	Verify with other tests
	SF 55	167.2-172.87	5.67	-41.15	-233.33	GRT, SLT	Potential Acid Generator
	SF 56	172.87-180.67	7.8	32.09	250.31	SST, DO	Probably Excess Neutralising Minerals
	SF 57	180.67-183.7	3.03	36.79	111.46	DO	Probably Excess Neutralising Minerals
	SF 58	183.7-186.86	3.16	11.70	36.96	SST, SLT	Verify with other tests
	SF 59	186.86-189.02	2.16	5.82	12.58	TIO, SST	Verify with other tests
	SF 60	189.02-200.16	11.14	8.18	91.10	QZT, EOH	Verify with other tests
Z124029	05.05		4.0	40.00	10.01	10/ 007	
	SF 25	6.9-8.1	1.2	10.86	13.04	KV, SST	Verify with other tests
	SF 26	8.1-61.46	53.36	40.29	2149.79	DO	Probably Excess Neutralising Minerals
	SF 27	61.46-73.68	12.22	35.58	434.84	SST, SLT, GRT	Probably Excess Neutralising Minerals
	SF 28	73.68-74.72	1.04	13.97	14.53	C, CSH	Verify with other tests
	SF 29	74.72-98.23	23.51	48.32	1136.01	SST, SLT	Probably Excess Neutralising Minerals
	SF 5	98.23-98.88	0.65	30.59	19.88	SST, SLT	Potential Acid Generator
	SF67 <sup>b</sup>	98.88-102.83	3.95	36.42	143.85	C4L	Probably Excess Neutralising Minerals
	SF 6	102.83-104.6	1.77	38.72	68.53	GRT	Probably Excess Neutralising Minerals
	SF 30	104.6-132.74	28.14	11.93	335.73	SST, SLT, GRT	Verify with other tests
	SF 31	133.63-136.28	2.65	-16.76	-44.40	SST, SLT, GRT	Potential Acid Generator
	SF 32	138.64-158.45	19.81	-15.74	-311.79	SST, SLT, GRT	Potential Acid Generator
	SF 33	158.45-165.43	6.98	20.70	144.49	SST, SLT, TIL	Verify with other tests
	SF 34	165.43-174.53	9.1	87.03	791.98	GR, EOH	Probably Excess Neutralising Minerals
Z124030	SF 35	11.6-12.52	0.92	10.53	9.68	SST	Verify with other tests
	SF 36	12.52-20.8	8.28	108.52	898.51	DO, KV	Probably Excess Neutralising Minerals
	SF 37	20.8-41.35	20.55	12.23	251.38	SST, SLT	Verify with other tests
	SF 38	41.35-49	7.65	54.92	420.17	DO	Probably Excess Neutralising Minerals
	SF 39	49-98.3	49.3	30.49	1503.21	SST, SLT	Probably Excess Neutralising Minerals
	SF68	99.2-99.42	0.22	-229.70	-50.54	C5H (Coal)8	Potential Acid Generator
	SF 40	98.3-100.45	2.15	16.54	35.56	C, SLT	Verify with other tests
	SF69	100.45-100.8	0.35	17.98	6.29	C5L (Coal)9	Verify with other tests
	SF 41	100.8-133.37	32.57	7.21	234.93	SST, SLT	Verify with other tests
	SF70	133.37-133.92	0.55	-44.34	-24.39	C4H (Coal)10	Potential Acid Generator
	SF 9	133.92-135.56	1.64	-9.33	-15.29	GRT	Potential Acid Generator
	SF71 <sup>b</sup>	135.56-139.83	4.27	78.46	335.03	C4L	Probably Excess Neutralising Minerals
	SF 10	139.83-140.6	0.77	3.00	2.31	SST	Verify with other tests
	SF 42	140.6-164.48	23.88	19.44	464.18	C, SST, SLT, KV	Verify with other tests
	SF 43	167.1-168.2	1.1	70.97	78.07	SLT	Probably Excess Neutralising Minerals
	SF 44	168.2-185.66	17.46	91.09	1590.45	SST, TIO, LAV	Probably Excess Neutralising Minerals

Side name	Lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
F142441	TC18	5.91	2.96	No Acid Potential	Acid under certain conditions
	TC75	76.08	38.04	No Acid Potential	No Acid Potential
	TC38	38.31	19.16	No Acid Potential	No Acid Potential
	TC83	31.20	15.60	No Acid Potential	No Acid Potential
	TC7	2.20	1.10	Acid under certain conditions	Acid under certain conditions
	TC35	0.03	0.02	Likely Acid Generator	Likely Acid Generator
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F142446	TC66	26.61	13.31	No Acid Potential	No Acid Potential
	TC57	73.57	36.79	No Acid Potential	No Acid Potential
	TC 62	9.30	4.65	No Acid Potential	No Acid Potential
	TC10	44.54	22.27	No Acid Potential	No Acid Potential
	TC4	43.40	21.70	No Acid Potential	No Acid Potential
	TC51	0.49	0.24	Likely Acid Generator	Likely Acid Generator
	TC41	0.39	0.19	Likely Acid Generator	Likely Acid Generator
	TC33	4.49	2.25	No Acid Potential	Acid under certain conditions
	TC87	8.57	4.28	No Acid Potential	No Acid Potential
F142471	TC65	23.52	11.76	No Acid Potential	No Acid Potential
	TC55	32.30	16.15	No Acid Potential	No Acid Potential
	TC61	30.90	15.45	No Acid Potential	No Acid Potential
	тС3	7.77	3.88	No Acid Potential Acid under certain	Acid under certain conditions Acid under certain
	TC107 <sup>b</sup>	3.80	1.90	conditions	conditions
	TC11	0.00	0.00	Likely Acid Generator	Likely Acid Generator
	TC112 <sup>b</sup>	0.11	0.06	Likely Acid Generator	Likely Acid Generator
	TC84	12.73	6.37	No Acid Potential	No Acid Potential
	TC105 <sup>b</sup>	8.37	4.18	No Acid Potential	No Acid Potential
	TO114 <sup>b</sup>	2.64	1.00	Acid under certain	Acid under certain
	TC114 <sup>b</sup>	2.64	1.32	conditions	conditions
0105010	TOOL	252.40	100 75	No Apid Detertiol	No Apid Datastial
O105016	TC25	253.49	126.75	No Acid Potential	No Acid Potential
	TC113	49.73	24.86	No Acid Potential	No Acid Potential
	TC48 TC43	<u>158.89</u> 3.71	79.44 1.86	No Acid Potential Acid under certain conditions	No Acid Potential Acid under certain conditions
				Acid under certain	Acid under certain
	TC101	2.29	1.15	conditions	conditions
	TC34	0.04	0.02	Likely Acid Generator	Likely Acid Generator
	TC24	20.04	10.02	No Acid Potential	No Acid Potential Acid under certain
	TC14	6.65	3.33	No Acid Potential	conditions
	TC79	0.02	0.01	Likely Acid Generator	Likely Acid Generator

## Appendix 11: Interpretation of the NP/AP ratio of Mine 1

Side name	Lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
P110030	TC49	16.93	8.46	No Acid Potential	No Acid Potential
	T52	4.27	2.13	No Acid Potential	Acid under certain conditions
	Т99	4.91	2.45	No Acid Potential	Acid under certain conditions
	TC1	3.04	1.52	Acid under certain conditions	Acid under certain conditions
	TC5	34.42	17.21	No Acid Potential	No Acid Potential
	TC74	2.66	1.33	Acid under certain conditions	Acid under certain conditions Acid under certain
	TC117 <sup>b</sup>	2.47	1.23	Acid under certain conditions	Acid under certain conditions Acid under certain
	TC104	2.09	1.04	Acid under certain conditions	conditions Acid under certain
	TC109 <sup>b</sup>	2.49	1.25	Acid under certain conditions	conditions
	TC82	4.03	2.02	No Acid Potential	Acid under certain conditions
	TC27	1.59	0.79	Acid under certain conditions	Likely Acid Generator
	TC108 <sup>b</sup>	0.32	0.16	Likely Acid Generator	Likely Acid Generator
	TC98	2.14	1.07	Acid under certain conditions	Acid under certain conditions
	TC106 <sup>b</sup>	0.14	0.07	Likely Acid Generator	Likely Acid Generator
P110087	ТС9	618.77	309.39	No Acid Potential	No Acid Potential
	TC13	9.91	4.95	No Acid Potential	No Acid Potential
	TC97	176.60	88.30	No Acid Potential	No Acid Potential
	TC58	1.04	0.52	Acid under certain conditions	Likely Acid Generator
	TC23	1.53	0.77	Acid under certain conditions	Likely Acid Generator
	TC94	64.52	32.26	No Acid Potential	No Acid Potential
	TC91	0.23	0.12	Likely Acid Generator	Likely Acid Generator
	TC45	0.46	0.23	Likely Acid Generator	Likely Acid Generator
	TC53	0.27	0.14	Likely Acid Generator	Likely Acid Generator Acid under certain
	TC6	2.25	1.12	Acid under certain conditions	conditions
	TC73	0.34	0.17	Likely Acid Generator	Likely Acid Generator
R146043	TC93	89.25	44.62	No Acid Potential	No Acid Potential
	TC116	37.50	18.75	No Acid Potential	No Acid Potential
	TC47	301.61	150.80	No Acid Potential	No Acid Potential
	TC59	65.64	32.82	No Acid Potential	No Acid Potential
	TC29	52.45	26.22	No Acid Potential	No Acid Potential
	TC16	74.01	37.01	No Acid Potential	No Acid Potential
	TC28	0.13	0.06	Likely Acid Generator	Likely Acid Generator
	TC110 <sup>b</sup>	0.67	0.33	Likely Acid Generator	Likely Acid Generator
	TC40	2.57	1.29	Acid under certain conditions	Acid under certain conditions
	TC15	2.91	1.46	Acid under certain conditions	Acid under certain conditions
	TC86	22.39	11.19	No Acid Potential	No Acid Potential

Side name	Lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
	TC92	0.04	0.02	Likely Acid Generator	Likely Acid Generator
	TC21	0.16	0.08	Likely Acid Generator	Likely Acid Generator
	TC118 <sup>b</sup>	0.56	0.28	Likely Acid Generator	Likely Acid Generator
T139228	TC70	59.85	29.93	No Acid Potential	No Acid Potential
	TC44	9.62	4.81	No Acid Potential	No Acid Potential
	TC36	227.59	113.79	No Acid Potential	No Acid Potential
	TC63	22.90	11.45	No Acid Potential	No Acid Potential
	TC30	85.59	42.79	No Acid Potential	No Acid Potential
	TC39	0.86	0.43	Likely Acid Generator	Likely Acid Generator
	TC76	0.57	0.28	Likely Acid Generator	Likely Acid Generator
	TC96	1.23	0.62	Acid under certain conditions	Likely Acid Generator
	TC78	0.00	0.00	Likely Acid Generator	Likely Acid Generator
	TC56	0.00	0.00	Likely Acid Generator	Likely Acid Generator
	TC60	1.33	0.67	Acid under certain conditions	Likely Acid Generator
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W569001	TC69	185.03	92.51	No Acid Potential	No Acid Potential
	TC80	0.66	0.33	Likely Acid Generator	Likely Acid Generator
	TC54	6.89	3.44	No Acid Potential	Acid under certain conditions
	TC111 <sup>b</sup>	11.07	5.53	No Acid Potential	No Acid Potential
	TC89	4.43	2.21	No Acid Potential	Acid under certain conditions
	TC102 <sup>b</sup>	1.25	0.63	Acid under certain conditions	Likely Acid Generator
	TC22	9.94	4.97	No Acid Potential	No Acid Potential
	TC103 <sup>b</sup>	0.03	0.01	Likely Acid Generator	Likely Acid Generator
	1	1	1	T	1
Y106048	TC32	0.01	0.00	Likely Acid Generator	Likely Acid Generator
	TC12	1.63	0.81	Acid under certain conditions	Likely Acid Generator
	TC8	7.79	3.89	No Acid Potential	Acid under certain conditions
	TC95	1.53	0.77	Acid under certain conditions	Likely Acid Generator
	TC19	0.12	0.06	Likely Acid Generator	Likely Acid Generator
Z145168	TC31	121.54	60.77	No Acid Potential	No Acid Potential
	TC115	15.22	7.61	No Acid Potential	No Acid Potential
	TC100	350.76	175.38	No Acid Potential	No Acid Potential
	TC37	31.63	15.82	No Acid Potential	No Acid Potential
	TC68	34.49	17.24	No Acid Potential	No Acid Potential
	TC85	21.32	10.66	No Acid Potential	No Acid Potential
	TC67	0.76	0.38	Likely Acid Generator	Likely Acid Generator
	TC42	0.00	0.00	Likely Acid Generator	Likely Acid Generator

Side name	Lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
Z145198	TC71	10.61	5.31	No Acid Potential	No Acid Potential
	TC90	51.14	25.57	No Acid Potential	No Acid Potential
	TC26	53.82	26.91	No Acid Potential	No Acid Potential
	TC2	0.00	0.00	Likely Acid Generator	Likely Acid Generator
	TC72	1.55	0.78	Acid under certain conditions	Likely Acid Generator
	TC88	1.46	0.73	Acid under certain conditions	Likely Acid Generator
Z145199	TC46	112.26	56.13	No Acid Potential	No Acid Potential
	TC17	0.01	0.01	Likely Acid Generator	Likely Acid Generator
	TC50	17.92	8.96	No Acid Potential	No Acid Potential
	TC77	60.97	30.49	No Acid Potential	No Acid Potential
	TC81	0.57	0.29	Likely Acid Generator	Likely Acid Generator
	TC20	12.88	6.44	No Acid Potential	No Acid Potential
	TC64	0.34	0.17	Likely Acid Generator	Likely Acid Generator

Side name	lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
R100001	SF 11	37.34	18.67	No Acid Potential	No Acid Potential
	SF 12	582.88	291.44	No Acid Potential	No Acid Potential
	SF 13	4.27	2.13	No Acid Potential	Acid under certain conditions
	SF61	23.68	11.84	No Acid Potential	No Acid Potential
	SF 14	17.24	8.62	No Acid Potential	No Acid Potential
	SF 1	1.94	0.97	Acid under certain conditions	Likely Acid Generator
	SF62	15.09	7.55	No Acid Potential	No Acid Potential
	SF 2	4.86	2.43	No Acid Potential	Acid under certain conditions
	SF 15	4.86	2.43	No Acid Potential	Acid under certain conditions
	SF 16	8.71	4.36	No Acid Potential	No Acid Potential
	SF 17	4.07	2.03	No Acid Potential	Acid under certain conditions
	SF 18	15.62	7.81	No Acid Potential	No Acid Potential
	SF 19	548.89	274.45	No Acid Potential	No Acid Potential
			-		
R100002	SF 20	36.99	18.50	No Acid Potential	No Acid Potential
	SF 21	7.79	3.90	No Acid Potential	Acid under certain conditions
	SF 3	3.96	1.98	Acid under certain conditions	Acid under certain conditions
_	SF63	12.33	6.16	No Acid Potential	No Acid Potential
	SF 4	1.13	0.56	Acid under certain conditions	Likely Acid Generator
	SF 22	1.23	0.62	Acid under certain conditions	Likely Acid Generator
	SF 23	5.65	2.82	No Acid Potential	Acid under certain conditions
	SF 24	13.04	6.52	No Acid Potential	No Acid Potential
	1	T	1		T
Z124027	SF 45	106.63	53.31	No Acid Potential	No Acid Potential
	SF 46	425.12	212.56	No Acid Potential	No Acid Potential
	SF 47	23.68	11.84	No Acid Potential	No Acid Potential
	SF 48	50.17	25.08	No Acid Potential	No Acid Potential
	SF 49	1.10	0.55	Acid under certain conditions	Likely Acid Generator
	SF64	4.29	2.15	No Acid Potential	Acid under certain conditions
	SF 50	13.23	6.61	No Acid Potential	No Acid Potential
	SF 51	8.33	4.17	No Acid Potential	No Acid Potential
	SF65	0.11	0.05	Likely Acid Generator	Likely Acid Generator
	SF 52	0.45	0.23	Likely Acid Generator	Likely Acid Generator
	SF 7	0.08	0.04	Likely Acid Generator	Likely Acid Generator
	SF66	63.86	31.93	No Acid Potential	No Acid Potential
	SF 8	2.24	1.12	Acid under certain conditions	Acid under certain conditions
	SF 53	14.70	7.35	No Acid Potential	No Acid Potential
	SF 54	4.05	2.03	No Acid Potential	Acid under certain conditions
	SF 55	0.36	0.18	Likely Acid Generator	Likely Acid Generator
	SF 56	40.35	20.17	No Acid Potential	No Acid Potential

Side name	lab No.	Neutralising Potential Ratio(NP/AP) for Open System	Neutralising Potential Ratio(NP/AP) for Closed System	Interpretation Open System	Interpretation Closed System
	SF 57	204.48	102.24	No Acid Potential	No Acid Potential
	SF 58	4.52	2.26	No Acid Potential	Acid under certain conditions
	SF 59	1.69	0.85	Acid under certain conditions	Likely Acid Generator
	SF 60	1.70	0.85	Acid under certain conditions	Likely Acid Generator
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Z124029	SF 25	175.09	87.55	No Acid Potential	No Acid Potential
	SF 26	42.55	21.28	No Acid Potential	No Acid Potential
	SF 27	292.87	146.44	No Acid Potential	No Acid Potential
	SF 28	4.92	2.46	No Acid Potential	Acid under certain conditions
	SF 29	13.30	6.65	No Acid Potential	No Acid Potential
	SF 5	1.58	0.79	Acid under certain conditions	Likely Acid Generator
	SF67	13.68	6.84	No Acid Potential	No Acid Potential
	SF 6	6.30	3.15	No Acid Potential	Acid under certain conditions
	SF 30	5.02	2.51	No Acid Potential	Acid under certain conditions
	SF 31	0.35	0.18	Likely Acid Generator	Likely Acid Generator
	SF 32	0.41	0.20	Likely Acid Generator	Likely Acid Generator
	SF 33	17.43	8.72	No Acid Potential	No Acid Potential
	SF 34	124.87	62.44	No Acid Potential	No Acid Potential
Z124030	SF 35	460.36	230.18	No Acid Potential	No Acid Potential
	SF 36	566.83	283.41	No Acid Potential	No Acid Potential
	SF 37	13.06	6.53	No Acid Potential	No Acid Potential
	SF 38	29.19	14.59	No Acid Potential	No Acid Potential
	SF 39	27.40	13.70	No Acid Potential	No Acid Potential
	SF68	0.11	0.05	Likely Acid Generator	Likely Acid Generator
	SF 40	2.69	1.34	Acid under certain conditions	Acid under certain conditions
	SF69	3.97	1.98	Acid under certain conditions	Acid under certain conditions
	SF 41	1.41	0.70	Acid under certain conditions	Likely Acid Generator
	SF70	0.19	0.09	Likely Acid Generator	Likely Acid Generator
	SF 9	0.43	0.22	Likely Acid Generator	Likely Acid Generator
	SF71	95.76	47.88	No Acid Potential	No Acid Potential
	SF 10	1.81	0.90	Acid under certain conditions	Likely Acid Generator
	SF 42	4.98	2.49	No Acid Potential	Acid under certain conditions
	SF 43	83.12	41.56	No Acid Potential	No Acid Potential
	SF 44	206.27	103.14	No Acid Potential	No Acid Potential